

- Ecotoxicological effects of transformed silver and
- 2 titanium dioxide nanoparticles in the effluent from a
- 3 lab-scale wastewater treatment system
- 4 Anastasia Georgantzopoulou † *, Patricia Almeida Carvalho ‡ , Christian Vogelsang † , Mengstab
- 5 $Tilahun^{\dagger}$, $Kuria\ Ndungu^{\dagger}$, $Andy\ M.\ Booth^{\Box}$, $Kevin\ V.\ Thomas^{\dagger, \parallel}$ and $Ailbhe\ Macken^{\dagger}$
- 6 [†]NIVA, Norwegian Institute for Water Research, Gaustadalleen 21, 0349, Oslo, Norway
- 7 [‡]SINTEF Materials and Chemistry, Forskningsveien 1, 0373, Oslo, Norway
- 8 SINTEF Ocean, Brattørkaia 17C, 7010, Trondheim, Norway
- 9 Queensland Alliance for Environmental Health Sciences (QAEHS), University of Queensland,
- 10 20 Cornwall Street, Woolloongabba, Queensland, 4102 Australia

ABSTRACT

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In this study, a lab-scale wastewater treatment plant (WWTP), simulating biological treatment, received 10 μg/L Ag and 100 μg/L TiO₂ nanoparticles (NPs) for five weeks. NP partitioning was evaluated by size fractionation (>0.7 μm, 0.1-0.7 μm, 3 kDa-0.1 μm, <3 kDa) using inductively coupled plasma mass spectrometry (ICP-MS), single particle ICP-MS and transmission electron microscopy. The ecotoxicological effects of the transformed NPs in the effluent were assessed using a battery of marine and freshwater bioassays (algae and crustaceans) and an in vitro gill cell line model (RTgill-W1). TiO₂ aggregates were detected in the effluent, while Ag NPs (0.1 to 0.22 μg/L) were associated with S, Cu, Zn. Fractionation showed that >80% of Ag and Ti were associated with the effluent solids. Increased toxicity was observed during weeks 2-3 and the effects were species-dependent; with marine epibenthic copepods and algae being the most sensitive. Increased reactive oxygen species formation was observed in vitro followed by an increase in epithelial permeability. The effluent affected the gill epithelium integrity in vitro and impacted defense pathways (upregulation of multixenobiotic resistance genes). To our knowledge, this is the first study to combine a lab-scale activated sludge WWTP with extensive characterization techniques and ecotoxicological assays to study the effects of transformed NPs in the effluent.

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INTRODUCTION

The production and use of consumer products containing Ag and TiO ₂ NPs continues to
increase ^{1,2} and due to their widespread use and application they can enter sewage streams and
wastewater treatment plants (WWTPs). Their presence in the influent of WWTPs has been
reported in several studies ³⁻⁷ . Ag and TiO ₂ NPs tend to be associated with particulate matter and
appear to be relatively efficiently removed from the wastewater during primary and secondary
treatment ^{3,4,6,8,9} , the extent of removal however depends on the design and efficiency of the
operating conditions ⁶ . Ag-based and TiO ₂ NPs have been detected in wastewater effluents ^{6,9}
making their release in surface waters through effluent discharge possible, which can potentially
be an important exposure route for aquatic organisms in receiving waters.
Nanoparticles undergo a combination of physical and chemical transformations in environmental
media (e.g. wastewaters) ¹⁰ that may influence their behavior, bioavailability and toxicity ^{11,12} .
Their behavior may differ from their pristine NP counterparts, thereby making comparisons and
predictions between transformed and pristine NPs difficult. It has been reported that Ag NPs are
sulfidized to various degrees in wastewater streams and during transport to WWTPs ^{8,13} .
Furthermore, studies using a pilot WWTP fed with municipal wastewater spiked with Ag NPs,
showed a transformation to Ag ₂ S while some of the Ag NPs detected in the effluent were still in
the pristine metallic form ¹⁴ . Even though most NPs present in the natural environment are likely
to have undergone some form of physicochemical transformation, very few effects studies have
employed transformed NPs ¹⁵⁻¹⁷ or NPs in environmentally relevant media such as
wastewaters ^{12,18,19} . One recent study has shown that Cu NP transformation through a septic tank
led to a lack of toxicity in a zebrafish embryo hatching assay ¹⁵ . A decreased toxicity was also
observed for the freshwater amphipod <i>H. Azteca</i> exposed to Ag NPs transformed through an

activated sludge simulation system¹⁷ while another study showed an increased zebrafish embryo 54 toxicity in the effluent of a similar system dosed with 4-16 mg/L Ag NPs²⁰. Studies using 55 56 sulfidized Ag NPs through wastewater treatment processes demonstrated that although Ag₂S NPs are less soluble, they can still be bioavailable to different organisms^{21,22} and induce toxicity. 57 though at lower levels compared to pristine Ag NPs²³. This highlights the need of a better 58 59 understanding of the behavior of NPs, their transformation and their toxicity in complex media. 60 It remains a challenging task to detect and quantify NPs at low, but environmentally relevant 61 concentrations ($< \mu g/L$) in complex matrices such as wastewater, effluent, sewage sludge, and surface waters²⁴. As a result, most environmental fate studies and toxicological assessments are 62 conducted at much higher concentrations than those expected to be found in the environment²⁰, 63 and studies taking into account relevant exposures at more realistic conditions are scarce 15,16. 64 65 There is a need to develop a better understanding of the environmental impact of transformed NPs at environmentally relevant concentrations²⁵. 66 67 The current study investigates the potential hazard of transformed Ag and TiO₂ NPs through 68 advanced biological treatment processes present in complex environmental media such as 69 WWTP effluents at environmentally relevant NP concentrations. A lab-scale pre-denitrifying 70 WWTP system with pre-conditioned activated sludge was established and continuously fed with 71 artificial wastewater dosed with 10 μg/L Ag and 100 μg/L TiO₂ NPs for a period of 5 weeks. 72 The system was combined with a battery of marine and freshwater bioassays and NP 73 characterization techniques to evaluate the hazard potential of transformed Ag and TiO₂ NPs. 74 Sequential filtration combined with ICP-MS was applied to characterize the different size 75 fractions (associated with settling solids, colloidal matter, nanoscale and dissolved). Both marine 76 (Skeletonema pseudocostatum, Tisbe battagliai) and freshwater (Raphidocelis subcapitata,

