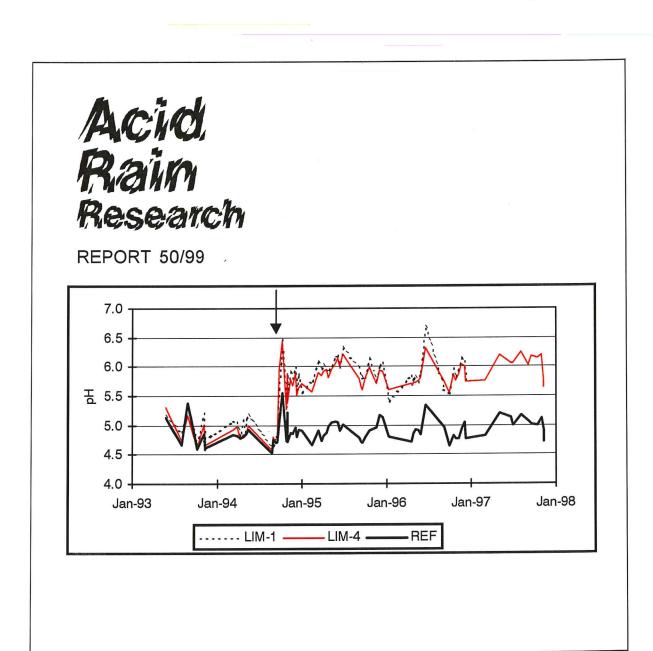


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Whole-catchment Application of Dolomite to an Acidified Forest Ecosystem in Gjerstad, Southern Norway



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Abstract

As part of the research programme "Counteractions Against Acidification in Forest Ecosystems" (Miljøtiltak i skog), a coniferous-forested catchment was limed with dolomite in September 1994. 3 t ha⁻¹ of coarse dolomite powder were spread on 0.8 km² by helicopter. The liming resulted in an immediate improvement in runoff water quality relative to an adjacent reference catchment. pH, Ca, Mg and ANC (acid neutralising capacity) increased and inorganic Al decreased. Favourable water quality was maintained for 3 years. NO₃-concentrations increased the second year after liming, whereas concnetrations of total N and TOC were not significantly changed. Liming did not affect concentrations of 7 trace metals (As, Cd, Cu, Fe, Ni, Pb, Zn) whereas concentrations of Mn, Co, and Zn decreased. Only minor changes in soil solution, and only at some of the lysimeters were detected. Steep slopes, thin soils, high amounts of precipitation and thus dominance of surface and subsurface flow in this catchment may explain the rapid response in runoff. During the first three years after liming there have been no significant effects on tree growth and vitality (crown density and crown colour). This experiment shows that liming of forested catchments is a viable method to obtain long-term improvement in water quality and positive effects for acidsensitive aquatic organisms.

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Counteractions Against Acidification in Forest Ecosystems

Whole-catchment Application of Dolomite to an Acidified Forested Ecosystem in Gjerstad, Southern Norway

Preface

The research program "Counteractions against acidification in forest ecosystems" (Miljøtiltak i skog) was initiated in 1991 as a Norwegian Forest Research Institute (NISK) program with funding from the Ministry of Agriculture. In 1993 the program was reorganised and a five-year research program led by NISK was started. Three co-operating institutes, NISK, the Norwegian Institute for Water Research (NIVA) and the Norwegian Institute for Nature Research (NINA), were main actors. Later on the Agricultural University participated.

The program was divided into an initial literature study and five separate research activities thereafter. This report gives background information and summarises the results from sub-project IV, the whole-catchment application of dolomite to an acidified forested ecosystem in Gjerstad, southern Norway.

We would like to thank the forest owner Olav Fjærbu for careful sampling and maintenance of monitoring stations during the project period and for offering his forest as a research site in an early phase of the program.

This work has been financed by the Ministry of Agriculture, the Ministry of Environment, the four counties Rogaland, Vest-Agder, Aust-Agder and Telemark and NIVA.

Grimstad, September 16, 1999

Atle Hindar

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Summary

As part of the research programme "Counteractions Against Acidification in Forest Ecosystems", a forested catchment dominated by pine (Pinus sylvestris) and Norway spruce (Picea abies) was limed with dolomite in September 1994. 3 t ha⁻¹ of coarse dolomite powder were spread on 80 ha by helicopter. Mean pre-liming stream water qualities (n=16 or 18) at four independent stations were in the range: pH 4.6 - 4.9; $0.85 - 1.28 \text{ mg L}^{-1}$ Ca; $0.30 - 0.41 \text{ mg L}^{-1}$ Mg; $194 - 275 \mu \text{g L}^{-1}$ reactive Al (Alr) and 34 - 103 μg L⁻¹ inorganic monomeric Al (Ali). Dolomite application resulted in an immediate increase in runoff water quality relative to an adjacent reference catchment. For the period September 1994-December 1996 mean values at the main outlet of the limed catchment were: pH 5.86; 1.51 mg L⁻¹ Ca; 0.81 mg L⁻¹ Mg; 15 μg L⁻¹ Ali. A small increase in NO₃ was found at the main outlet of the limed catchment the second winter after liming. Total N and TOC were not changed. Mn, Co and Zn showed significantly decreasing trends when compared to the reference stream, whereas Fe, Pb, Cd, Cu As and Ni remained relatively unaffected by the dolomite application. Cr was mostly below detection limits. Only minor changes in soil solution, most clearly for Mg, could be detected after liming. Steep slopes, thin soils, high amounts of precipitation and thus dominance of surface and subsurface flow in this catchment may explain the rapid response in runoff. During the first three years after liming there have been no significant effects on tree growth and vitality (crown density and crown colour). This experiment shows that liming of forested catchments is a viable method to obtain long-term improvement in water quality and positive effects for acid-sensitive aquatic organisms.

1. Introduction

1.1 Background

In areas in southern Norway characterised by widespread fish-death due to acidification of surface waters, so far no negative effects of acid rain on forest health can be documented (Aamlid *et al.* 1998). Forests tend to remove base cations, scavenge air pollutants and concentrate pollutants due to increased evapotranspiration (Jenkins *et al.* 1990). In the long run, acidification of forest soils may therefore be expected in vulnerable areas if the soil has a limited base cation supply. The supply of nitrogen from atmospheric deposition may stimulate forest growth and thereby also the loss of base cations from the soil. Although atmospheric deposition of sulphur in Norway has decreased during the last decade, the situation for the forest ecosystems is still uncertain, and the need to examine various counteractions against acidification of the soils has been recognised.

Episodic acidification, enhanced by sea salts, may be particularly important for sensitive aquatic organisms in forested areas (Hindar *et al.* 1994; 1995a). Afforestation may have undesired impacts on aquatic ecosystems, and the question of whether afforestation should be augmented with measures to prevent this extra acidification has been raised.

Several countermeasures against forest soil acidification are available for forest management practices, including both silvicultural measures and application of chemicals. The first group involves use of broadleaved trees, both in pure stands or mixed within conifer stands (Frank 1994).

Changes in forestry management may reduce the removal of base cations but will not prevent strong acid anions from entering soil solution and runoff. Strong acid anions, especially SO_4 , are the driving force in soil acidification when balanced by base cations (BC, mainly Ca and Mg). Poorly buffered soils have a very limited capacity of base cation production because of the low weathering rate. In such soils strong acid anions may also be balanced by acid cations, especially H^+ and $Al(OH)_n^{(3-n)+}$.

A second group of countermeasures includes application of carbonates (liming), ash and various commercial fertilisers. The addition of carbonates as calcitic and dolomitic limestone increase the base saturation of the soils and thereby prevent leaching of H⁺ and toxic Al-species. Liming may thus be regarded as desirable for both forest soils and aquatic systems in acidified areas, and has been practised for a long time as a compensatory or ameliorating measure in forest soils (Hüttl and Zöttl 1993; Kreutzer 1995) and in acidified lakes and rivers (Olem 1991; Henrikson *et al.* 1995). Fertilisation with the goal of revitalisation of forest stands suffering from nutrient deficiencies has also proved successful (Hüttl 1991).

The question of whether liming should be launched on a practical scale in Norwegian forestry has been raised several times during the last decade (Nilsen *et al.* 1994). The inter-institutional research program "Counteractions against acidification in forest ecosystems" (Miljøtiltak i skog) was initiated in 1991. Main goals for the program were to:

- Give a status and evaluate potential and practical measures, such as liming, vitality fertilisation, use of broad-leaved trees and other silvicultural methods, that may counteract negative effects of acid deposition in forests.
- Perform research on the effects of selected measures on trees, ground vegetation, animal life, forest soils, soil solution and the runoff water quality.
- Give necessary basis for making decisions about future activities on a practical scale.

1.2 Definitions and targets

Liming is the addition of carbonates, oxides and hydroxides of calcium and magnesium to terrestrial and aquatic ecosystems to neutralise soil or surface water. This is consistent with the definition of Olem (1991), although he included all bases. Operational liming of forest soil has gradually changed towards dolomitic materials in Germany (Feger *et al.* 1995) due to the anticipated positive effect of Mg supply to prevent Mg deficiency. Dolomite and calcite mixture was also recommended for operational liming in Sweden (Lindström *et al.* 1993). However, some of the Swedish experiments were carried out using calcitic limestone, i.e. products composed of CaCO₃ with only a few percent Mg.

The targets for terrestrial liming operations may be many, e.g. the soil pH, the soil (Ca+Mg)/Al ratio, tree growth, forest health, stream water pH, stream water inorganic Al, a trout population and so on. Various application strategies may be used to achieve one or more of these goals (Henrikson *et al.* 1995), and lime quality, lime dosages, and application strategies may vary. If the goals are clarified, reasonable liming strategies may be recommended, if the potential effects are adequate and acceptable.

The main targets for forest soil liming should be to avoid the negative effects on forest soils of strong acid inputs and to avoid unwanted effects on the terrestrial and aquatic ecosystems. Another target may be improved living conditions for fish in the stream water draining the limed forest. If all the potential positive effects of forest liming are achieved all the negative effects may also occur, see section 1.3. If no undesired effects can be accepted, no positive effects may occur either. The real challenge is to maximise the positive and minimise the negative effects.

1.3 Potential effects of application of dolomite

Moderate addition of dolomitic limestone to the topsoil of forest ecosystems may be intermediate in both positive and negative effects. In this report the main focus is on reduced toxicity towards aquatic organisms.

1.3.1 Soil solution and stream water

The added dolomite gradually dissolves and releases Ca and Mg that can be adsorbed in the organic layer of the top soil. Vertical movement of these base cations to deeper soils then takes place, but this is a slow process, and may occur in the range of 1 cm year⁻¹ (Brahmer 1994). Fast changes in the chemical composition of soil solution are therefore not likely. The first change may be increased concentration of Mg in the soil solution due to the more mobile nature of this element relative to Ca (Brahmer 1994). On the other hand dolomite often contains almost four times as much Ca on a weight basis, thereby counteracting the dominance of Mg-increase over Ca-increase in the soil solution.

Although the change in soil solution may be slow, increased amounts of base cations in the topsoil may be important for runoff water quality when hydrological events are characterised by overland or subsurface flow. Overland and subsurface flow is promoted by thin soils and steep slopes, typical of acidified areas in Norway. High amounts of precipitation increase the likelihood for a high water table and saturated soils. Dominance of a more lateral movement of the runoff and reduced residence time of rainwater in the soil may be the result. Under such circumstances runoff acidity may be exchanged with the added base cations in the top soils, thus increasing the concentration of Ca and Mg and also increasing pH.

Dolomite addition thus may result in several positive effects. Increased base saturation of the topsoil, eventually also at greater depths in the soil, and increased pH will reduce Al-mobilisation. This in turn decreases the leaching of toxic Al to soil solution and surface water. Also, increased concentrations of base cations should be expected. If no significant changes in SO₄ and NO₃ occur, liming with dolomite will therefore result in increased ANC (acid neutralising capacity).

In acidified regions low pH and high Al-concentrations in runoff may occur during episodes caused by sea salt inputs (Hindar *et al.* 1994; 1995a), snowmelt and other hydrological events. Increased base saturation in the forest soil and increased pH of the runoff after liming may reduce this Al transport, both by decreasing the preferential ion exchange with H⁺ and Al in acid soils and by polymerising and precipitating ion exchanged Al in the overland flow. If forest liming in such areas reduces the Al transport and promotes terrestrial polymerisation of Al, the magnitude and frequency of episodes with toxic water will be reduced and mixing zones in streams and rivers characterised by unstable Alchemistry (Rosseland *et al.* 1992) may be avoided.

Undesirable chemical effects of terrestrial liming include increased NO₃-leaching and mobilisation of trace metals (Matzner *et al.* 1985; Persson *et al.* 1989; Hüttl and Zöttl 1993; Kreutzer 1995). This may partly be due to increased mineralisation of the surface organic (humus) matter, and may effect soil solution and surface water.

The liming-mediated change in the top soil pH may stimulate the bacterial degradation of organically bound N in the humic material to NH₄ due to a more favourable environment for these organisms (Kreutzer 1994). Nitrification, the bacterial mediated conversion of NH₄ to NO₃, may be stimulated and lead to decreased pH. If the supply of NO₃ is already in excess due to atmospheric deposition, the result may be increased leaching of NO₃ to the soil solution (Kreutzer 1995) and probably to runoff, with subsequent increased acidification and possibilities for increased supply of N to coastal areas. N leaching may thus reduce the deacidification effect of the dolomite in runoff. In less N-polluted areas, NO₃ released will probably be stored in the soil and consumed by microorganisms and terrestrial vegetation.

Mineralisation may result in increased mobilisation of organically-complexed trace metals (especially Cu, Pb and Fe) and result in elevated concentrations in the soil solution. This may be especially so if the deposition rate of the metals is high, and they have accumulated in the surface soils, as in the Höglwald experimental area in Germany (Kreutzer 1995). Solubility of Cd and Zn (and Cu) is pH dependent and these metals will probably not be transported out after liming. On the contrary, increased pH may result in increased retention. Dissolved inorganic Mn and Cd may be oxidised and precipitated at increased pH and decrease in the soil solution after liming (Kreutzer 1995).

The deposition of Cu in Norway is mainly due to local sources and is generally low (Berg *et al.* 1996). Deposition of Zn and Pb in 1996 was an order of magnitude higher than other long-range transported metals but has decreased by 60-80 % in the period 1978-1996 (Tørseth and Manø 1997). This reduction is also true for Cd. Compared to the situation in Germany deposition of trace metals in Norway is low.

1.3.2 Aquatic organisms

Forest liming, according to the Swedish concept (Lindström *et al.* 1993) tends to result in a relatively moderate water quality change in the runoff. Exceptions may be found in areas heavily polluted with nitrogen or trace metals, as referred to above. An important question is: will these changes in any way affect aquatic organisms?

Several investigations have documented that even small water quality changes or differences may be of significance for the survival of fish (Staurnes *et al.* 1993; Dalziel *et al.* 1995; Kroglund and Staurnes 1997) and invertebrates (Raddum and Fjellheim 1984). Decreased concentration of the toxic inorganic Al-species is probably most important for sensitive aquatic organisms and increased concentration of Ca may ameliorate Al toxicity (reviewed by Wood and McDonald 1987; Rosseland *et al.* 1990). Increased concentration of dissolved organic matter will detoxify a larger fraction of Al at low pH due to complex binding. Forest soil liming may therefore represent an effective liming strategy for aquatic systems if significant changes in these parameters occur.