Daphnia magna) organisms (algae and crustaceans) were used as model species to monitor the toxicity of the transformed NPs present in the effluent during the 5-week dosing period. The choice of organisms reflects that the behavior of NPs differs in marine and freshwater environments, the effects may vary depending on the species used²⁶ as well as the fact that in many countries the effluent is discharged in both freshwater and marine environments.

Furthermore, an *in vitro* model using the rainbow trout (*Oncorhynchus mykiss*) gill cell line RTgill-W1 was employed, representing a major interface between the organism and its environment that is one of the first sites impacted by waterborne chemicals. The model was used in addition to the standard bioassays for assessment of the effluent with minimal sample modification during the period of dosing of the WWTP system and cellular responses were assessed (metabolic activity, epithelial integrity, reactive oxygen species (ROS) formation and the gene expression of zonula occludens-1 and multixenobiotic resistance genes ABCB1, ABCC1 and ABCC2).

MATERIALS AND METHODS

Nanoparticles and chemicals. Polyvinylpyrrolidone (PVP)-coated Ag NPs (Econix 25 nm, aqueous suspension) were obtained from Nanocomposix (Czech Republic). TiO₂ NPs (NM-101, primary particles of 5 nm) were obtained from the Joint Research Centre (JRC Repository, Ispra, Italy) and have been extensively characterized previously²⁷. A stock dispersion of TiO₂ NPs in 0.22 μm filtered MilliQ (2.56 mg/ml) was prepared in a Scint-Burk glass vial and sonicated in ice water for 13 min with a calibrated probe sonicator according to the FP7 EU NANoREG sonication protocol²⁸. The NP stock dispersions were then analyzed with scanning transmission

electron microscopy (STEM), single particle (sp-ICP-MS, see sections below) and dynamic light scattering (DLS) (Supporting Information; SI). AgNO₃ (Sigma-Aldrich) was used as an ionic control.

Lab-scale wastewater treatment plant. The lab-scale WWTP was a pre-denitrifying activated

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sludge treatment system comprised of a 6.5 L non-aerated denitrifying reactor, an 8 L aerated nitrifying reactor with automatic temperature (20°C) and pH (7.2) control and a 5.1 L settler (SI; Figure S1). The activated sludge used in the system was collected from Bekkelaget WWTP, Oslo, Norway. To adapt the activated sludge to the synthetic medium and to wash out any NPs transferred to the system together with the initial sludge, the system was continuously fed (hydraulic retention time ~15 h) synthetic wastewater without NPs for a period of 10 weeks. The composition and characteristics of the synthetic wastewater and a detailed description of the system operation and the parameters measured are presented in the SI. Sludge was continuously removed from the denitrifying reactor to maintain a solids retention time (SRT) of \sim 15 days. During the adaptation period effluent samples from the reference system without NPs were collected weekly and served as "background controls". After the adaptation period the synthetic medium was dosed with a continuous supply of 10 μg/L Ag NPs and 100 μg/L TiO₂ NPs to the denitrifying reactor for a period of 5 weeks. The synthetic wastewater containing Ag and TiO₂ NPs was freshly prepared every 2-3 days. Effluent samples were collected weekly and used to evaluate the influence of NP transformation on the battery of bioassays (performed within 48 h of effluent collection). The COD and total inorganic N removal was 81±8 % and 71±16 %, respectively (SI).

Ag and TiO₂ NP characterization (STEM/EDS, sp-ICP-MS). Ag, TiO₂ NP stock dispersions or effluent samples were imaged using STEM, while elemental point analysis and mapping were

122 performed with energy-dispersive X-ray spectroscopy (EDS). A detailed description of the 123 STEM-EDS method is presented in the SI. 124 The effluent samples as prepared for STEM were transferred to Eppendorf tubes, vortexed for 30 125 s, sonicated for 30 min, and then diluted with MilliQ water prior to single particle ICP-MS (sp-126 ICP-MS) analysis for particle concentration and size. The sp-ICP-MS analytical protocol and data analysis (using the single particle RIKILT calculation tool²⁹, Wageningen, The Netherlands) 127 are similar to those described elsewhere^{9,29} (detailed description of the sp-ICP-MS method in SI). 128 129 Ag and TiO₂ fractionation (filtration, ultrafiltration and ICP-MS). Samples from the 130 influent, nitrifying and denitrifying reactors, as well as the effluent (collected from the overflow 131 of the settler), were collected weekly and fractionated using a series of membranes with 132 decreasing pore size immediately upon sample collection. The samples were filtered sequentially 133 through a 0.7 µm filter membrane (glass microfiber GF/F, Whatman, GE Healthcare Life 134 Sciences), a 0.1 µm membrane (Durapore membrane filter, Millipore) and finally centrifuged 135 through a 3 kDa cut-off membrane (Amicon Ultra-15, Millipore, 5000g for 1 h) to obtain the 136 soluble fraction present in the filtrate sample. The 0.7 µm filters were dried at 45°C for 2 h and 137 kept in microwave tubes until further analysis (solid-associated fraction or particles >0.7 μm). 138 The solids-associated (>0.7 μ m), particulate (0.1-0.7 μ m), NP (3 kDa cut-off - 0.1 μ m) and the 139 soluble fraction (3 kDa filtrate) were analyzed by ICP-MS (see SI for details). 140 Skeletonema pseudocostatum growth inhibition assay. The marine algae were cultured in ISO media³⁰ prepared from filtered natural seawater (35 ppt salinity), and maintained at 20°C under 141 142 continuous light and shaking according to the ISO 10253 standard. Dilution water used for the 143 exposure assays was a modified version of the ISO media with a reduced concentration (1/5) of

trace elements and EDTA to minimize free metal ion complexation ³¹ and possible impacts on the
toxicity profile of the effluent. The effluent was spiked with concentrated ISO media stock
solutions to reach the elemental concentrations present in the dilution water. Artificial sea salts
(Coral Pro Salt) were added to reach 35 ppt salinity. Increasing concentrations (5 concentrations:
6.2-100%) of effluent or pristine NPs and AgNO ₃ were placed in a 12-well plate (1.35 ml/well,
triplicates). Exponentially growing algae were counted with a hemocytometer and 150 μl of
$1\cdot10^5$ cells/ ml were added to each well (final algal concentration $1\cdot10^4$ cells/ml). An artificial
seawater control was prepared by spiking artificial sea salts (to achieve 35ppt) into clean dilution
water. Filtered natural seawater with reduced trace elements and EDTA concentrations served as
an untreated control while "background" effluent control was also included. The algal cell
density and growth was assessed daily for 72 h by measuring fluorescence (excitation 530 nm:
emission 685 nm, Victor ³ Multilabel plate reader, PerkinElmer). The specific growth rate
(logarithmic increase in biomass) and the percent growth inhibition over the exposure period was
calculated according to the ISO standard.
Raphidocelis subcapitata growth inhibition assay. The freshwater algae were cultured in EPA
media ³² and maintained at 20°C under continuous light and shaking according to the OECD 201
guideline. The effluent was spiked with concentrated nutrient stock solutions to achieve the same
concentration as the standard media. Trace elements and EDTA were used at a reduced
concentration (1/5). 1.35 ml of increasing concentrations of effluent (5 concentrations: 6.2-
100%), pristine NPs or AgNO ₃ were placed in a 12-well plate. Finally, 150 μ l of algae (5·10 ⁵
cells/ml) in exponential growing phase were added per well (final algae concentration $5 \cdot 10^4$
cells/ml). Dilution water (MilliQ water supplemented with the concentrated stock solutions and
1/5 trace elements-EDTA) served as an untreated control and effluent collected during the

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stabilization period served as a "background" effluent control. The algal cell number and growth was measured daily for 72 h (fluorescence measurement, excitation 485 nm; emission 685 nm, Victor³ Multilabel plate reader, PerkinElmer). Effects of effluent on ROS formation (marine and freshwater algae). Exponentially growing algae were centrifuged and re-suspended in dilution water to achieve a concentration of 4·10⁶ cells/ml. 25 µl of cell suspension was placed in each well of a 96-well plate (final algal concentration 1·10⁶ cells/ml) and incubated in the dark with 25 µl DCFH-DA 20 µM (final concentration 10 µM) for 1.5 h under shaking conditions. At the end of the incubation period, 150 µl of effluent (serially diluted in dilution water) was added to each well and incubated for 1 h. At the end of the exposure period, DCF fluorescence was measured at wavelengths of 485 nm excitation and 535 nm emission. H₂O₂ was used as a positive control. **Daphnia magna acute toxicity assay.** Daphnids were maintained in M7 media³³ and fed with R. subcapitata every other day. Daphnids <24 h old were used for the assay, which was performed in 6-well plates as previously described³⁴ and according to OECD 202 guideline. Five daphnids per well were used in quadruplicate and were exposed to increasing concentrations of effluent (5 concentrations: 6.25-100%). Moderately hard EPA water was used for dilutions of the effluent³⁵. Daphnids in EPA water served as an untreated control while exposure to effluent collected during the stabilization period served as a "background" effluent control. The effects of pristine Ag NPs as well as spiked in background effluent (0.005-0.32 mg/L) were also evaluated. Daphnid mobility was assessed after 24 and 48 h. Tisbe battagliai acute toxicity assay. T. battagliai were maintained in filtered (0.22 um) seawater obtained from the outer Oslofjord and fed a mixed diet of Rhodomonas baltica and

<i>Isochrysis galbana</i> . Copepods of 6 ± 2 days old were used for the assay as previously
described ³⁶ . Tests were performed in 12-well plates with 5 animals (4 replicates per treatment) in
each well containing ~4 ml of test solution. Artificial salts (Coral Pro Salt) were added to the
effluent to reach a salinity of 35 ppt, with further dilutions made in the natural seawater used for
culture maintenance. The effects of increasing concentrations of the effluent (5 concentrations:
6.25-100%), Ag NPs (0.08-1.3 mg/L), TiO_2 NPs (0.01-10 mg/L) or $AgNO_3$ (0.01-0.16 mg/L) in
seawater or spiked in background effluent were assessed after 24 and 48 h of exposure. MilliQ
water spiked with artificial sea salts acted as an artificial seawater control. Natural seawater
served as an untreated control.
RTgill-W1 in vitro model in transwell inserts. The rainbow trout gill epithelial cell line RTgill-
W1 ³⁷ was provided by Prof. Kristin Schirmer (EAWAG, Switzerland). Cells were cultured in
Leibovitz's L-15 medium (L-15, Gibco, ThermoFischer Scientific) supplemented with 5% fetal
bovine serum (FBS, Gibco, ThermoFischer Scientific) and 1% gentamicin solution (10 mg/ml,
Sigma-Aldrich), and maintained at 19 °C in an incubator in the absence of CO ₂ . The cells were
seeded in 12-well transwell inserts (Millicell Hanging Cell Culture Insert, 1.0 μm , Merck
Millipore) at a concentration of $1.8 \cdot 10^5$ cells/ml (0.5 ml cell suspension/insert). The basolateral
compartment was filled with 1.5 ml of complete L-15 cell culture medium in a 12-well receiver
plate (Merck Millipore). Cells were allowed to grow for 10 days and form a confluent
monolayer. The media was renewed every other day.
Metabolic activity and epithelial integrity. On day 10, the cells were exposed for 24 h to
increasing concentrations of the freshly collected effluent from the system (filtered through a 0.2
μm filter; serial dilutions with a dilution factor of 2), the pristine NPs or AgNO3. Dilutions were
performed in L15/ex media as previously described ^{37,38} . Cells in L15/ex media served as an