1.4 The whole-catchment liming project in Gjerstad

The question whether liming should be launched on a practical scale in Norwegian forestry may be answered if adequate results exist. Due to lack of integrated studies on soil, water and vegetation under Norwegian conditions, a catchment study with dolomite application was initiated in 1993. The project in Gjerstad, which is presented in this report, is one of five projects under the research program.

The purpose of the experiment was to study the effects of dolomite application to a forested area on soil solution chemistry, runoff water chemistry, soil chemistry, vegetation, tree growth and vitality. Few research programmes have focused on the effects of forest liming on both soil solution, stream water and the significance of water quality changes for aquatic organisms. One of the main purposes of the activities in the Gjerstad-project is to link these topics.

Results from this experiment have been published as part of the Acid Reign'95 Conference in Gothenburg (Hindar *et al.* 1995b) and in annual and final reports in Norwegian from the research programme. This report presents the results in more detail and covers the pre-liming and post-liming periods (until end of 1997).

2. Site description and liming

2.1 Site selection

Several possible areas were visited prior to the establishment of the project sites. The catchment was chosen to satisfy the following criteria:

- acidified due to long-range transported air-pollutants; pH in runoff in the range 4.5-5.5 and markedly elevated Al-concentrations
- mixed forest stands, typical for the moderately productive forests of the area
- mixed, but not too poor, growing conditions (nutrient and water availability) for ground vegetation
- not too close to the sea, due to possible special effects related to sea-salts
- two defined catchments with brooks suited for sampling and monitoring of runoff, one for liming operation and one as reference.

A mixed forest in Gjerstad, Aust-Agder County, southern Norway (58° 53' N, 9° 00' E) was chosen for investigations of the effects of forest soil liming. Fugleliåsen (denoted LIM and FU; 84 ha) was selected as liming area and Spjøtåsen (denoted REF and SP; 41 ha) as reference. The limed catchment was further divided in 3 by the sampling stations LIM-1 (40 ha), LIM-2 and LIM-3, whereas LIM-4 is the main sampling station (**Figure 1**).

2.2 Forest cover

The catchments are forested with a mixture of mainly Norway spruce (*Picea abies*) and Scots pine (*Pinus sylvestris*) stands. A relatively small amount of broad-leaved trees are mixed within the conifer stands. The forest consists of old mature stands and with a certain amount of newly regenerated areas (less than 15 years).

Table 1. Catchment areas (ha), standing volume (m³), mean age (yr) and area distribution according to tree age classes and impediment.

	Size	Standing volume (m ³)			Mean	Middle aged	Regenerated	Imped.
	(ha)	Total	Spruce	Pine	age (yr)	+ old forest	areas	_
LIM	84.4	9900	3400	4500	67	70 %	17 %	13 %
REF	40.8	6000	1600	3600	77	84 %	9 %	7 %

2.3 Precipitation and deposition

Mean annual precipitation is about 1200 mm and mean annual discharge about 900 mm (29 L s⁻¹ km⁻²). Mean total wet + dry deposition of S and N in the project period 1994-1996 at the Birkenes monitoring station 70 km SW of Gjerstad was 0.90 g m⁻² and 1.52 g m⁻², respectively. At the Solhomfjell monitoring station 20 km NW of Gjerstad the mean wet deposition of S and N was 0.61 g m⁻² and 0.91 g m⁻², respectively, in the same period. Deposition of trace metals at Solhomfjell in 1996 was as follows (numbers in mg m⁻² yr⁻¹): 4.95 Zn; 2.01 Pb; 0.82 Cu; 0.26 Ni; 0.18 As; 0.16 Cr; 0.042 Cd; 0.02 Co (all deposition data from Tørseth and Manø 1997).

2.4 Liming

In September 1994 a total of 240 tonnes of coarse-grained dolomite was spread over the Fugleliåsen catchment, except for two small (about 1000 m^2) ponds, by helicopter. This gives a mean dose of approximately 2.9 t ha⁻¹. The grain-size distribution of the dolomite was 10 % > 1.7 mm; 90 % > 0.18 mm. The Ca, Mg and water content were 23, 12 and 1 % by weight, respectively.

3. Methods

3.1 Hydrology and water chemistry

3.1.1 Discharge

The limed catchment was divided into three sub-catchments (**Figure 1**), of which two (sampling stations LIM-1 of one of the sub-catchments and LIM-4 at the outlet of the limed catchment) have been included in the water chemistry monitoring in the whole project period. Monitoring in LIM-2 and LIM-3 of the two other sub-catchments were terminated in the summer of 1995.

Water flow was recorded beginning 26 May 1993 at 15-minute intervals at calibrated 90° (at LIM-1) and 120° (at LIM-4 and REF) V-notch weirs.

3.1.2 Runoff water sampling

Volume-weighted stream water samples were collected biweekly beginning 26 May 1993 at LIM-1, LIM-4 and REF. Point samples were collected biweekly at the other two stations in the limed catchment. Water samples were sent to NIVA by post and were analysed 2-3 days after sampling.

Samples for analysis of trace metals were taken directly in 100-ml polyethylene bottles. Before sampling the bottles were washed in 3% HNO $_3$ for at least 24 hours and thereafter rinsed with demineralised water. After the washing procedure the PE-bottles were refilled with demineralised water until sampling. Before sampling the PE-bottles were emptied for demineralised water and immediately filled with the sample.

3.1.3 Chemical analyses – water and soil solution

Chemical analyses of all major ions and Al-fractions were carried out at NIVA according to standard procedures (**Table 2**). Trace metal analyses were performed with ICP-MS at the Norwegian Institute for Air Research (NILU). In the event that concnetrations were reproted below the detection limit, these were set at ½ the detection limit.

3.1.4 Calculations and statistical analyses

Ali (inorganic monomeric Al) is defined as the difference between Alr (reactive Al) and Alo (organic monomeric Al). Organic N is defined as the difference: total N - $(NO_3-N + NH_4-N)$.

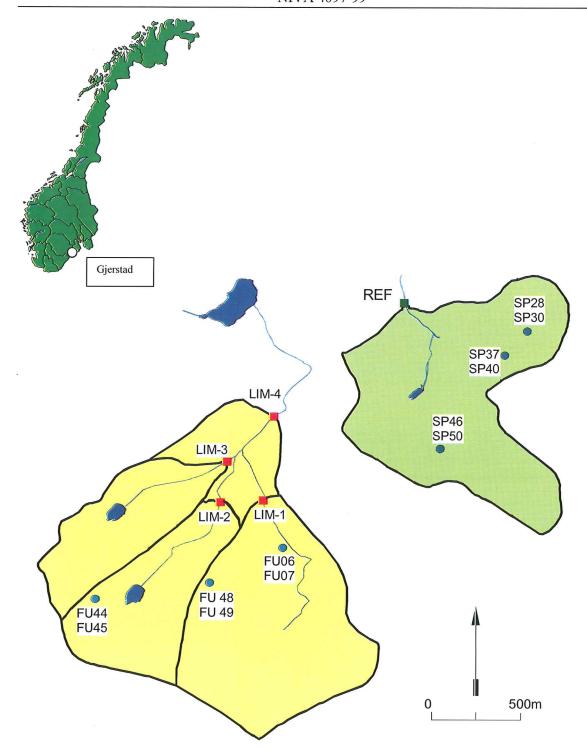


Figure 1. Situation of the limed and reference catchments in Gjerstad. Stations for surface water sampling (LIM-1, LIM-2, LIM-3, LIM-4 and REF), soil chemistry and soil solution sampling (FU and SP stations) are shown. Runoff is registered continuously at LIM-1, LIM-4 and REF.

Table 2. Chemical analysis methods used at NIVA (major components) and NILU (trace metals). ICP=Inductive coupled plasma emission spectrometry; ICP-MS= Inductive coupled plasma emission spectrometry combined with mass spectrometry; FIA=flow injection analysis.

Parameter	Parameter name	Unit	Analytical method
pН	рН	- log [H ⁺]	Potentiometric
Cond	Conductivity	mS m ⁻¹ at 25 °C	Electrometric
Ca	Calcium	$mg L^{-1}$	ICP
Mg	Magnesium	mg L ⁻¹	ICP
Na	Sodium	mg L ⁻¹	ICP
K	Potassium	$mg L^{-1}$	ICP
Cl	Chloride	mg L ⁻¹	Ion chromatography
SO_4	Sulphate	mg L ⁻¹	Ion chromatography
NO_3	Nitrate	μg N L ⁻¹	Automatic colorimetry
$\mathrm{NH_4}$	Ammonium	μg N L ⁻¹	Automatic colorimetry
Alk	Alkalinity	μeq L ⁻¹	Potentiometric titration
		1	to pH 4.5
TOC	Total organic carbon	mg L ⁻¹	Oxidation to CO ₂ and
Alr	Reactive aluminium	μg L ⁻¹	then IR detector Automatic colorimetry
Alo	Non-labile aluminium	μg L ⁻¹	Automatic colorimetry
7110	ron-idone aranimum	μgL	after ion exchange
SiO_2	Silica	mg L ⁻¹	Photometry (FIA)
Tot N	Total nitrogen	μg L ⁻¹	Automatic colorimetry
Tot P	Total phosphorus	μg L ⁻¹	Automatic colorimetry
As	Arsenic	$\mu g L^{-1}$	ICP-MS
Cd	Cadmium	$\mu g L^{-1}$	ICP-MS
Co	Cobalt	$\mu g L^{-1}$	ICP-MS
Cu	Copper	μg L ⁻¹	ICP-MS
Fe	Iron	μg L ⁻¹	ICP-MS
Mn	Manganese	μg L ⁻¹	ICP-MS
Ni	Nickel	$\mu \mathrm{g} \mathrm{L}^{ ext{-}1}$	ICP-MS
Pb	Lead	$\mu g L^{-1}$	ICP-MS
Zn	Zinc	μg L ⁻¹	ICP-MS

ANC (Acid neutralising capacity), in μ eq L^{-1} , is defined as: ANC = $(Ca^{2^+} + Mg^{2^+} + Na^+ + K^+)$ - $(Cl^- + SO_4^{2^-} + NO_3^-)$ if $NH_4^+ \approx 0$. All concentrations are in μ eq L^{-1} .

Non-marine Na (Na*) in μeq L⁻¹ is calculated as:

 $Na^* = Na^+ - 0.859 * [Cl^-]$, assuming that Cl is entirely derived from sea salts. If the Na/Cl ratio departs from the ratio in the sea (0.859), this indicates sources or sinks of Na in the catchment.

Net change in transport of Ca and Mg from the limed catchment is calculated from water flow and the difference between simultaneously taken samples from LIM-4 and REF. This difference was then corrected for the mean pre-liming difference between the two catchments. This correction was +0.14 mg L^{-1} for Ca and -0.01 mg L^{-1} for Mg.

The data material was tested statistically by different methods:

Runoff water samples;

- 1. Systematic pre-liming variability between sub-catchments was tested with simple ANOVA. The tests were performed on the means of calculated differences of simultaneously taken samples at independent (no autocorrelation) LIM-stations and the REF-station. This difference, which reduces the contribution of variance from seasonal variability, is referred to as e.g. Ca-diff.
- 2. Changes due to liming were also tested by use of this dataset and ANOVA.
- 3. Trends after, and due to, liming were tested by use of simple regression on the differences. General trends, as reflected in the reference catchment, were thereby eliminated.
- 4. Changes in water chemistry in the limed catchment (LIM4) relative to the reference catchment (REF) at the point of time of liming (September 1994) were tested by Random Intervention Analysis (RIA) (Carpenter *et al.* 1989). RIA compares differences in concentrations in paired, chronologically-ordered samples for the pre- and post-liming periods with the change obtained from 1000 randomly-ordered sets of differences.

3.2 Soil and soil solution chemistry

3.2.1 Soil sampling

Soil were sampled in summer 1994 before liming at the 6 plots (3 each in the limed and reference catchments) used for analysis of ground vegetation (**Figure 1**). At each plot 1 sample was collected from the humus horizon and 2-3 samples from the mineral soil horizon.

3.2.2 Chemical analyses – soils

Soils were extracted in water for measurement of pH and in 1 M NH₄NO₃ for determination of exchangeable cations. Bulk density, loss on ignition, and Kjeldahl N were measured on bulk soil. Contents of 31 elements was determined by ICP (Inductive coupled plasma emission spectrometry) on acid digests of the bulk soil. Analyses were performed at NISK using standard procedures.

Cation exchange capacity (CEC; NH₄NO₃ extract) was calculated as the equivalent sum of H⁺, Al, Ca, Mg, Na, and K. Base saturation (%) is defined as the sum of (Ca+Mg+Na+K)/CEC.

3.2.3 Soil solution sampling

60 Prenart lysimeters were installed prior to liming (August 1994) in the 6 locations selected for vegetation analyses (**Figure 1**), at 3 sites in Fugleliåsen (FU) and 3 in Spjøtåsen (SP); two lysimeters for each sampling depth. The samples were collected following a 24-hour evacuation period. The soil solution was collected from 5 and 15 cm depth and stored in 2 litre Prenart bottles, which were covered to minimise reaction catalysed by sun light or alga growth. The soil solution samples were filtered through 2 μm membranes and stored for less than 48 hours prior to analysis at NIVA. Sampling began in September 1994. There are no pre-liming samples of soil solution.

3.3 Forest inventory

Forest inventories were carried out during summer 1994 in both limed and reference areas. The inventory was made as a systematic sample plot inventory. Lines were laid out with individual distances of 100 m in the forest and along each line sample plots were placed with 50-m distances (**Figure 2**). The plots were circular with an area of 200 m². Plots on mineral soil were permanently marked with a wooden pole in the centre. Plots that fell on peat-land, roads, water or barren rocks were classified as non-productive land. Plots that fell in boarder areas between stands were systematically moved 10 meters in one direction. A total of 122 (limed area) and 56 (reference area) plots were measured.

3.3.1 Tree measurement

In each sample plot all trees more than 5 cm in diameter at breast height were measured by diameter and permanently marked with tree number. Tree species were noted.

Sample trees were picked out with relascope. Relascope factor 2 was used and each third of the trees falling within the relascope were picked out for further measurements. These sample trees were measured for height above ground, height to living crown and different kinds of damages were noted according to a manual. Sample cores were taken from all sample trees and annual ring width was measured for the latest 40 years. From each sample plot, where possible, the two largest trees (by diameter) without damages were picked out for site index estimation. In addition to the variables already mentioned, bark thickness was measured and annual height growth visually estimated on these trees. All annual rings were measured for breast height age determination on these trees. The annual rings were measured in laboratory.

3.3.2 Site index, volume- and cutting class calculations

Site index for each sampling plot was estimated from age at breast height and height of the two largest trees (Tveite 1977). Curves for diameter-to-height relationship for each tree species were constructed for the two areas together. Tree height was then estimated for all trees. Volume was calculated using functions for spruce, pine and birch (all broad-leaved trees are treated as birch in this connection) (Braastad 1966, Vestjordet 1967, Brandseg 1967) with diameter and height as independent variables. Cutting class (for definition see text in **Figure 24**) is dependent on site index, species and age and the area distribution of cutting classes have been calculated from the sample plot data for the two areas.

3.4 Intensive monitoring plots

In each of the two areas, 10 sample plots (25 x 30 m) were subjectively laid out (**Figure 3**). The criteria for placing the different plots were to cover the vegetation gradient in the limed catchment, and a comparable gradient in the reference catchment. The gradient varied from dry to wet and from fertile to infertile soils. Within each sample plot an area of 5×10 m was used for intensive registration of vegetation (Eilertsen *et al.* 1994). On each plot steepness and exposition were measured.

3.4.1 Registration in the plots

All trees more than 5 cm in breast height were measured by diameter and permanently marked with a number. Height and crown height was measured on about half of the trees in 1994. Height increment was visually estimated on the trees and bark thickness was measured. In 1998 height was just measured on ¼ of the trees. Damages were also noted for these trees.

The 3 largest trees by diameter (spruce or pine) were cored to pith, and age and annual ring width were measured in the laboratory.

Each autumn, diameter was measured on all trees and tree vitality evaluated. The vitality evaluation is based on tree crown density and colour. The crown density (%) is defined as the amount of needles in tree crowns compared to what could be expected on a sound tree in that particular area. The crown colour is an overall impression of the colour based on a scale from 1 to 4: 1 is normal green, 2 slightly yellow, 3 average yellow and 4 strongly yellow. The result from the evaluation is based on the same trees each year and the work has been done by the same person and follows the same criteria as for the Norwegian forest monitoring programme (Groeggen 1997).

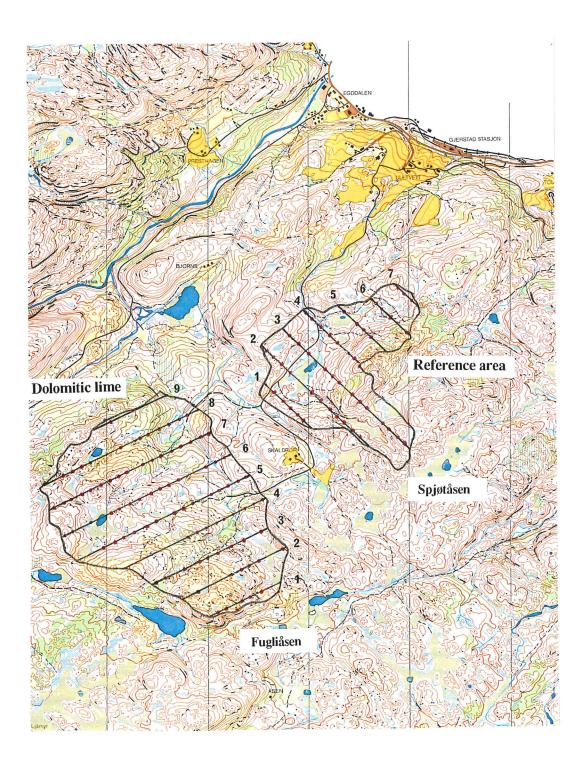


Figure 2. Map showing the two catchments and the sampling lines in each area. Red dots indicate plots where trees were measured. Scale approx. 1:20000.

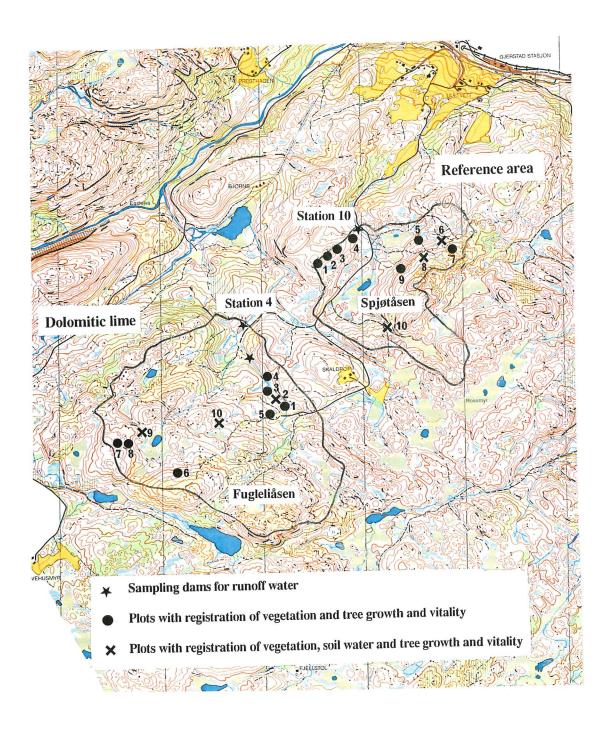


Figure 3. The location of the intensive monitoring plots. All plots were used for vegetation investigations and some for soil solution sampling. Scale approx. 1:20000.

3.4.2 Statistical analysis

Based upon height and diameter the volume of the sample trees was estimated using functions for different tree species (Braastad 1966, Vestjordet 1967, Brandseg 1967). A linear regression between diameter and volume on these trees was used to estimate volume for all trees. The function developed from the 1994 registration was also used for the 1998 data. Volume increment in the 4-year period was calculated as the difference in standing volume between the two years.

Site index (Tveite 1977) was estimated on the basis of age and height of the 3 trees per plot. Site index curves for the dominant tree species was used to determine the site index of the plots.

The increment in the 4-year period from 1994 to 1998 was compared to increment functions developed for Norwegian conditions based on increment investigations on permanent sample plots (Blingsmo 1988). The increment function has standing volume, stand age and site index as explanatory variables. Functions for pine, spruce and birch (used for all broad-leaves) were used and the increment was weighted with the tree species share of total volume.

4. Results

4.1 Precipitation and hydrology

Monthly precipitation amount at the nearby meteorological station Åsbø in Gjerstad and water discharge at the three monitoring stations are shown in **Figure 4** and **Figure 5**.

According to the runoff map of Norway average discharge in the area is 900 mm yr⁻¹ (29 L s⁻¹ km⁻²). Based on the measured runoff and the subcatchment areas of stations LIM-1, LIM-4 and REF specific discharge for the four years 1994-1997 were calculated (**Table 3**).

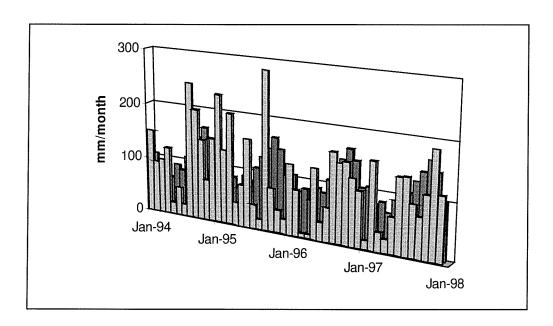
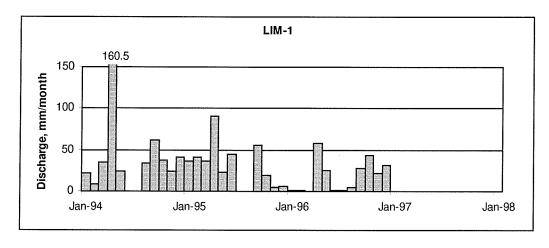
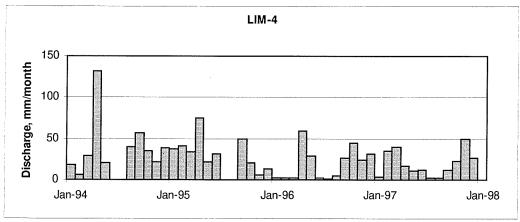


Figure 4. Monthly precipitation amount at the meteorological station (3520 Åsbø) in the monitoring period (in the front) and the 30-year monthly normal precipitation (in the back). Data from the Norwegian Meteorological Institute.

1994 and the hydrological year from summer 1994 to summer 1995 was very wet compared to the following years. Only the year 1995 was close to normal. 1996, 1997 and especially the hydrological year 1995-1996 were dry. Lower discharge in 1996 and 1997 compared with the initial period after liming is also evident when accumulated discharge is calculated (**Figure 6**).





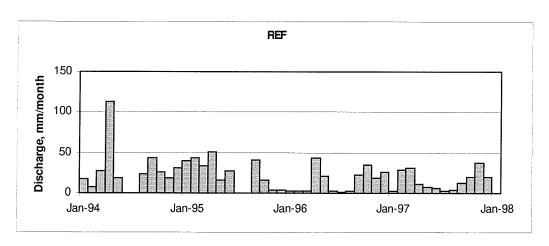


Figure 5. Monthly discharge at LIM-1, LIM-4 and REF. Discharge in 1997 was not measured at LIM-1.

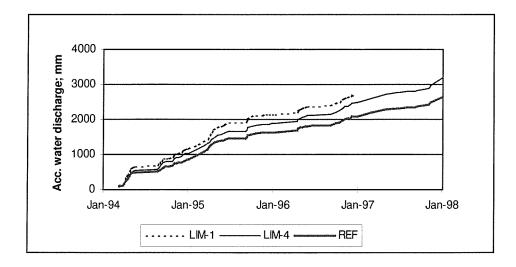


Figure 6. Accumulated water discharge (in mm) at monitoring stations LIM-1, LIM-4 and REF for the period 1994-1997.

Table 3. Calculated specific discharge (mm yr⁻¹) for catchments LIM-1, LIM-4 and REF. Results are given for calendar years and for hydrologic years (1 July to 30 June).

Calendar year		1994	1995	1996	1997
	LIM-1	1057	839	517	612
	LIM-4	1034	855	602	609
	REF	1082	918	599	
Hydrologic year		1994-1995	1995-1996	1996-1997	
	LIM-1	1104	410	653	
	LIM-4	1126	486	640	
	REF	1170	464		

4.2 Runoff water quality

4.2.1 Preliming characteristics

Significant pre-liming variability between subcatchments is reflected in pH, Ca, Mg, ANC, Alfractions and TOC but not in SO₄ and NO₃ (**Table 4**). This reflects the relatively uniform response of S and N deposition on SO₄ and NO₃ in runoff within the catchment and the more non-uniform geochemical characteristics. The areas draining to LIM-2 and LIM-3 are more acid-sensitive that the LIM-1 catchment. These two areas also had the most toxic water before liming, as seen in lower pH and ANC and higher Ali.

Prior to liming the stream water in both catchments was chronically acidified (**Table 4** and **Table 5**). pH was generally between 4.5 and 5.5 (**Figure 7**), and mean concentrations of reactive aluminium (Alr) were 248 µg L⁻¹ and 191 µg L⁻¹ in the two catchments (**Table 5**). About 70 % of the Al was organically bound. The relatively high organic Al-fraction can be related to the generally high contents of total organic matter (TOC) in the streams (6-8 mg TOC L⁻¹).

4.2.2 Post-liming characteristics

After liming significant increases in pH, Ca, Mg and ANC were found for all LIM-stations relative to the REF-station. **Table 5** summarises this effect for LIM-4. The immediate response in pH, which increased to 5.5-6.0, is clearly seen in **Figure 7**. At the main outlet of the limed catchment (at LIM-4) the mean concentration of Ca and Mg increased by 0.35 and 0.44 mg L⁻¹, respectively. Acid

Table 4. Pre-liming mean values and st. dev. (in brackets) of 4 independent sampling stations (n=17 for LIM-2; n=19 for LIM-1, LIM-3 and REF) and of LIM-4 (n=19). Letters a-d denote significant differences (p<0.05) between columns based on multiple range tests.

LJM-4	44(06)	(7) (7)	16 (6)ab	4.84 (0.19)bc	1.13(0.29)b	0.36 (0.08)b	11 (11)bc	248 (69)b	176 (59)	72 (19)b	7.0 (2.6) ab
REF	4.3 (0.7)	50 (57)	17 (6)b	4.79 (0.20)b	1.26 (0.33)b	0.36 (0.08)b	23 (17)ď	191 (56)a	157 (49)	34 (14)a	8.5 (2.9)b
LIM-3	4.4 (0.4)	60 (57)	17 (6)ab	4.82 (0.20)bc	1.09 (0.23)b	0.32 (0.05)ab	5 (10)b	268 (74)b	164 (52)	103 (35)c	6.1 (2.4)a
LIM-2	4.2 (0.5)	67 (73)	24 (5)c	4.63 (0.10)a	0.85(0.19)a	0.30 (0.06)a	-6 (9)a	273 (63)b	184 (51)	89 (27)c	6.8 (2.4)ab
LIM-1	4.6 (0.8)	74 (88)	13 (6)a	4.94 (0.23)c	1.22 (0.37)b	0.41 (0.11)c	17 (14)cd	233 (72)ab	175 (60)	58 (20)b	8.1 (3.4)b
	mg L	ng L-I	${\sf hed}\ { m L}^{\text{-}{\sf I}}$	-	${ m mg}{ m L}^{-1}$	${ m mg}{ m L}^{-1}$	μ ed Γ^{-1}	µg L'.	µg L⁻'	$\mu \mathrm{g} \mathrm{L}^{'}$	mg L ⁻¹
	SO_4	NO3-N	*+ H	$\widetilde{\mathrm{pH}^*}$		Mg*	ANC*	Alr*	Alo	Al1*	T0C*

neutralising capacity (ANC) increased from 11 to 49 μeq L⁻¹, whereas a decrease was seen in the unlimed catchment (**Table 5**. Liming did not result in significant changes in Na, K, NH₄, SO₄, Cl, Alo, tot N, organic N, TOC or organic C/N.

After the initial increase during the first 10 months after liming, no significant trends were detected for base cations, pH, ANC or aluminium species at LIM-4. There were thus no signs of re-acidification.

A significant decrease in Alr (p<0.001) due to liming was found for LIM-1 and LIM-4 (**Table 6** and **Figure 9**). The decrease in Alr-concentration was rather modest, however, from 42 and 47 μ g L⁻¹ preliming to -10 and -9 μ g L⁻¹ post-liming for LIM-1 and LIM-4, respectively, relative to REF. This means that the Alr-concentration in the limed catchment was only slightly lower than in the reference catchment after liming. A corresponding decrease from 24 and 38 to -22 and -21 μ g L⁻¹ was found for the difference in the inorganic monomeric fraction (Ali-diff). Ali is supposed to include the toxic Alspecies, and the mean concentration of this fraction was 17 μ eq L⁻¹ after liming (**Table 5**). No significant liming-related change in Alo or TOC was found.

The Random Intervention Analysis also shows that pH, Ca, Mg, and ANC concentrations increased and Alr and Ali decreased significantly in LIM-4 relative to REF in September 1994, the point of time of liming (**Table 6**). RIA indicates no significant changes in TOC or NO₃.

A significant (p<0.05) increasing trend was found for TOC at LIM-1, but not at LIM-4. For sulphate and the organic C/N-ratio this was reversed; a significant increasing trend at LIM-4, but not at LIM-1.

Table 5. Mean stream water values and std. deviation (in brackets) for different variables in the limed (LIM-4) and reference (REF) catchments before and after treatment in September 1994. n=17-19 preliming and n=55-57 post-liming for all variables. * denotes significant change (p<0.05) of mean or median at LIM-4 due to liming and at REF after the time of liming. Significance of change at LIM-4 is based on differences of simultaneously taken samples according to ANOVA or the non-parametric Kruskall-Wallis test, see also **Table 6**.