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untreated control. At the end of the exposure period the media was removed and replaced with L15/ex media containing 100 µM alamar blue solution. Cells were incubated for 1 h and fluorescence was measured at wavelengths of 530 nm excitation and 590 nm emission (Victor³ Multilabel plate reader, PerkinElmer). The alamar blue solution was then removed and replaced with 0.1 mg/ml lucifer yellow (LY, Sigma-Aldrich) solution as a marker for paracellular permeability. The cells were incubated for 2 h before the inserts were removed from the receiver plates and fluorescence was measured at wavelengths of 485 nm excitation and 535 nm emission (Victor³ Multilabel plate reader, PerkinElmer). Quantitative real time PCR (qPCR). After exposure of the RTgill-W1 cells in transwell inserts, the exposure medium was removed, the cells were washed in PBS and were collected with 300 µl RLT plus buffer (Qiagen) supplemented with 1% mercaptoethanol. Total RNA was extracted using RNeasy Plus Mini Kit (Qiagen) according to the manufacturer's instructions and as previously described³⁹. The RNA purity and concentration were determined using a Nanodrop ND1000 spectrophotometer while RNA integrity was determined with an Agilent Bioanalyzer RNA 6000 nano series kit (Agilent technologies, USA). The qPCR was performed as previously described³⁹ (protocol details can be found in SI). Effects of effluent on ROS formation (in vitro). RTgill-W1 cells were seeded in 96-well plates at a concentration of 5·10⁵ cells/ml (100 µl cell suspension/well). After 24 h, the media was removed and fresh media containing 25 µM DCFH-DA in L15/ex media was placed in each well (100 µl solution/well). After a 1 h incubation, the DCFH-DA solution was removed and replaced with increasing concentrations of effluent (5 concentrations: 6-100%), Ag NPs, TiO₂ NPs or AgNO₃ diluted in L15/ex. Fluorescence was measured after 1 and 2 h of exposure at wavelengths of 485 nm excitation and 535 nm emission. H₂O₂ was used as a positive control.

Statistical analysis. Statistical analysis was performed with GraphPad Prism 6 (GraphPad Software, La Jolla, CA 92037, USA). Values are expressed as means ± standard deviation. Significant differences between the different treatments and control were analyzed with one-way analysis of variance (ANOVA) followed by Dunnet's multiple comparison test or nonparametric Kruskal-Wallis test followed by Dunn's multiple comparison test. Statistical significance was defined at p<0.05. Dose-response curves, EC₁₀ and EC₅₀ values were obtained with GraphPad Prism 6 (GraphPad Software, La Jolla, CA 92037, USA) using a logistic four-parameter model. Principal component analysis (PCA) of the parameters and effects observed with the different bioassays was performed with XLSTAT 2018 (SI).

RESULTS AND DISCUSSION

Ag and TiO_2 Nanoparticle characterization. The physicochemical properties determined for the Ag and TiO_2 NP stock dispersions in MilliQ water are summarized in the SI (Figures S2-3, Table S1). The Ag NPs were spherical with a mean diameter of 26.5 ± 0.7 nm and 58.8 ± 0.19 nm according to sp-ICP-MS and DLS measurements, respectively. DLS and sp-ICP-MS analyses showed an average TiO_2 aggregate size of 640.7 ± 9.2 and 278 ± 15 nm, respectively. STEM imaging indicated that TiO_2 NPs were porous and formed large aggregates consisting of individual particles below 10 nm, confirming previous reports on this material²⁷. In synthetic wastewater and seawater TiO_2 aggregates of 969 ± 19 nm and 1375 ± 76.7 nm, respectively were measured with DLS (SI; Table S1). Ag NPs in synthetic wastewater, seawater and the exposure media used in the bioassays ranged from 57.3 ± 0.17 to 59.5 ± 0.18 nm as measured with DLS, suggesting a stability of the PVP-coated Ag NPs in the different media. The higher (~2x) particle size obtained for both pristine Ag NPs and TiO_2 with DLS is probably related to the inherent properties of the instrument, light scattering techniques such as DLS require higher

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concentrations that can result in aggregation that could influence the analytical signal⁴⁰. With sp-ICP-MS low concentration levels can be detected in more complex or natural environmental samples. Therefore, multiple analytical techniques are necessary especially for low NP concentrations in environmental samples.