		LIM-4		R	EF
		Before	After	Before	After
H^{\dagger}	μeq L ⁻¹	15.9 (6.2)	1.6 (0.88)*	17.3 (6.3)	12.7 (4.5)*
pН	units	4.84 (0.19)	5.86 (0.23)*	4.79 (0.20)	4.93 (0.18)*
Ca	mg L ⁻¹	1.13 (0.29)	1.51 (0.25)*	1.26 (0.33)	1.29 (0.28)
Mg	mg L ⁻¹	0.36 (0.08)	0.81 (0.17)*	0.36 (0.08)	0.38 (0.08)
Na	mg L ⁻¹	1.81 (0.35)	1.73 (0.20)	1.81 (0.39)	1.74 (0.25)
K	mg L ⁻¹	0.21 (0.11)	0.21 (0.09)	0.17 (0.08)	0.20(0.12)
NH ₄ -N	μg L ⁻¹	23 (31)	40 (56)	19 (14)	52 (58)
SO ₄	mg L ⁻¹	4.4 (0.6)	4.3 (0.9)	4.3 (0.7)	4.0 (0.9)
C1	mg L ⁻¹	2.2 (0.8)	2.6 (0.7)	2.2 (0.8)	2.5 (0.7)*
NO ₃ -N	μg L ⁻¹	70 (71)	98 (98)**	50 (57)	59 (58)
ANC	μeq L ⁻¹	11 (11)	54 (19)*	23 (17)	17 (14)
Alr	μg L ⁻¹	248 (69)	167 (40)*	191 (56)	176 (38)
Alo	μg L ⁻¹	175 (59)	152 (40)	157 (49)	140 (30)
Ali	μg L ⁻¹	72 (19)	15 (12)*	34 (14)	37 (18)
Tot N	μg L ⁻¹	314 (163)	355 (115)	328 (115)	358 (124)
Org N	μg L ⁻¹	223 (116)	217 (80)	261 (106)	247 (100)
TOC	mg L ⁻¹	7.0 (2.6)	7.4 (2.1)	8.5 (2.9)	8.4 (2.3)
Org C/N		34 (6)	35 (4)	34 (4)	35 (6)

Table 6. Results of random intervention analysis comparing difference in concentrations in paired samples collected from LIM-4 (limed) and REF (reference) for the periods pre- and post-liming.

			H⁺	pН	Ca	Mg	Alr	Ali	TOC	ANC	NO ₃ -N
			μeq L ⁻¹		mg L ⁻¹	mg L ⁻¹	μg L ⁻¹	$\mu g L^{-1}$	mg L ⁻¹	$\mu eq L_{\stackrel{-}{l}}$	μg L ⁻¹
pre-liming	average	LIM4	15.9	4.84	1.14	0.36	250	73	7.2	11	70
n=18	average	REF	17.3	4.80	1.28	0.37	194	34	8.7	24	50
	avg. diff-p	re = LIM4-REF	-1.4	0.04	-0.14	0.00	56	38	-1.5	-13	20
post-liming	average	LIM4	1.6	5.86	1.51	0.81	167	15	7.4	54	98
n=57	average	REF	12.7	4.93	1.29	0.38	176	36	8.4	17	59
	avg. diff-pe	ost = LIM4-REF	-11.1	0.93	0.22	0.42	-9	-21	-1.0	36	40
change in difference between LIM4 and REF at point of time of liming, Sept. 94											
-	= diff-post		-9.7	0.89	0.36	0.43	-65	-60	0.5	49	19
	RIA level	of significance, p<	0.001	0.001	0.001	0.001	0.001	0.001	n.s.	0.001	n.s.

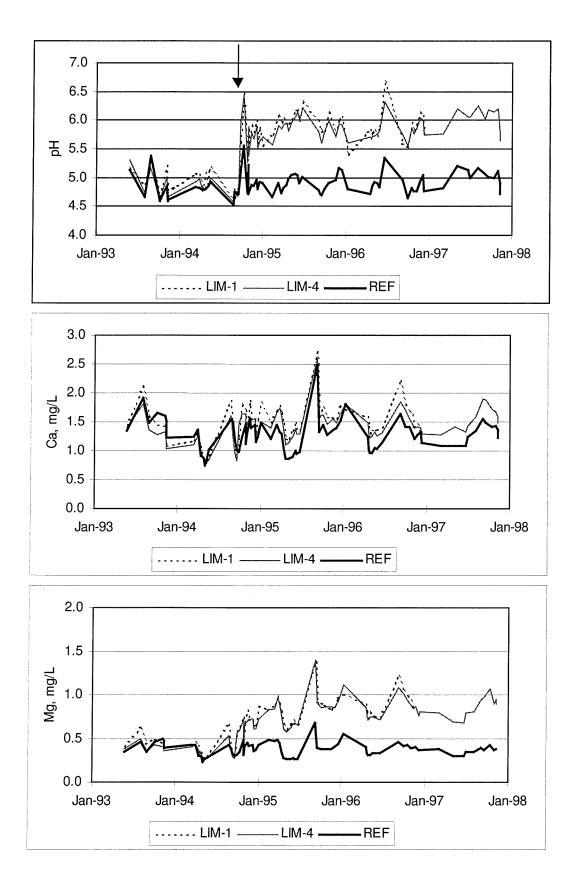


Figure 7. pH, Ca and Mg in streams of limed (LIM-1 and LIM-4) and reference (REF) catchment in the Gjerstad forest. Dolomite was spread in September 1994 (arrow).

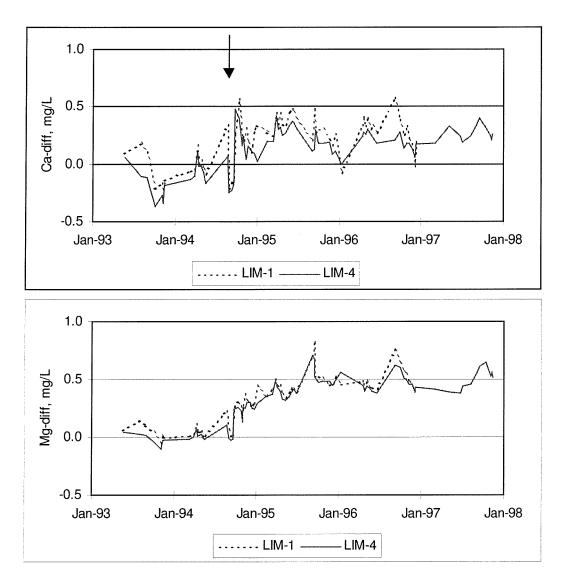
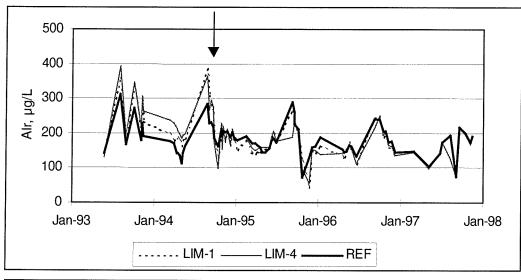


Figure 8. Differences in Ca and Mg concentrations between limed and reference streams in the Gjerstad forest. Dolomite was spread in September 1994 (arrow).



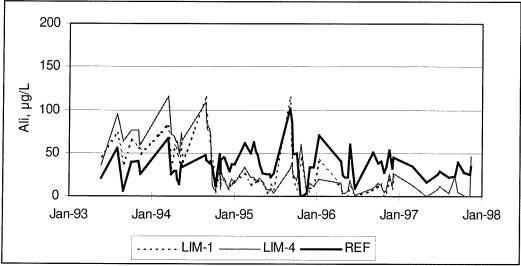
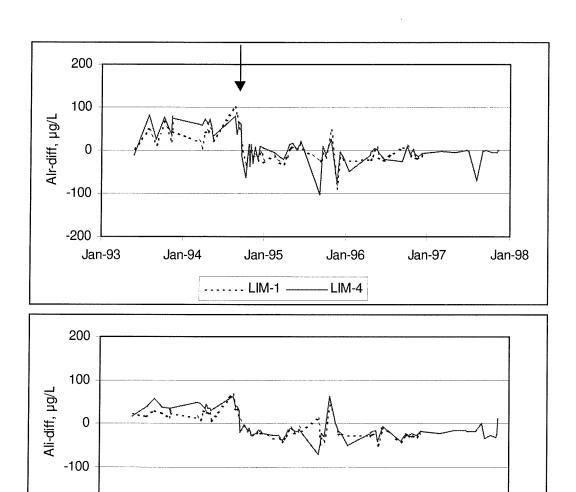


Figure 9. Reactive (Alr) and inorganic monomeric (Ali) aluminium in streams of limed (LIM-1 and LIM-4) and reference (REF) catchment in the Gjerstad forest. Dolomite was spread in September 1994 (arrow).



-200

Jan-93

Jan-94

Figure 10. Differences in concentrations of reactive (Alr) and labile (Ali) aluminium between limed and reference streams in the Gjerstad forest. Dolomite was spread in September 1994 (arrow).

Jan-96

LIM-4

Jan-97

Jan-98

Jan-95

----- LIM-1

Sea salts were not supposed to have significant impacts on the water chemistry, in terms of episodic acidification and Mg supply to the forest soil, due to the distance from the coast. However, in January 1995 (four months after liming) a strong increase in stream Cl concentrations indicated relatively large inputs of salts, and this resulted in negative values for non-marine Na and the lowest estimated ANC in the unlimed catchment in the project period **Figure 11**. The limed catchment also showed a drop in ANC, but not to negative values.

The seasonal variability was relatively large for concentrations of NO₃ (**Figure 12**). Some of the problems this might have on the statistical tests were eliminated by use of the differences between data-pairs of limed and reference samples. But also the differences were characterised by seasonal variability (**Figure 12**). This was found for both the preliming and post-liming period. The streams do not appear to differ until the second winter following liming. The result of the ANOVA-test of the preliming and post-liming NO₃-diff for LIM-1 and LIM-4 was a small but significant increase in the medians at the p<0.05 level in LIM-4, but not in LIM-1. Randon intervention analysis showed no significant difference (**Table 6**).

The organic N fraction, the C/N-relationship of the dissolved organic matter and total phosphorus concentration did not change significantly after liming (**Table 5** and **Figure 13**).

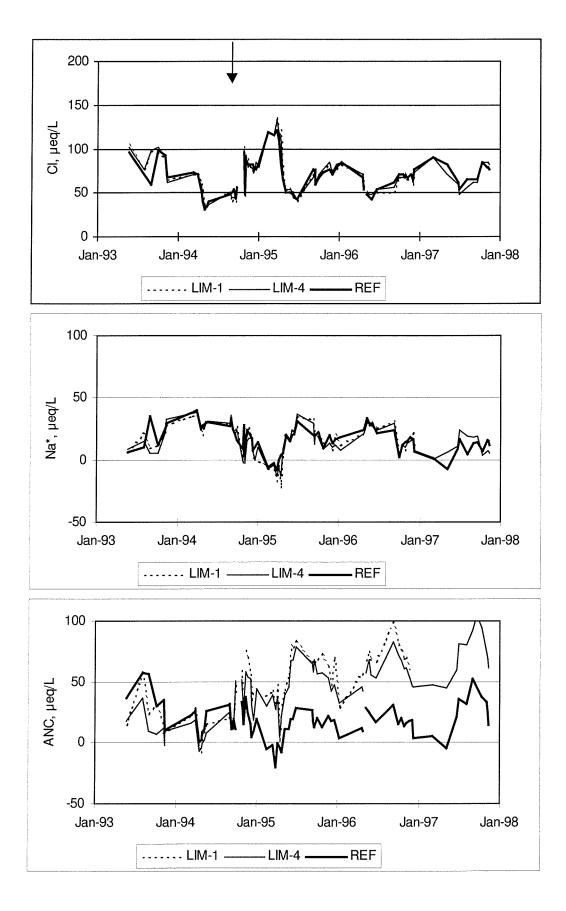


Figure 11. Chloride, non-marine Na (Na*) and ANC in streams of limed (LIM-1 and LIM-4) and reference (REF) catchment in the Gjerstad forest. Dolomite was spread in September 1994 (arrow).

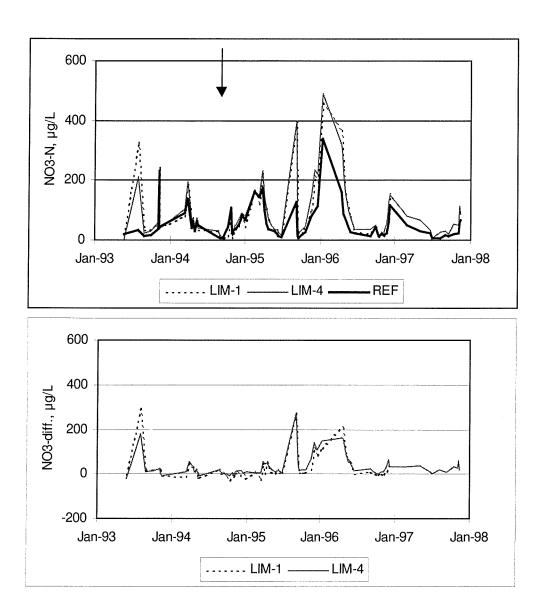
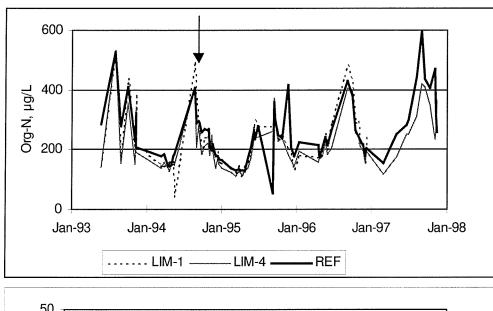


Figure 12. NO₃-N in streams of limed (LIM-1 and LIM-4) and reference (REF) catchments in the Gjerstad forest. The NO₃-N-diff. is shown in the lower panel. Dolomite was spread in September 1994 (arrow).



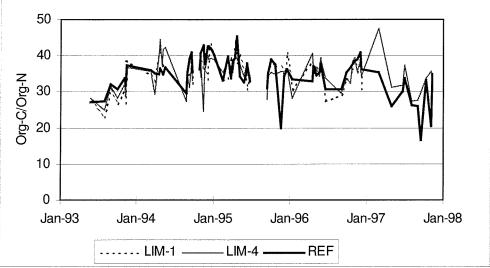


Figure 13. Dissolved organic N and the Org C/Org N ratio in streams of limed (LIM-1 and LIM-4) and reference (REF) catchment in the Gjerstad forest. Dolomite was spread in September 1994 (arrow).

4.2.3 Trace metals

No pre-liming data of trace metals are available and thus changes from pre-liming to post-liming period cannot be analysed. However, concentrations in the limed stream relative to the reference stream were analysed for differences and post-liming trends.

Relative to REF none of 10 trace metals in monthly point samples were significantly higher after liming (**Table 7**). In fact, Cd, Co, Fe, Mn, Ni, Pb and Zn were significantly lower in the stream of the limed catchment, which may be due to inherent differences between the streams or may be an effect of the liming. Only Mn, Co and Zn showed significantly decreasing trends in runoff from the limed catchment during the 3-year monitoring period (**Figure 14**). Fe, Pb, Cd, Cu, As and Ni remained relatively unaffected by the dolomite application. Cr was mostly below detection limits in both limed and unlimed catchments.

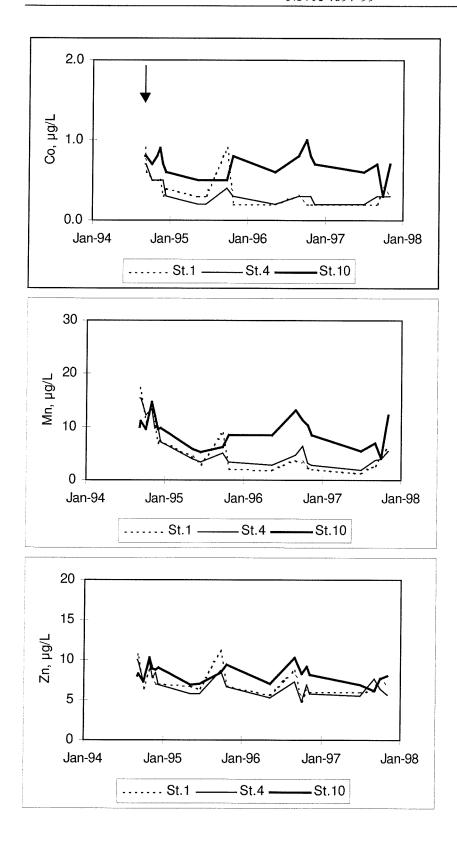


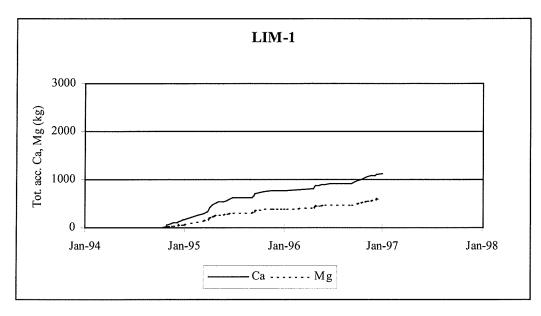
Figure 14. Cobalt (Co), manganese (Mn) and zinc (Zn) in streams of limed (LIM-1 and LIM-4) and reference (REF) catchment in the Gjerstad forest. Dolomite was spread in September 1994.

Table 7. Mean and st. dev. or median (in $\mu g \, L^{-1}$) of nine trace metals from point samples of streams in limed (LIM4) and unlimed (REF) catchments. All samples (n=18) are from the post-liming period. * denotes significantly (p<0.05) lower mean or median values for LIM-4 than for REF. Other metals are not significantly different. Concnetrations in the samples reported under the detection limit were set at ½ the detection limit.

	LIM	[-4	REF	· ·	
	Mean (st.dev.)	Median		Mean (st.dev.)	Median
As	0.27 (0.15)			0.36 (0.20)	
Cd*			0.03		0.05
Co*	0.32 (0.11)			0.68 (0.17)	
Cu	0.52 (0.17)			0.58 (0.21)	
Fe*			122		201
Mn*	5.7 (3.4)			9.0 (3.0)	
Ni*			0.8	, ,	1.05
Pb	0.64 (0.19)			0.72 (0.16)	
Zn*	6.8 (1.3)			8.2 (1.2)	

4.2.4 Ca and Mg transport

The total transport of Ca and Mg at LIM-4 during the first 3 years after liming was 2090 and 1150 kg, respectively (**Figure 15**; transport at LIM-1 is also shown). Based on the pre-liming and post-liming differences between samples from LIM-4 and REF a net transport of 518 kg Ca and 576 kg Mg from the applied dolomite was calculated for the first 3 years after liming (**Figure 16**). This corresponds to 0.9 % and 2.0 % of the added amounts of these elements. An annual transport of less than 1 % of the dose for both Ca and Mg was thus found in this experiment.



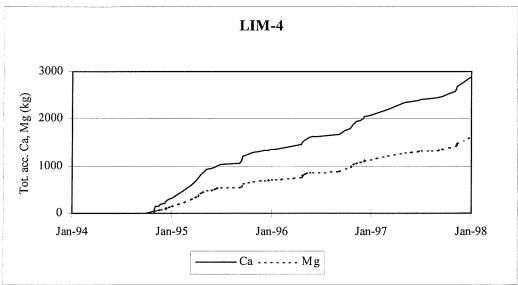
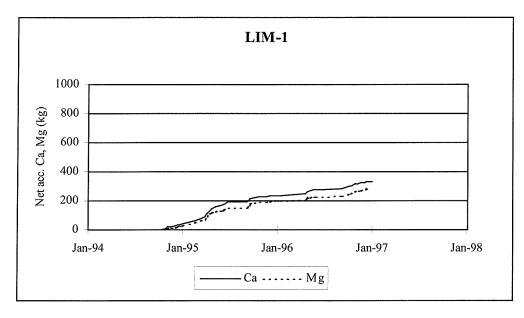


Figure 15. Accumulated total amount of Ca and Mg transported out at stations LIM-1 and LIM-4 of the limed catchment.



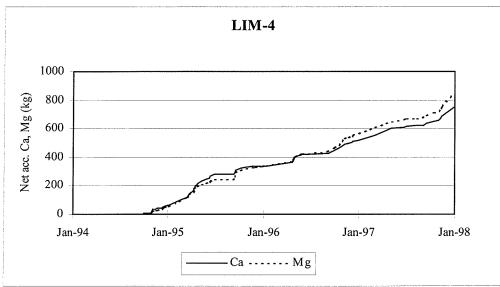


Figure 16. Accumulated net (due to liming) amount of Ca and Mg transported out at stations LIM-1 and LIM-4 of the limed catchment, relative to that expected based on transport from the reference catchment (REF).

4.3 Soil chemistry

The soil data at the lysimeter stations represent the composition of the organic (5 cm) and mineral (15 cm) soil layer before liming. Data for six locations, three located in the limed catchment and three located in the control catchment are given in **Table 8** and **Table 9**.

The results show that the soils vary widely in chemical composition. This reflects the wide range in site characteristics. The 3 sites in each catchment were intentionally chosen to cover a range in vegetation, soil and moisture conditions.

4.3.1 Soils of the control catch ment

Stations SP 28 and SP 30.

This location is covered with ferns and deciduous trees. The ground water table is high, and there were never problems in filling the lysimeter bottles. The pH values in the water samples extracted from the organic and mineral soil were relatively high (4.82 and 4.9, respectively).

Stations SP 37 and SP 40

Station SP 37 is located in a hollow in the ridge hill, while SP 40 is located in a rather dry and poor slope just above SP 37. The dominating forest is coniferous. The soil solution samples normally had a vigorous yellow colour and the volume at SP 40 was usually low, especially at 15 cm.

Stations SP 46 and SP 50

These stations are located in a shady area beneath large spruce trees. The ground vegetation was poorly developed. The site is moderately dry.

4.3.2 Soils of the limed catchment

Stations FU 06 and FU 07

Stations FU 06 and FU 07 were located in poor soil in a pine forest. The soil solution samples from this location normally were low in volume and of yellow colour. The soil quality can be compared to soils at stations SP 30 and SP 40.

Stations FU 44 and FU 45

These stations are located in the bottom of a small valley with a mixed forest of coniferous and deciduous trees, among them a number of birches. The ground water level is high and the lysimeter bottles were filled up with water within a few hours.

Stations FU 48 and FU 49

Station FU 48 was placed in a boggy soil close to a stream, while FU 49 was located higher up on the ground in between the roots of a big alder. The lysimeter FU 49 seldom gave water samples at all.

Taken together the 6 soils at the 3 sites in the reference catchment are similar chemically to the 6 soils of the 3 sites in the limed catchment (**Figure 17**). The heterogeneity of the sites is reflected in the large standard deviations about the means. The soils are typical for coniferous forests in Norway, although they have rather low C/N ratios. This may reflect the long-term deposition and storage of N in the soil.

■ reference

☐ reference

Ilimed

■ limed

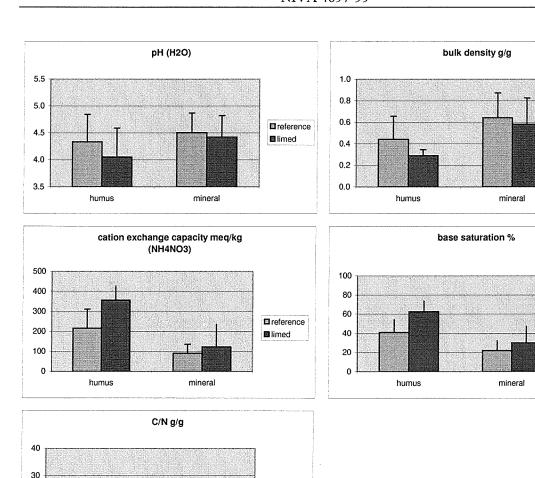


Figure 17. Soil chemistry at the 6 sites in the reference and limed catchments at Gjerstad, sampled in summer 1994 prior to liming. Mean and standard deviation of the samples (n=6 for humus; n=12-15 for mineral soil) are shown.

□ reference

■ limed

mineral

20

10

humus

Table 8. a. Pre-liming composition of organic soil in three locations in the reference catchment (SP) and three locations in the limed catchment (FU). B.D. = bulk density; L.O.I. = loss on ignition; Kjel-N = Kjeldahl-N.

	S	mmol kg-1	63.62	81.42	78.26	50.28	30.53	16.53	58.34	62.16	121.6	83.99	73.84	68.34	9	53.44	16.53	81.42	26.08	9	78.05	58.34	121.60	23.17
		mmol kg ⁻¹ 1	6.29	6.38	8.25	4.93	6.14	4.86	8.62	7.73	13.81	9.12	6.78	7.4	9	6.14	4.86	8.25	1.24	9	8.91	6.78	13.81	2.54
		mmol kg-1	104.3	15.5	16.9	23.6	150.8	40.6	22.1	21.4	23.5	16.6	44.8	49.7	9	58.62	15.50	150.80	56.10	9	29.68	16.60	49.70	13.89
s, ICP	×	mmol kg ⁻¹	19.2	16.4	26.9	23.8	22.4	10.8	26.7	20.3	25.8	20.7	19.3	18.2	9	19.92	10.80	26.90	5.76	9	21.83	18.20	26.70	3.54
total analyses, ICF	Ca	mmol kg ⁻¹	28.6	20.8	33.3	50.5	62.6	42.1	83.3	79.8	134.2	78.4	50.2	111.8	9	39.65	20.80	62.60	15.29	9	89.62	50.20	134.20	29.31
	Al	mmol kg ⁻¹	1285	716	260.2	114.2	427.7	152.9	50.66	61.59	403.2	819.1	113.2	91.24	9	536.17	114.20	1285.00	483.31	9	256.50	50.66	819.10	305.37
	calc. C/N		16.15	16.75	21.65	26.26	17.16	14.11	29.57	27.04	15.89	19.39	17.34	16.57	9	18.68	14.11	26.26	4.46	9	20.97	15.89	29.57	5.86
	Kjel-N	mmol kg ⁻¹	1167	1717	1484	1009	616	386	1081	1141	2061	1559	1410	1325	9	1063.17	386.00	1717.00	505.34	9	1429.50	1081.00	2061.00	355.40
		%	52.78	80.53	89.95	74.18	29.59	15.25	89.49	86.38	91.72	84.64	68.47	61.48	9	57.05	15.25	89.95	29.82	9	80.36	61.48	91.72	12.37
	Dry wt.	%	90.4	88.3	90.1	92.7	94.7	9.76	90.1	8.06	88.9	88.1	92.7	93.5	9	92.30	88.30	09.76	3.42	9	89.06	88.10	93.50	2.11
	B.D.	g g_	0.526	0.28	0.236	0.263	0.578	0.767	0.22	0.231	0.293	0.285	0.354	0.349	9	0.44	0.24	0.77	0.22	9	0.29	0.22	0.35	90.0
	$^{\mathrm{hd}}$	H ₂ O	4.86	4.82	3.87	3.6	4.44	4.42	3.55	3.56	4.51	4.88	3.79	4.02	9	4.34	3.60	4.86	0.51	9	4.05	3.55	4.88	0.54
HUMUS GJERSTAD, 1994															ref	ref	ref	ref	ref	limed	limed	limed	limed	limed
MUS GJER		site	34 SP28	36 SP30	44 SP37	48 SP40	55 SP46	60 SP50	67 FU06	68 FU07	113 FU44	114 FU45	118 FU48	119 FU49	SP	SP	SP	SP	SP	FU	FU	FU	FU	FU
HU		OBS																		count			max	

Table 8. b. Pre-liming composition of organic soil (extracts in NH_4NO_3 (1M)) in three locations in the reference catchment (SP) and three locations in the limed catchment (FU). CEC = cation exchange capacity; BS = base saturation.

	S	mmol kg-1	4.1	5.03	6.29	3.72	4.14	1.24	3.81	4.07	6.64	5.23	5.14	3.55	9	4.09	1.24	6.29	1.67	9	4.74	3.55	6.64	1.16
	Ъ	mmol kg-1	0.1	0.1	0.86	2.86	0.19	0.15	1.88	2.11	0.48	0.1	1.06	1.35	9	0.71	0.10	2.86	1.09	9	1.16	0.10	2.11	0.78
	Na	mmol kg-1	3.48	5.47	7.45	4	1.56	0.98	8.54	7.26	12.97	8.76	5.4	5.2	9	3.82	0.98	7.45	2.42	9	8.02	5.20	12.97	2.85
sions	Mg	mmol kg-1	4.28	10.48	12.23	13.96	98.9	4.07	17.53	17.43	18.58	8.47	25.3	26.76	9	8.65	4.07	13.96	4.19	9	19.01	8.47	26.76	6.57
exchangeable ions	×	mmol kg-1	8.09	14.27	24.31	17.53	7.41	3.85	21.95	17.61	22.58	14.07	15.29	12.28	9	12.58	3.85	24.31	7.59	9	17.30	12.28	22.58	4.23
J		mmol kg ⁻¹	13.79	35.64	29.48	46.21	35.08	20.98	78.6	72.99	127.5	67.84	40.62	90.25	9	30.20	13.79	46.21	11.53	9	79.63	40.62	127.50	28.67
	Al	mmol kg ⁻¹	34.58	62.94	62.9	14.12	21.12	9.33	7.16	10.7	36.76	47.54	15.4	4.16	9	34.67	9.33	65.90	24.59	9	20.29	4.16	47.54	17.68
	BS	%	31.6	24.7	32.3	48.7	51.4	57.1	54.7	52	73.4	58	57.6	79.4	9	40.97	24.70	57.10	13.09	9	62.52	52.00	79.40	11.13
	CEC	μeq kg ⁻¹	150.9	219.3	356.4	291.5	180.5	96.2	407.6	395.5	446.4	302.5	265	316.9	9	215.80	96.20	356.40	95.16	9	355.65	265.00	446.40	70.80
															ref	ref	ref	ref	ref	limed	limed	limed	limed	limed
		OBS site	34 SP28	36 SP30	44 SP37	48 SP40	55 SP46	60 SP50	67 FU06	68 FU07	13 FU44	114 FU45	18 FU48	119 FU49	$_{ m SP}$	SP	SP	$_{ m SP}$	SP	FU	FU	FU	FU	FU
		0										-	-		count	average	min	max	S.D.	count	average	min	max	S.D.