Ag and TiO₂ NP transformation in the lab-scale WWTP. Sequential filtration and ICP-MS analysis of the individual effluent fractions showed that >80% of the Ag and Ti measured was associated with suspended solids (>0.7 µm fraction) present in the effluent samples (Figure 1, Figure S4). The highest concentrations of both total Ag and Ti were observed in effluents from weeks 2 and 5. The Ti levels in the fraction >0.7 μm ranged from 0.9-24.2 μg/L, with the highest concentration measured at week 2. The dissolved Ag concentration was in the range of 0.005-0.021 µg/L (Table 1). The highest dissolved Ag concentrations were observed in effluents collected after 1 and 3 weeks of NP dosing, and corresponded to 7-8% of the total Ag measured during those weeks. The Ag concentration present in the NP fraction ranged from 0.1-0.22 µg/L, with the highest concentrations measured in the effluent samples collected in weeks 1, 3 and 5 (0.22, 0.14 and 0.17 µg/L, respectively). The Ti present in the 0.1 µm and 3 KDa fractions could not be distinguished and quantified separately, therefore the values are reported as Ti >0.7 µm and <0.7 µm. A previous study with sequencing batch reactors showed that a significant fraction of Ag was associated with colloidal material (below 0.45 µm)⁴¹ and biosolids in the sludge and effluent of a pilot WWTP¹⁴.

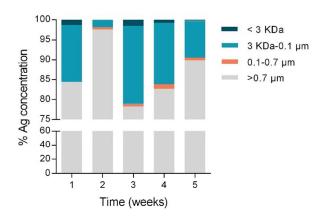


Figure 1. Effluent characterization and distribution of the total Ag present in the effluent of the lab-scale WWTP system during the 5 weeks of continuous dosing of the system.

Table 1. Ag and Ti concentrations (μ g/L or μ g/g effluent suspended solids) in each effluent fraction during the 5 weeks of operation and continuous dosing of the lab-scale WWTP system.

	Ag concentration								Ti concentration			
	Total		>0.7 μm		nano-Ag		3 KDa filtrate		>0.7 μm		<0.7µm	
Effluent Sample	μg/L	μgAg/gSS	μg/L	μgAg/gSS	μg/L	μgAg/gSS	μg/L	μgAg/gSS	μg/L	μgTi/gSS	μg/L	μgTi/gSS
wk 1	0.74	47.34	0.51	32.21	0.22	13.82	0.02	1.31	0.90	57.50	0.14	8.67
wk 2	5.99	72.15	5.84	70.41	0.11	1.28	< 0.005	0.06	24.20	291.52	0.13	1.55
wk 3	0.72	66.28	0.56	51.88	0.14	12.98	0.01	1.01	1.00	92.17	0.16	14.81
wk 4	0.65	47.90	0.54	39.60	0.10	7.37	< 0.005	0.37	2.50	183.15	0.10	7.50
wk 5	1.80	333.22	1.62	299.75	0.17	30.59	< 0.005	0.93	5.40	999.30	0.15	27.04

The effluent collected during the 4th week of system operation was analyzed by STEM in combination with EDS to determine both the presence and transformation of Ag and TiO₂ NPs. Electron microscopy images showed the presence of particles with high mass (bright contrast), while EDS analysis indicated that Ag-rich particles were associated with S, Cu and Zn (Figure 2A). STEM also showed the presence of TiO₂ polycrystalline aggregates (~50 nm) (Figure 2B)

comprised of primary particles below 10 nm which were similar to the initially dosed particles. The association of Ag present in WWTP with elements such as Cu, Zn and S is in accordance with previous studies reporting the presence of Ag particles associated with S in sludge^{14,42} and effluent samples¹⁴. It has recently been shown that secondary nano-sized Ag particles of approximately 20 nm diameter associated with S from organic or inorganic source are formed from dissolved silver from Ag NPs (80 nm, PVP coated) in batch systems with wastewater effluent and mixed liquor¹⁰.

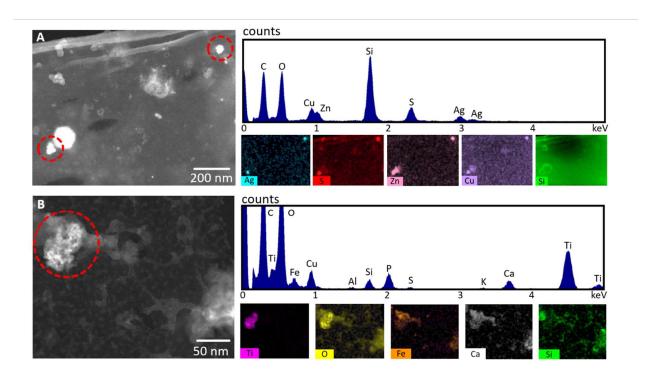


Figure 2. STEM images of (A) Ag-rich and (B) TiO₂ particles from the lab-scale WWTP, together with sum spectra of the encircled regions and elemental maps. Particles were detected in the effluent collected during the 4th week of dosing and operation of the system.

Single particle ICP-MS analysis of effluent samples collected during the 5 weeks of operation of the system confirmed the presence of Ag and TiO₂ NPs, indicating they occurred within the size ranges 20.5-31.6 nm and 110.9-124.8 nm, respectively (SI; Table S1). Sp-ICP-MS is a very promising technique for the identification and quantification of metallic NPs in complex matrices⁴³, including wastewater and effluents⁴⁴⁻⁴⁶. The technique has low detection limits⁴⁷ and requires highly diluted samples that are very relevant for environmental samples, as well as when realistic exposures are to be studied. However, distinction between Ag complexes and species or Ag bound colloids cannot be made⁴⁵.

Effects of effluents on algal growth and ROS formation. A 20-40% growth inhibition of the marine algae, *S. pseudocostatum*, was observed upon exposure to effluents at the highest effluent concentration (100%; Ag and Ti exposure concentrations of 6 and 24 μg/L, respectively), with effluent from week 2 showing the strongest effect (40% growth inhibition relative to untreated control) (Figure 3). However, results from the DCFH-DA assay indicated that no formation of ROS occured for any of the tested effluents (SI; Figure S5). Exposure to the background effluent alone did not result in any significant effect on algal growth. These concentrations are below the respective no effect concentration (NOEC) values obtained for *S. pseudocostatum* in this study (1 mg/L and 10 mg/L for Ag and TiO₂ NPs). This suggests that the presence of solids and elevated NH₄ concentrations (3.3 mg/L) contribute to the observed effects and not just the total Ag and Ti present in the effluents (Table S2, Figure S8). Differences in toxicity of Ag NPs aged in crude and final wastewater have been reported and decreased toxicity was related to the sample physicochemical parameters and increased complexity⁴⁸.