Table 9. a. Preliming composition of mineral soils at three locations in the reference catchment (SP) and three locations in the limed catchment (FU). B.D. = bulk density; L.O.I. = loss on ignition; Kjel-N = Kjeldahl-N.

MIINER 1994	MIINERAL horizon GJERSTAD, 1994	D,							total analyses, ICP	ss, ICP		mmol kg	l kg-l		
			Ηd	B.D.	Dry wt.	L.O.I.	Kjel-N	calc. C/N	AI	S	×	Мо	ž	۵	V.
OBS	site		H_2O	g g.	%	% I	mmol kg ⁻¹	g g-1			 	a !	3	1	מ
	1 FU06		3.80	89.0	98.2	15.76	231	24.366	49.24	19.5	7.6	6	2.93	4	11.38
	2 FU06	7	4.32	0.995	9.86	5.94	84	25.255	153.2	12.1	8.8	21.2	2.22	2.9	4 57
	3 FU07	-	3.78	0.518	97.2	22.95	404	20.288	98.24	28.5	11.2	13.8	2.55	6.7	22.71
	4 FU07	2		0.533	8.96	28.02	409	24.467	97.95	27	9.6	13.4	2.91	2.8	20.52
	5 FU07	n	4.44	0.843	26	11.1	159	24.933	391.2	18.8	12.1	54.2	2.42	7.6	10.07
	6 FU44		4.19	0.203	90.1	90.35	1884	17.127	293.7	136.4	14.8	28.2	4.22	26.9	103.5
		7	4.67	0.719	97.3	14.74	373	14.113	270.1	18.9	4.4	10.6	2.51	8.5	19.91
		_	4.81	0.229	88	84.34	1543	19.521	1120	45.7	8.6	12.3	2.55	47.3	97.54
	9 FU45	7	5.16	0.305	89.1	59.7	1182	18.038	1976	50.1	7.8	22.9	3.18	52	89.77
	10 FU45	m ·		0.334	88.8	59.06	1128	18.699	1970	49.6	9.8	22.5	3.5	54.3	88.18
	11 FU48		4.28	0.543	6.96	20.1	482	14.893	295.7	29	11.3	85.9	4.43	11.6	25.48
•	12 FU48	7	4.64	0.811	9.76	8.58	175	17.51	399.9	24.7	11.3	128	4.86	7.6	11.95
	13 FU49	-	4.36	0.627	9.76	15.44	413	13.352	293.1	31.6	8.3	104	5.19	10.4	22.28
	14 FU49	7	4.65	0.795	2.76	8.37	160	18.683	551.6	32.3	10.2	209.1	6.44	8.9	13.92
•	15 SP28	_	4.90	0.343	89.3	57.21	1432	14.268	1411	31.7	15.5	104.8	3.07	46.1	76.39
	16 SP28	7		0.35	68	58.33	1436	14.507	1445	31.2	18.2	108.5	3.25	7.44	81.15
•	17 SP28	က	4.91	0.519	92.5	31.51	703	16.008	1367	28.3	13.5	121.9	3.75	29.1	34.66
. • •	18 SP30	_	4.83	0.272	87.3	73.05	1650	15.812	1506	24.1	9.3	36.4	2.59	52.3	87.22
	19 SP30	7	4.88	0.623	94.8	20.63	478	15.414	1285	26.4	12.9	141.2	3.61	22.7	25.55
. 4 '			4.11	0.498	97.1	18.59	344	19.3	193.6	10.3	6.4	10.9	2.45	10	13.34
. 7		7	4.33	0.953	8.86	4.71	106	15.869	166.9	7.4	4.7	9.2	1.86	4	4.5
		က		0.981	8.86	4.74	103	16.436	168.5	8.7	9.5	9.3	1.84	3.7	4.38
. 4		_	3.89	0.581	98.1	13.28	201	23.596	189.2	12.4	21.1	48.5	2.25	6.7	10.04
. 4		7	3.97	0.932	6.86	5.4	66	19.481	149.6	8.6	15	41.7	2.18	3.8	3.89
. 4		_	4.53	0.62	92.8	21.58	394	19.561	547.7	50	15.4	135.6	5.76	41.9	26.95
. 1	26 SP46	7	4.75	0.698	96	13.15	249	18.861	649.5	47.9	10.1	146.3	5.99	37	16.13
. 4			4.47	0.611	98.1	10.98	252	15.561	146.4	29.7	7.5	31.7	4.66	16.4	12.19
. 1	28 SP50	7		0.718	98.2	10.63	246	15.433	150.1	31.9	∞	32.7	4.26	17	12.58
. 1	29 SP50	3	4.49	0.93	98.2	8.08	181	15.943	222.2	25.1	9.9	50.9	4.3	14.7	10.29

MIINER/	AL horizo	MINERAL horizon GJERSTAD,							total analyses, ICP	es, ICP		mmc	mmol kg ⁻¹		
1			Hd	B. D.	Dry wt.		Kjel-N	calc. C/N	Al	S C	×	Mg	Z	Д.	V.
OBS	site		H_2O	89. 1	%	%	mmol kg ⁻¹	99,				o	1	•	2
count	FU	limed	12	14	14		14	14	14	14	14	14	14	14	14
average	FU		4.43	0.58	95.1		616.2	19.4	568.6	37.4	9.6	52.5	3.6	18.0	38.7
min	FU		3.78	0.20	88.0		84.0	13.4	49.2	12.1	4.4	9.0	2.2	2.9	4.6
max	FU		5.16	1.00	9.86		1884.0	25.3	1976.0	136.4	14.8	209.1	6.4	54.3	103.5
S.D.			0.40	0.25	4.03	29.01	575.00	4.04	650.98	30.77	2.49	59.25	1.27	18.90	37.38
count	SP	ref	12	15	15		15	15	15	15	15	15	15	15	15
average	SP		4.51	0.64	95.4		524.9	17.1	639.8	24.9	11.6	9.89	3.5	23.3	28.0
min	SP		3.89	0.27	87.3		0.66	14.3	146.4	7.4	4.7	9.2	1.8	3.7	3.9
max	$_{ m SP}$		4.91	0.98	6.86		1650.0	23.6	1506.0	50.0	21.1	146.3	6.0	52.3	87.2
S.D.			0.37	0.23	3.98		533.79	2.55	578.97	13.46	4.80	51.44	1.33	17.19	29.15

Table 9. b. Preliming composition of mineral soils (extracts in NH₄NO₃ (1M)) at three locations in the reference catchment (SP) and three locations in the limed catchment (FU). CEC = cation exchange capacity; BS = base saturation.

v	mmol/kg	0.49	0.37	0.72	89.0	6.0	3.94	99.0	2.95	2.06	2.21	1.7	0.95	1.5	1.8	4.28	4.05	2.51	4.62	1.58	0.0	0.22	0.23	0.64	0.26	1.93	1.45	1.11	1.12	69.0
۵	mmol/kg	0.16	0.1	0.1	0.1	0.1	0.97	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.22	0.1	0.1	0.1	0.1	0.14	0.1
Ž	mmol/kg	0.57	0.22	0.73	0.77	0.22	3.56	0.42	1.9	1.66	1.72	0.58	0.35	0.77	0.34	1.48	1.48	0.87	1.71	0.72	0.65	0.23	0.26	0.38	0.28	0.51	0.41	0.41	0.37	0.27
le ions Mo	mmol/kg	2.3	0.29	3.9	3.89	0.37	21.08	0.84	4.47	2.69	2.71	4.01	0.85	3.1	0.63	5.06	5.1	1.65	4.24	0.97	2.28	0.29	0.29	2.42	0.64	2.58	0.89	2.05	1.87	0.81
exchangeable ions K	mmol/kg	4.24	0.68	3.55	2.53	1.15	16.73	2.31	9.37	5.5	4.91	5.44	2.09	3.17	1.32	9.18	8.89	2.56	7.74	2.26	4.1	0.68	0.63	3.19	0.84	3.83	1.67	2.94	2.63	1.01
Ca	mmol/kg	6.97	0.5	15.95	15.57	1.11	116	4.65	33.38	27.11	26.46	5.86	1.62	6.04	0.41	15.39	15.2	7.46	13.85	4.54	5.51	0.77	8.0	7.29	1.41	10.26	2.79	7.85	7.2	2.31
Ā	mmol/kg	5.15	15.54	18.82	18.9	19.31	43.72	15.94	55	24.88	25.42	19.35	13.2	15.79	14.98	36.32	35.62	22.39	48.18	17.62	34.53	14.94	15.75	16.57	14.63	19.02	18.51	10.89	10.71	12.47
BS	%	37.5	5.3	34.8	33.3	7.3	63	23.4	35.8	20	48.5	28.7	16.1	30.3	8.4	32.3	32.1	25.8	25.2	23.1	15.7	6.5	6.3	24.5	9.5	31.7	14.6	34.7	32.9	16
CEC	µeq/kg	62.2	46.7	126.4	126.8	59.1	467.3	58.6	242.6	133.5	133.9	8.68	45.8	73.3	44.5	159.4	158.6	83.8	180.9	9.09	129.5	46.7	48.6	93.9	55	94.6	64.8	2.99	64.2	46.9
	sample	1	2	1	2	3	-	2	—	2	3	-	2	-	2	-	2	3	-	2		2	3	_	2	-	2	_	2	c
	OBS site	1 FU06	2 FU06	3 FU07	4 FU07	5 FU07	6 FU44	7 FU44	8 FU45	9 FU45	10 FU45		12 FU48	13 FU49		15 SP28	16 SP28				_		22 SP37				2	27 SP50	28 SP50	29 SP50

	S	mmol/kg	þ	14		0.4		1.04	15.1	17	0.2	4.5	1.50
	Ь	mmol/kg		14	0.2		1.0	0.23	15	0.1	0.1	0.0	0.03
	Na	mmol/kg		14	1.0	0.2	3.6	0.93	27.5	0.7	0.2	17	0.50
e ions	Mg	mmol/kg	A COLUMN TO THE PARTY OF THE PA										1.60
exchangeabl	×	mmol/kg											2.88
ð	Ca	mmol/kg		14	18.7	0.4	116.0	30.07	15	8.9	0.8	15.4	5.02
	Al	mmol/kg		14	21.9	5.2	55.0	12.84	15	21.9	10.7	48.2	11.30
	BS	%		14	30.2	5.3	63.0	17.06	15	22.1	6.3	34.7	96.6
	CEC	μeq/kg		14	122.18	44.50	467.30	113.18	15	90.28	46.70	180.90	45.35
				limed	limed	limed	limed	limed	ref	ref	ref	ref	ref
	•	site		FU	FU	FU	FU	FU	SP	SP	SP	SP	SP
	8 }	OBS		count	average	min	max	S.D.	count	average	min	max	S.D.

4.4 Soil solution

The lysimeters were installed in September 1994. Samples collected through 31 October 1994 were disregard here due to possible effects of disturbance caused by the installation.

The chemical composition of soil solution varies widely between the sites, in part reflecting the heterogeneity of soil and site properties. For example, pH in the humus layer was about 4 at site SP40, FU 06 and FU 07, but 5.5-6.0 at site SP28, FU 44 and FU 45 (**Figure 18** and **Figure 19**). There were no trends over time in pH at any of the lysimeters in either the limed or reference catchments.

Mg concentrations in soil solution also varied from site-to-site (**Figure 20** and **Figure 21**). At 5 of the lysimeters in the limed catchment the Mg concentrations show a statistically significant (p< 0.05) increase during the 3 years following liming. There were no significant trends in Ca concentrations (**Figure 22** and **Figure 23**).

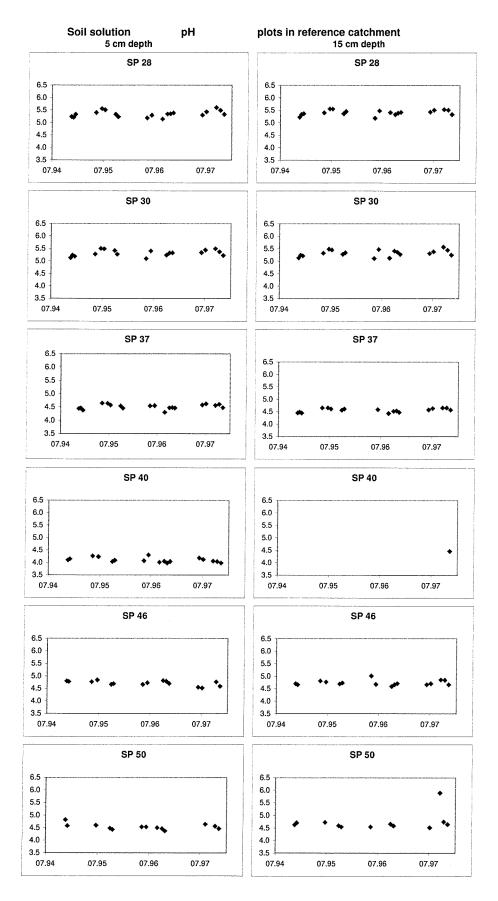


Figure 18. pH in soil solution in plots in the reference catchment.

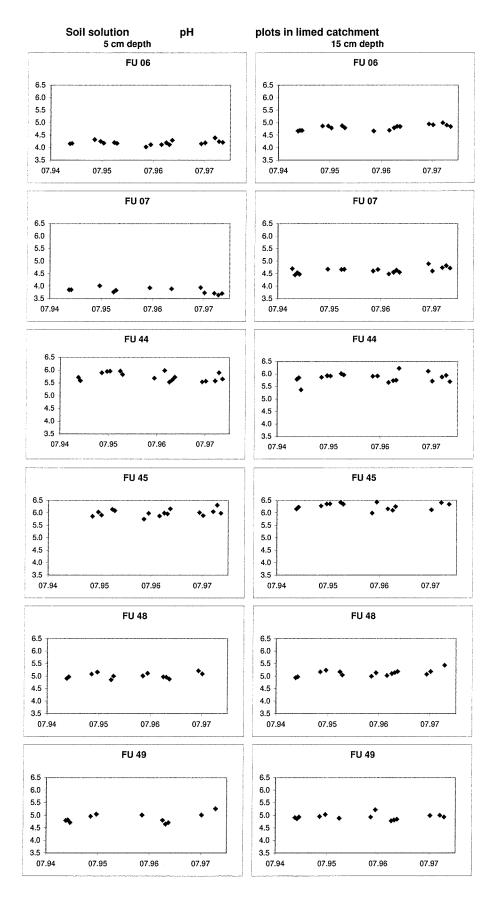


Figure 19. pH in soil solution in plots in the limed catchment.

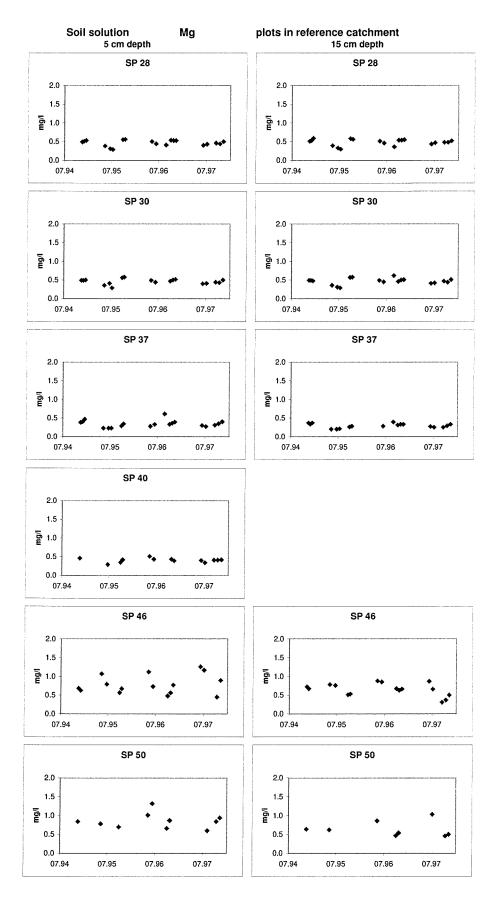


Figure 20. Mg concentrations in soil solution in plots in the reference catchment.

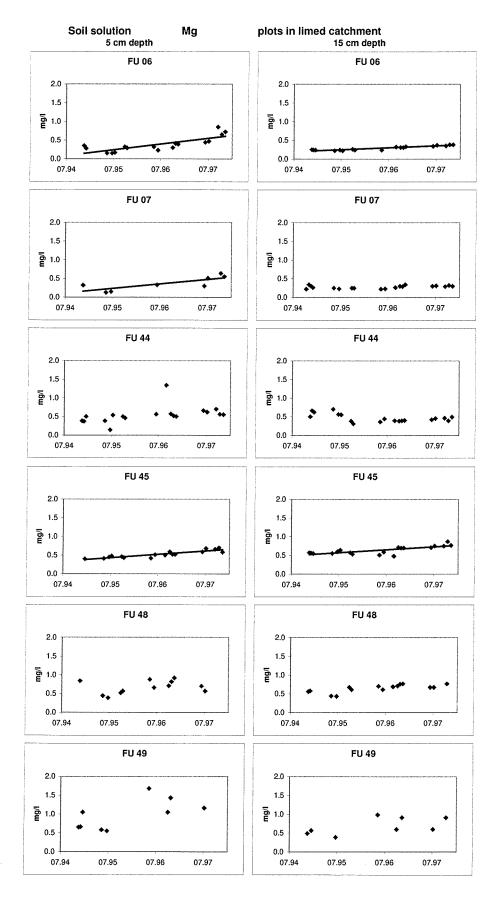


Figure 21. Mg concentrations in soil solution in plots in the limed catchment. Significant linear trends indicated by regression lines.

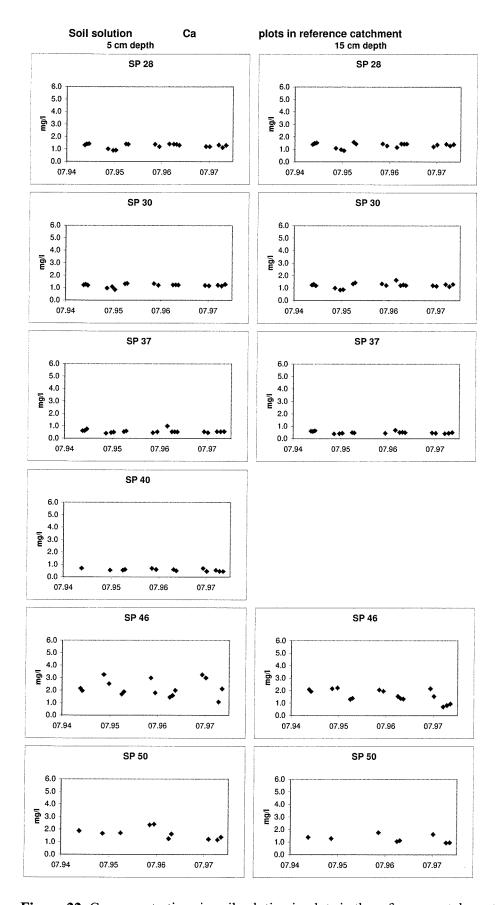


Figure 22. Ca concentrations in soil solution in plots in the reference catchment.

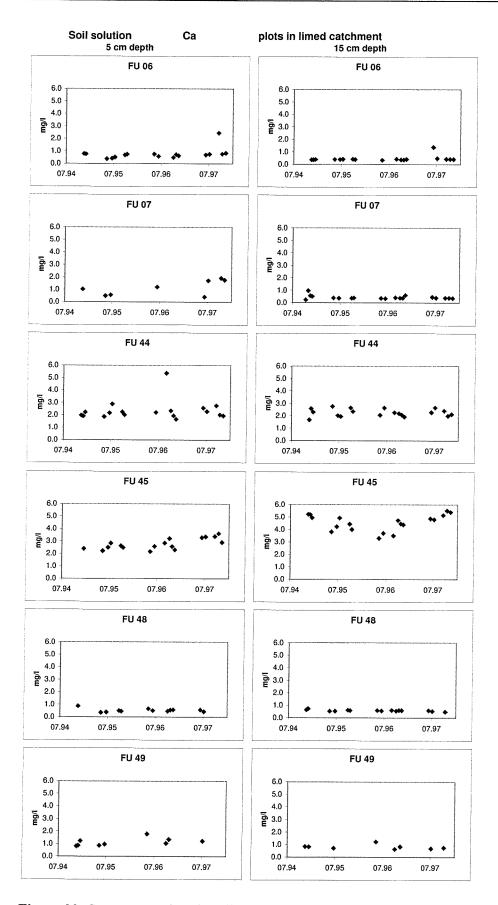
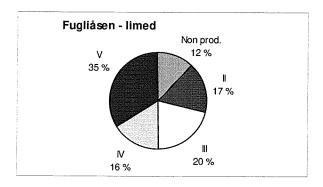


Figure 23. Ca concentrations in soil solution in plots in the limed catchment.

4.5 Forest investigations

4.5.1 Forest condition

In **Figure 24** the distribution of different cutting classes is shown in the two catchments. The sampling intensity is approximately 3% for both areas, and the stand composition is fairly alike. In the reference area there is less young and middle-aged forest. Here, about 70% of the forest area is classified as old production forest or old forest, while in the limed catchment this proportion is 50%.



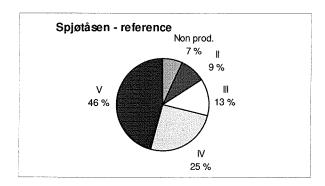


Figure 24. Area distribution of different cutting classes in the two catchments. Cutting classes according to following definition: I- Forest under regeneration, II - Regenerated areas and young forest, III - Young production forest, IV - old production forest, V - old forest.

Table 10 and **Table 11** show aggregated figures for forest condition in the two catchments. The share of Scots pine to Norway spruce is higher in the reference area (70/30) compared to the limed area (56/44). A small fraction of broad-leaved trees in the middle aged and old stands is typical for this region. The clear-cut areas in the catchments are mainly dominated by birch or trembling aspen.

Table 10. Some key figures for the forest stands in the two catchments.

	Fugleliåsen - limed area	Spjøtåsen - reference area
Area	84.4 ha	40.8 ha
Total standing volume	9920 m ³	6027 m ³
Mean diameter	140 mm	157 mm
Mean height	10.4 m	11.4 m
Average volume per tree	0.157 m ³	0.187 m ³
Mean age	67 yr	77 yr
Number of sample plots	122	56
Sample intensity	2.9%	2.7%
Site index (H ₄₀)	12.4	11.8

Table 11. Volume in different cutting classes and by tree species in the two catchments.

Fugleliåsen - limed catchment

				Volume (m	13)		
Cutting class	Spruce	Pine	Birch	Aspen	Oak	Other broadleaves	Sum
III	230	445	177	177	124	18	1171
IV	703	990	92	144	101	3	2033
V	2516	3025	192	502	306	51	6592
Sum	3449	4460	461	823	531	72	9796

Spjøtåsen - reference catchment

				Volume (m	3)		
Cutting classes	Spruce	Pine	Birch	Aspen	Oak	Other broadleaves	Sum
III	48	231	8	108	16	2	413
IV	467	1015	36	114	1	1	1631
V	1050	2318	117	214	57	47	3803
Sum	1565	3564	161	433	74	50	5847

4.5.2 Intensive monitoring plots

The increment in the 4-year period 1994-1998 is compared to the estimated increment based on increment function (**Figure 25**) The overall level of estimated increment is 70% for the reference catchment and 77% for the limed catchment. This difference is not significant. As seen from **Figure 25** the variation in residuals is rather large. One plot, no 6 in the reference catchment, has a very low increment compared to that expected from the function. This is probably caused by low vitality and declining trees. The reason for this has not been investigated further.

The average crown density and crown colour are shown in **Figure 26** and **Figure 27**. The crown density is also relatively low, with many of the stands with less than 85% crown density for most of the period. The level varies little from year-to-year; no effect of liming can be extracted. The same holds for the crown colour. The year-to-year variation is more pronounced for crown colour. An increase in colour value (more yellow trees) was found in 1995. Then a steady decrease in value (more green trees) has been noted in both catchments.

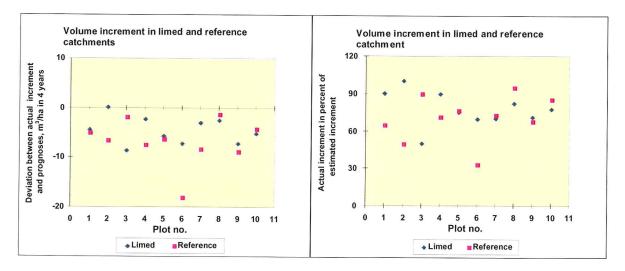


Figure 25. Differences between measured and estimated increment on the intensive monitoring plots. Residuals (m³ ha⁻¹ in 4 years) (left) and measured increment in percent of estimated (right).

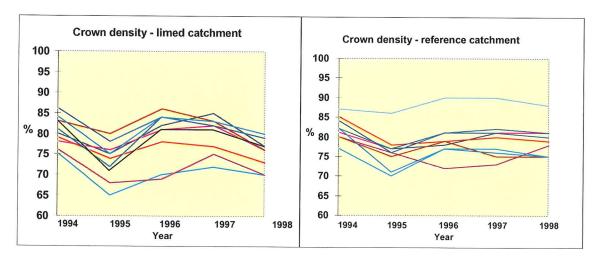


Figure 26. Average crown density on intensive monitoring plots in the period in the two areas.

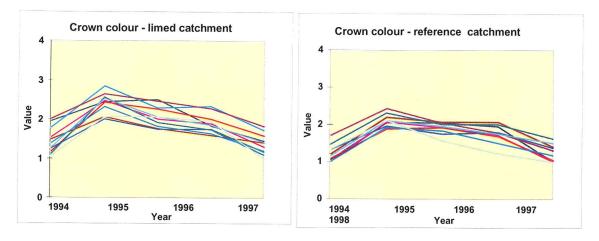


Figure 27. Average crown colour in the intensive monitoring plots during the period in the two catchments.

In **Figure 28** the percentage amount of green trees in each plot can be seen. For most of the plots an almost complete recovery in crown colour compared to 1994 has occurred. Still some stands have low proportions of green trees.

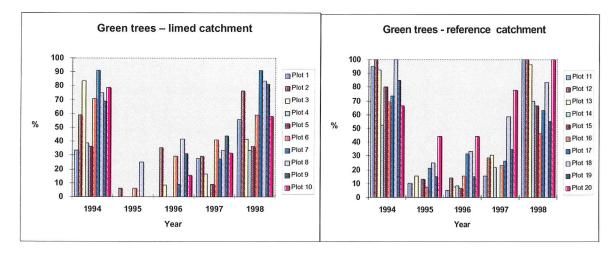


Figure 28. The amount of green trees on the intensive monitoring plots in the limed catchment and reference catchment in the different years.

5. Discussion

The choice of coarse dolomite and the dose of 3 t ha⁻¹ for liming the 80 ha forested ecosystem in Gjerstad was partly based on recommendations for forest soil liming in Sweden, as summarised by Nihlgård *et al.* (1996). Results from the whole-catchment liming at Tjønnstrond (Traaen *et al.* 1997) with the same dose of more fine-powdered calcite indicated that significant improvement of the water quality was likely. Higher doses may change the soil pH too much, whereas ion exchange of accumulated Al and H⁺ in the soil with the added Ca and Mg and production of organic acids from decomposition of organic matter may increase the acidity of the soil solution if lower doses are used.

Due to the anticipated variability of many important factors for acidity and liming effects within heterogeneous catchments, mixed forests and climatic regions, we chose a fixed dose and a paired catchment experiment in a forest ecosystem that might be regarded as typical for large acidified areas in Norway. An initial split in the limed catchment for stream water analyses and parallel sampling of soil solution at different stations were included to study in-catchment variability.

5.1 Soil chemistry

Sites for soil chemistry was selected mainly for vegetation purposes. Large gradients in nutrient status and water supply were two central criteria for this selection. As a result large variability in most of the soil chemical compounds was expected, and this is indeed what was found.

The chemistry of the soils at the two Gjerstad catchments is typical for soils in southern Norway in coniferous forests on nutrient-poor, siliceous moraine. Soils on well-drained sites are acidic and exhibit podsolic character, while those on adjacent wetter areas are typically organic-rich and humified. Both the cation exchange capacity (CEC) and the base saturation, two key parameters with

respect to acidification of soil and mitigation by liming, are well within the ranges expected for coniferous forest soils of southern Norway.

Whether liming has resulted in changes in soil chemistry such as increase base saturation remains to be seen. A resampling of soils is scheduled for 1999. Liming at Tjønnstrond, for example, resulted in significant increase in base saturation (Traaen *et al.* 1997).

5.2 Soil solution chemistry

In principal one should expect a close relation between chemistry of soil and chemistry of soil solution at each individual site. The wide range in soil chemistry should be reflected in a corresponding wide range of soil solution chemistry. This is in part the case for pH in the humus layer. In both catchments soils with low pH (water) yield soil solution with low pH (Figure 29). This is the case for the humus layer. The mineral soil horizons do not show this relationship, however. Similarly for both Mg and Ca there is a tendency for the humus layers with higher contents of exchangeable base cations to have higher concentrations in soil solution (Figure 30). Again there is no such relationship for the mineral horizons.

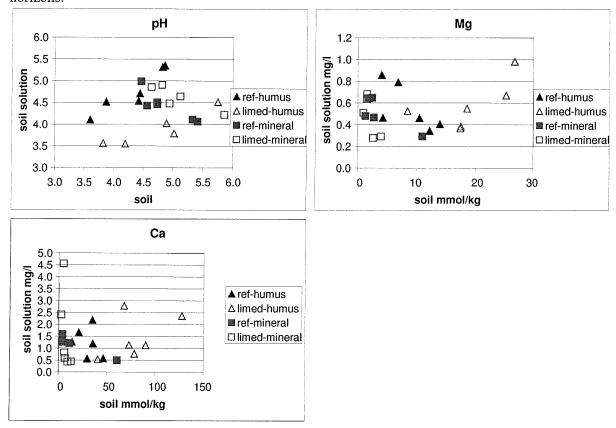


Figure 29. A comparison of soil and soil solution chemistry in the humus and mineral soil horizons at two catchments at Gjerstad. The soils were sampled in summer 1994. The soil solution data are means for samples from each lysimeter collected November 1994 through November 1997.