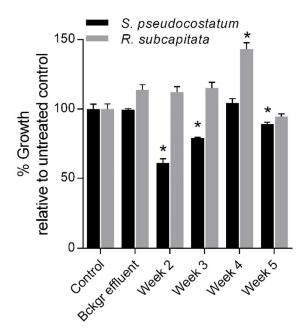


Figure 3. Percentage growth of *S. pseudocostatum* (black bars) and *R. subcapitata* (grey bars) exposed to effluents collected in weeks 2-5 (100% and 50% effluent concentration for *S. pseudocostatum* and *R. subcapitata*, respectively) and the background effluent. Algal growth inhibition was determined after 72 h of exposure. Asterisks denote statistical significance at p<0.05.

In contrast to the inhibitory effects of the effluent on *S. pseudocostatum* growth, there was evidence of hormetic effects in the freshwater algae, *R. subcapitata* exposed to effluent concentrations <50%. These effects were most apparent after exposure to effluent collected from week 4 and showed significant stimulatory effects on growth compared to the control (40% increase in growth compared to control) (Figure 3). The stimulatory effects in *R. subcapitata* growth were accompanied by a significant increase in the ROS formation (1.6-1.9-fold compared to untreated control) (SI; Figure S5) and increased cell aggregation (observed by microscopy,

data not shown). The ROS formation was positively correlated with the total Ag and Ti

concentration, total N and suspended solids present in the effluents (Figure S8). A similar
response of cell aggregation has been previously reported upon exposure of the green algae
Chlamydomonas reinhardtii to CuO-polystyrene core-shell NPs ⁴⁹ and Chlorella vulgaris and
Dunaliella tertiolecta to Ag NPs ⁵⁰ . It has been suggested that cell aggregation is a defense
mechanism that decreases the amount of exposed surface to xenobiotics. Moderate stress and low
ROS levels can lead to hormetic effects that can in turn induce the defense system ⁵¹ . The results
from the current study indicate that responses to the effluent exposure are species-dependent,
possibly due to differences in cell size, surface area and cell wall composition. Studies with
green algae and cyanobacteria exposed to Ag NPs have also shown differences in cell viability
and ROS response between species attributed to different biological properties and the
production of extracellular polymeric substances ⁵² . Moreover, the NP behavior depends on the
media composition that can result in different responses, TiO2 aggregates of 1369 nm were
observed in the presence of Cl in the higher ionic strength media of S. pseudocostatum compared
to 650 nm aggregates in R. subcapitata media while the Ag NPs seemed to be stable in both
media. The formation of insoluble AgCl(s) and dissolved silver chloride species depends on the
Cl/Ag ratio ⁵³ which could further explain differences in effects observed between the freshwater
and marine algae.
Effects of effluents on T. battagliai and D. magna. Exposure to effluents collected weekly
during the operation of the system led to a 20-45% increase in mortality of <i>T. battagliai</i> (at 100%)
effluent concentration), while no effect was observed from the background effluent (Figure 4A).
The highest significant mortality was observed upon exposure to effluents collected in weeks 2

and 5 (35 and 45% mortality compared to untreated control, respectively). Spiking the

background effluent with increasing concentrations of Ag NPs elicited a reduction in toxicity at
the lowest Ag NP concentration (0.08 mg/L) compared to pristine Ag NPs, but still caused a
significant increase in mortality at most concentrations (Figure 4B). Spiking the background
effluent also resulted in a 1.9x increase in the EC_{50} value compared to the pristine Ag NPs (0.09
and 0.17 mg/L, respectively) although the EC50 values were not statistically significant (Figure
4B). TiO ₂ NPs did not have any effect on mortality at any of the concentrations tested (0.01-10
mg/L).
Although the total Ag concentration in the effluents (5.99 $\mu g/L$ or 72.15 $\mu g/gSS$) exceeded the
NOEC for Ag NPs (0.005 mg/L), and was at a similar level to the EC_{10} obtained in this study
(0.0076 mg/L), no adverse effects on daphnid mobility were observed following 48 h exposure to
either the effluents or the background effluent. Spiking of the background effluent with
increasing concentrations of Ag NPs led to a significant decrease in mobility, but resulted in an
$16x$ increase in the EC $_{50}$ value compared to the pristine Ag NPs (0.16 and 0.0098 mg/L,
respectively) (Figure 5). TiO ₂ NPs did not affect daphnid mobility.

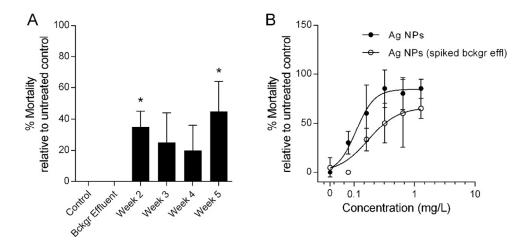


Figure 4. Percentage mortality of *T. battagliai* following 48 h exposure to (A) effluents collected in weeks 2-5 and (B) increasing Ag NP concentrations as received or spiked in the background effluent. Asterisks denote statistical significance at p<0.05.

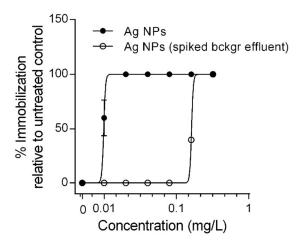


Figure 5. Percentage immobilization of *D. magna* juveniles following 48 h exposure to increasing Ag NP concentrations and Ag NP-spiked background effluent. Background effluent was collected during the system stabilization period (prior to spiking). No effects of effluents collected in weeks 2-5 and background effluent were observed.

A clear reduction in the toxicity of Ag NPs to D. magna was observed when exposed to the

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effluent collected from the lab-scale WWTP system (containing transformed Ag NPs) compared to pristine Ag NPs. Unlike D. magna, the marine copepod T. battagliai exhibited a clear response following exposure to the week 2-5 effluents (statistically significant in weeks 2 and 5). The difference in response between the two species may result from a combination of NP behavior in more complex WWTP effluents and differences in the feeding behavior of the two organisms. D. magna is a planktonic filter feeding organism⁵⁴ while T. battagliai is an opportunistic feeding epibenthic organism⁵⁵. Therefore, *T. battagliai* is likely to be directly in contact with particles associated with effluent solids that may settle out during the exposure period. T. battalgiai are non-selective grazers as well as filter feeders and feed on suspended particles along with detritus that settles out of the water column⁵⁶. These differences in feeding habit could explain the increased sensitivity of the copepods compared to daphnids when exposed to the WWTP effluent. In contrast to this D. magna was 10x more sensitive to pristine Ag NPs compared to T. battagliai (Figure 4 and 5). Therefore, the complete absence of effects in D. magna exposed to any of the collected effluents reinforces the idea that NPs present in the effluent are associated with the solids settling on the bottom of the vessels, reducing direct exposure and ingestion by the daphnids. To further confirm this, T. battagliai and D. magna were exposed to the background effluent spiked with increasing concentrations of Ag NPs which led to decreased toxicity relative to the pristine Ag NPs. However, for T. battagliai the EC₅₀ value only increased 2 times, whereas for D. magna the EC₅₀ value increased 16 times. This indicates the presence of solids in the effluent,

as well as the potential formation of precipitates, reduces the bioavailability of the Ag NPs to the

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daphnids compared to T. battagliai. This is in accordance with previous studies where reduced toxicity of AgNO₃ spiked into untreated effluent was observed for the freshwater green algae C. reinhardtii¹⁹ and the protective effects of background effluent were observed towards Cu interference with zebrafish hatching¹⁵. Furthermore, a decrease in the bioavailability of Ag from AgNO₃-exposed algae (C. reinhardtii) was observed in wastewaters, and suggested to be due to the presence of ligands¹². It has been previously demonstrated that sulfidation⁵³, the presence of natural organic matter⁵⁷ and thiol- or selenide-containing compounds such as cysteine⁵⁸ can reduce the Ag NP dissolution rate and lead to protective effects due to Ag⁺ complexation and decreased bioavailability^{59,60}, partially explaining the reduced toxicity of Ag NPs spiked in background effluent. The differences in EC50 increase trends of Ag NP-spiked background effluent compared to pristine Ag NPs between the 2 organisms can also be attributed to differences in media composition and ionic strength. The formation of AgCl precipitates in media with high Cl content such as in seawater can further impact the Ag⁺ availability and subsequent toxicity^{53,61}. Species-specific differences were related to the degree of Ag NP sulfidation, the exposure route and species sensitivity⁵³. Therefore, the effects of Ag NPs observed in the current study are considered organismdependent, with (epi)benthic organisms having the highest exposure risk due to directly ingesting sedimented and aggregated NPs or NPs bound to effluent solids. In addition, the media composition can impact the NP speciation and behavior leading to increased TiO₂ NP aggregation and formation of silver chloride species in media of increasing ionic strength. Effects of effluents on RTgill-W1 cells. The in vitro fish gill cell line model was employed in the current study as the gill is a key site for xenobiotic uptake and it is continuously exposed to water-borne contaminants⁶². Furthermore, the gills express enzymes involved in xenobiotic

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metabolism and transport. Exposure to the 1-5 week effluents did not cause a statistically significant decrease in the metabolic activity of RTgill-W1 cells in transwell inserts (Figure 6). A 40% decrease in the epithelial integrity (Figure 6), which coincided with a 2-fold increase in ROS formation (Figure S6), was observed upon exposure to effluent from week 3. However, no statistically significant effect was observed for any of the other effluents and no effect was observed for the "background" effluent for either endpoint. Previous studies have shown that primary fish gill cell cultures in permeable filter supports can tolerate apical water and varying osmotic conditions⁶³, river water⁶⁴, detect bioreactive metals^{64,65}, and have been used to study the uptake and transport of Ag NPs⁶⁶. In the current study, it has proven to be a good model system for whole effluent toxicity testing without the need for sample modification or alteration of the water chemistry prior to exposure. However, the concentrations of Ag and TiO₂ NPs measured in the effluent are considered too low to fully account for the effects observed in the metabolic activity and epithelial integrity assays. Given the complexity of the wastewater effluent, it appears that the combination of the presence of Ag NPs, ionic Ag and additional stressors such as NO₃ contribute to the overall response observed (Figure S8).

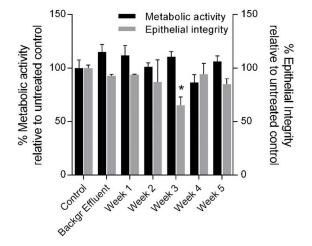


Figure 6. Percentage change in metabolic activity (left Y axis, black bars) and epithelial integrity (right Y axis, grey bars) of RTgill-W1 cells following exposure to effluents collected in weeks 1-5. Asterisks denote statistical significance at p<0.05.

As effects were observed on the epithelial integrity, and because the gill is a site of xenobiotic uptake and detoxification, the effects of the effluents on the gene expression of zonula occludens-1 (ZO-1) tight junction protein and multixenobiotic resistance genes in RTgill-W1 cells were studied. The ZO-1 gene was selected due to the decreased epithelial integrity observed in the paracellular permeability assay. Results showed ZO-1 mRNA levels were elevated after exposure to effluents collected on week 1 and 3 (SI; Figure S7). Previous studies have shown that the RTgill-W1 cells express functional tight junctions that can respond to certain modulators⁶⁷. In the current study, the RTgill-W1 cell model in transwell inserts showed an increased paracellular permeability followed by an increase in ZO-1 expression upon exposure to week 3 effluent, suggesting an impact on the epithelial integrity and a compromised barrier function. Moreover, the DCFH-DA assay indicated exposure to the week 3 effluent led to a 2-

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fold increase in ROS formation, suggesting a ROS-induced compromised epithelial integrity. It has previously been shown that oxidative stress can lead to a disruption of tight junctions in MDCK canine kidney cells⁶⁸. The multixenobiotic resistance (MXR) mechanism mediated by ATP binding cassette transporters is an important mechanism of defense against xenobiotics, which functions by extruding them or their metabolites out of the cell. The transporters are localized in tissues with a barrier function or involved in secretion and absorption, they transport a wide variety of compounds across cell membranes and it has recently been shown that NPs, including Ag NPs, can interfere with the MXR system^{69,70}. Due to their importance in cellular defense against xenobiotics, the multixenobiotic resistance genes ABCB1, ABCC1, ABBC2 were also investigated in the current study. Exposure to the effluents led to increased mRNA levels of ABCB1, ABCC1 and ABCC2 transporters, with ABCB1 (the most responsive) exhibiting increased expression levels in response to effluents from weeks 1-3 (3.4-fold increase upon exposure to effluent week 2) (SI; Figure S7). These results indicate an interference with the defense mechanism and potentially compromised protection against xenobiotics. The contribution of other trace elements and other unidentified stressors present in the effluent to the observed effects cannot be excluded. It also remains to be determined whether this observed change in gene expression also leads to transporter functional changes. **Environmental implications**. The combination of a lab-scale WWTP with detailed fractionation approaches, characterization techniques (TEM, sp-ICP-MS, sequential filtration/ICP-MS), a battery of marine and freshwater bioassays and an *in vitro* gill cell line model allowed the effects of transformed NPs to be investigated. This study shows that Ag NPs are transformed through simulated biological WWTP processes to particles associated with S, Cu and Zn. The resulting

hazard cannot be predicted based on exposures made in simplified media or determined by measuring the NP concentration and the dissolved fraction since the effluent is complex with additional stressors (e.g suspended solids, NH₄) either exacerbating or mitigating the effects depending on the organism, endpoint and media used. The transformed particles appeared to have a greater impact on epibentic copepods suggesting that they were still bioavailable despite their transformation. Differences in responses in marine vs freshwater algae and crustaceans highlight the importance of the media composition in the NP speciation that can lead to speciesspecific responses. The study reinforces the need to use multiple test species representing different environments and exposure routes, bioassays and endpoints to gain clearer understanding of the potential hazards of low level realistic concentrations of transformed nanomaterials and multiple stressors in environmental media of increased complexity. The results highlighting the difference in toxicity of pristine and transformed particles, emphasize the need for future studies using a broader range of weathered or transformed NPs in relevant exposure scenarios to provide a more accurate understanding of their potential impacts. The combination of complementary analytical techniques (TEM, sp-ICP-MS, sequential filtration/ICP-MS) was useful for the detection and characterization of low NPs concentrations in complex environmental matrices. Our results demonstrated that Ag and TiO₂ NPs show a strong association with solids, suggesting the potential for terrestrial organisms' exposure through biosolid^{21,42,71} application. Based on these conclusions future studies should focus on the effects of transformed NPs associated with the biosolids on terrestrial organisms and the factors contributing to species-specific responses.

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ASSOCIATED CONTENT

Supporting Information . Additional information is provided for the synthetic wastewater
composition, the lab-scale WWTP description and operation (and schematic; Figure S1), sample
preparation description for STEM/EDS and sp-ICP-MS, mass balance calculations for Ag and
TiO ₂ NPs, DLS measurements of TiO ₂ and Ag NPs stock dispersions in MilliQ water, synthetic
wastewater, seawater and exposure media, sp-ICP-MS measurements of NP stock dispersions
and effluents (Table S1), characteristics of the effluents collected in weeks 1-5 (Table S2) and an
overview of genes, primer sequences and protocol used for qPCR (Table S3). In addition, TEM
images of Ag and TiO ₂ NPs stock dispersions are provided (Figure S2, S3), fractionation of Ti
(Figure S4), effects of effluents on S. pseudocostatum and R. subcapitata ROS formation (Figure
S5), effects of effluents on RTgill-W1 ROS formation (Figure S6), gene expression (Figure S7)
and principal component analysis (PCA) of the physicochemical parameters and effects observed
in the different bioassays (Figure S8).

AUTHOR INFORMATION

Corresponding Author

* Anastasia Georgantzopoulou. E-mail: <u>anastasia.georgantzopoulou@niva.no</u>. Tel: +4798227741

Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

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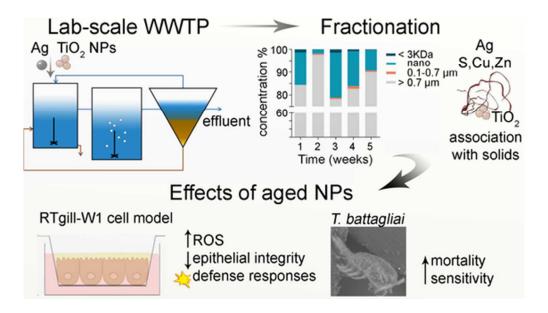
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