During the first 2.5 years following liming there have been very few indications of changes in soil solution chemistry. A small increase in Mg at only some of the locations has taken place. This was expected due to the slow movement of the dissolved base cations in other experiments (Brahmer 1994; Nohrstedt 1992). This contrasts with the rapid change of runoff water chemistry.

The most striking result from the Gjerstad experiment is the relatively uniform, acid and Al-rich runoff in the sub-catchments and the great variability in soil solution (**Figure 30**). This indicates that soil solution collected by the Prenart tension lysimeters is not representative of runoff. Similar observations on the lack of similarity between soil solution and runoff comes from other whole-catchment experiments such as the RAIN (Hauhs 1998) and CLIMEX (Jenkins *et al.* 1996) projects at Risdalsheia near Grimstad, about 60 km SW of Gjerstad and the NITREX project at Gårdsjön, near Gothenburg Sweden (Stuanes and Kjønaas 1998). In both these cases runoff responded more rapidly and consistently to changes in the chemical composition of deposition, while lysimeter data showed wide variations ranging from little or no response over many years, to immediate and persistent response. As at Gjerstad the lysimeter samples show great heterogeneity spatially.

There are several possible explanations for the lack of similarity between soil solution and runoff. First, the sampling technique for soil solution probably does not collect water in volumes proportional to that in runoff. Soil solution is comprised of water in pore spaces over a continuum of sizes, from macropores that have water only during saturated conditions, to micropores that retain water by capillary tension. The Prenart lysimeters under tension will collect water from a fraction of these pore sizes, but most probably some will be over-represented. The macropore flow will be underrepresented in volume. And since the lysimeters at Gjerstad were only evacuated 1 day each 14 days, the samples represent only a small time fraction as well.

Second, water movement varies greatly in time and space though heterogeneous uneven terrain such as is characteristic of the forested catchments at Gjerstad, and also Risdalsheia and Gårdsjön. The dominant pathway from soil surface to the stream is horizontal, not vertical. Incoming precipitation (or throughfall) percolates down through the soil undergoing chemical change underway until an impermeable surface is reached — either bedrock or a water-saturated zone. The water then moves horizontally towards the stream channel, and may alter chemical composition further, especially in the discharge regions near the stream. The runoff in the stream is a composite sample of water "packages" that have moved vertically and horizontally through the soil at a wide variety of distances and contact times.

Although lysimeter water is not representative for runoff, it may, however, give a good picture of the water quality available for uptake by roots. Thus the response of trees and ground vegetation to the liming treatment may first occur when the chemical composition of soil solution changes. At the NITREX experiment at Gårdsjön the runoff began to respond to N additions already during the first year of treatment, while the soil solution responded after 2-3 years and the vegetation yet later (Kjønaas *et al.* 1998).

5.3 Runoff chemistry

In runoff, the most striking effects of the dolomite treatment at Gjerstad were increased pH and concentrations of base cations, slightly reduced Al transport and change of Al species from highly toxic to non-toxic species. These effects are probably due to dissolution of dolomite in the topsoil and subsequent ion exchange of base cations in the humic layer. Lateral surface and sub-surface flow transport excess and ion-exchanged Ca and Mg to the stream.

Rapid, almost immediate, improvements in the stream water quality were found. This occurred even though the finest fraction of the dolomite was removed before spreading, in contrast to the liming material used in e.g. the Gårdsjön liming experiment, where the particles were in the range 0-30 μ m, a particle size fraction that was almost completely removed from the Gjerstad dolomite.

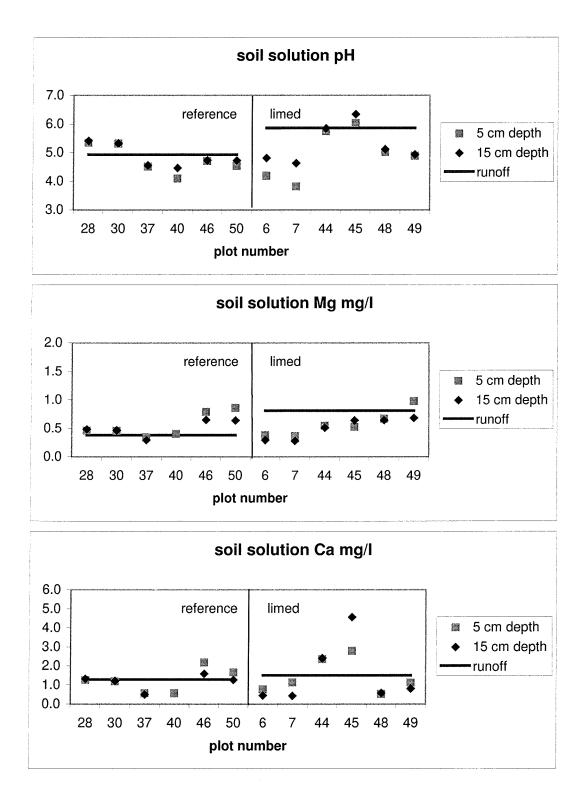


Figure 30. Comparison of mean concentrations in soil solution (6 sites at each catchment) and in runoff for samples collected during the post-liming period November 1994 through November 1997.

The post-liming increase in runoff concentrations of Ca and Mg was not significantly different in the Gjerstad experiment. Higher mobility of Mg relative to Ca was reflected in the relative increase. Whereas 0.5 % of the added Ca was transported out, 2.0 % of the added Mg was transported out of the limed catchment during the first 2.5 years.

Although the annual net transport of Ca and Mg was less than 1 % of the amount added, this was sufficient to cause a rapid response in runoff water quality after liming. The increase in water quality persisted throughout the period. No decreasing trends were documented after the initial increase in pH, base cations or ANC, and a longer monitoring period is needed to draw conclusions regarding duration and thereby mean annual liming costs.

These changes due to liming have turned the potentially toxic runoff into a water quality adequate for survival of acid sensitive organisms such as trout.

The results from Gjerstad generally agree with other comparable experiments with liming terrestrial areas. Hultberg *et al.* (1995) reported similar effects on runoff after application of 6 t ha⁻¹ of extremely fine-ground dolomite (particle size range: 0-30 µm) to a small catchment at the Gårdsjön research site, Sweden. A significant reduction of inorganic Al and Mn of 40-50 % was found. As in Gjerstad the mean annual flux of Mg was less than 1 % (7 % in 9 years). Fransman and Nihlgård (1995) also found increased pH, Ca and Mg and reduced Al, Fe and Mn after forest soil liming. Other experiments from Sweden with moderate lime doses (3-4 t ha⁻¹) showed only minor effects on pH and Al-concentration in stream water during a 3-year period (Westling and Skärby 1993).

In Schluchsee, Black Forest in Germany a minor increase in stream-water Ca was registered one year after liming with 4 t ha⁻¹ of pelleted dolomitic limestone, but the increase in Mg was as rapid as in Gjerstad (Brahmer 1994). Marked increases (after an initial peak) in both Ca and Mg in the soil solution at 30-cm depth were seen the year following liming. Increased Al-mobilisation at 30 cm was ascribed to the initial ion exchange processes after liming. Al did not increase in the stream water the first year after liming.

Kreutzer (1995) reported increased pH in the humic layer and in drainage water from this layer in the 6-year period after liming. No pH change in soil solution at 20-cm depth or drainage water from this layer was found. Drastic deprotonation of functional groups of the humic matrix was reported and both the cation exchange capacity and the base saturation increased. This stored buffer capacity in the topsoil. Six years after liming, dissolution of the added dolomite was regarded as 100 % and 70 % of the added Ca and 30 % of the Mg were still present in the humic layer. The low number for Mg reflects the more mobile nature of this element compared to Ca. Dissolution of lime was described by the equation: $m(t) = A^*e^{-0.643^*t}$, where A is the mass of carbonate added in tonnes, m is mass of carbonates remaining in tonnes, and t is time in years. According to this equation 80 % of the added dolomite in Gjerstad should be dissolved at the end of 1996.

Liming of the heathland Tjønnstrond-catchment in Norway with a 3 t ha⁻¹ dose of calcite powder resulted in a rapid and long-lasting increase in both pH and Ca and a more than 50 % reduction in reactive Al (Traaen *et al.* 1997). Liming of an upland, forested catchment (subcatchment IV) at Woods Lake, Adirondack Mountains, New York, USA, also resulted in improvement in water quality (Cirmo and Driscoll 1996). Liming of wetlands may increase the water quality of runoff, although the duration may be significantly shorter due to the more favourable dissolution properties of flushed bog surfaces and effective draining of dissolved liming material (Hindar *et al.* 1996).

Although the results from some of these experiments indicate only minor changes in pH, base cations and Al, even small water quality changes or differences may be of significance for the survival of

acid-sensitive fish and invertebrates. The significance of minor water quality changes may also be important for the duration of adequate liming effects on streamwater, tending to increase duration and thereby reducing costs if being part of a liming programme for aquatic systems.

Potential undesirable effects, such as increased NO₃ leaching and mobility of organically-complexed trace metals (Fe, Cu and Pb), may be expected after forest soil liming. So far, a significant but minor increase in the median concentration of NO₃ due to liming was found after 2.5 years of liming in Gjerstad. The seasonal variability was large and concentrations in the limed catchment were within the concentration range in the reference most of the year. A clear increase in concentration was found the second winter after liming. This "winter/spring" trend was also found at Tjønnstrond (Traaen *et al.* 1997). A possible explanation may be that liming in general stimulates decomposition of organic matter and that the result of this change is insignificant or masked during the first period after liming due to a delay in the build-up of the microflora and uptake of the produced NO₃ by the soil and vegetation during summer. Nitrification may still be significant in the snow-covered soil during winter, however, and the accumulated NO₃ from the dormant season may be flushed out with meltwater during spring melt, thus increasing the concentrations in late winter/spring. A longer data record is needed to draw any conclusion on the course of NO₃ in this experiment.

Warfvinge *et al.* (1996) summarised liming effects on runoff from several forested catchments in Sweden and found no increased N leaching during periods of 9-10 years. An alternative hypothesis to increased N leaching after liming was pointed out; increased N leaching due to shortage of base cations after several years of acidification was regarded as relevant for areas in Sweden that receive high amounts of atmospheric N. If liming would counteract this was not commented on, however.

At Schluchsee, increased NO₃ in soil seepage from the humus layer and both 30 and 80 cm soil depth were recorded (Brahmer 1994). As a consequence of the pH increase of two units in the O-horizon a shift in the population of nitrifiers from heterotrophic to autotrophic organisms took place (Feger *et al.* 1995). All mineralised N was converted to NO₃ in the limed topsoil, whereas NH₄ dominated the inorganic N fraction in the reference. However, no change in streamwater was found the following 3 years after liming (Brahmer 1994; Feger *et al.* 1995).

Kreutzer (1995) reported loss of N from the humus layer after 7 years of liming and also an increase of about 1 meq L⁻¹ in the NO₃ concentration in the drainage water leaving the root zone at 40-cm soil depth. The data record indicates a doubling of the NO₃ concentration at this depth due to liming already after one year. This is explained by a transformation of organically-bound N from the humus layer to dissolved organic N which was translocated to deeper soil depth and converted to NO₃ by stimulated nitrification in the mineral soil. No data for streamwater was presented, but increased concentrations of drainage-water NO₃ was calculated from soil solution data and a flux model. N deposition in Höglwald (12 kg N ha⁻¹) is of the same order of magnitude as in the Birkenes area but higher than at Solhomfjell closer to the Gjerstad site.

Increased mobility of potentially harmful trace metals complexed in the humic layer, especially Cu and Pb, may be expected after forest soil liming, if liming results in increased decomposition of organic matter (Kreutzer 1995). However, mobility of trace metals also depends on soil pH and increasing pH may stop leaching of metals related to accelerated soil acidification (Hüttl 1988). No increasing trends or elevated concentrations of the 10 trace metals relative to the reference stream were documented after the forest soil liming with dolomite in Gjerstad. This was probably due to the pH increase in runoff water, the moderate dose (3 t ha⁻¹) and low or moderate deposition of some of the trace metals in the area relative to e.g. German sites. Deposition of Cd, Cu and Pb in 1992-1993 at the Deuselbach monitoring station 200 km north of the Black Forest in Germany was higher than at Solhomfjell in Norway (Berg *et al.* 1996) by factors of 6.8, 4.6 and 1.5, respectively. Increased pH will increase the retention of some metals, whereas organically-complexed metals not will be mobilised if decomposition of organic matter is insignificant for this to occur. Hüttl (1988), in a

review of forest soil liming, concluded that the risk of trace metal mobilisation for limed soils is probably not more pronounced than for acidified soils.

The decrease in Mn, Zn and Co may or may not be desirable, depending on the resulting concentrations. Both Mn and Co are elements of importance for flora and fauna and increased retention, e.g. as the result of oxidation of inorganic monomeric Mn to MnO₂ at pH>5 and subsequent precipitation, may result in shortage. Mn limitation in lime-rich soils has been reported and appears to be common (Pearson and Adams 1967). In some acidified areas in Sweden large concentrations of dissolved inorganic Mn (> 1 mg L⁻¹) have been reported and a reduction is probably desirable. In Gjerstad, however, the concentrations are very low (5-15 μ g L⁻¹) and a reduction to < 5 μ g L⁻¹ after liming signals less availability of Mn for vegetation and forests and might be considered as a potential problem. For 985 Norwegian lakes the median Mn concentration was 2.5 μ g L⁻¹ (Skjelkvåle *et al.* 1999), about the same level as measured at LIM-1 and LIM-4 the last 1.5 years. Higher concentrations should be expected in streams with higher concentrations of organic matter such as is characteristic of forested ecosystems.

The liming effects on soil solution and stream water depend on hydrology, soil permeability and topography, i.e. the contact between runoff and limestone material. Steep slopes, thin soils and high precipitation, as in the Gjerstad area, was expected to promote rapid response in stream water due to anticipated dominance of overland and subsurface flow. Soil stratification in hillslope profiles is characterised by compact basal layers (Feger 1994), which also may promote lateral flow. Poorly mineralised humus, forming hydrophobic layers in the organic top layer of the coniferous forest may act in the same way. Less clear-cut effects in stream water in other forested ecosystems may be due to different topography, soil properties and climatic conditions. Possibilities of undesirable effects, like increased leaching of N and trace metals, is dependent on the deposition history of these elements in the actual sites. As deposition may vary to a great extent, e.g. between German and Norwegian sites, different results should be expected and this is indeed also what has been found. This confirms the rationale in selecting Norwegian sites for research activities before drawing conclusions on possible effects of forest soil liming in Norway.

Terrestrial liming, as the forest soil liming in Gjerstad and the alternative strategies represented by whole-catchment liming of non-forested catchments (Traaen *et al.* 1997) and wetland liming (Hindar *et al.* 1996), has certain advantages compared to more traditional (in Norway) lake liming and lime dosing methods. The Al-transport from the acidified catchment to watercourses will be reduced, and a more stable water quality is achieved throughout the year. Low doses of coarse-grained dolomite will probably minimise the undesirable effects on water quality. Terrestrial liming, also the forest soil liming concept, may therefore represent an interesting supplement to other liming methods for aquatic systems. A longer data record is needed, however, to draw conclusions on NO₃ leaching and costs.

5.4 Forest condition

An overall conclusion from the tree stand investigations is that the liming has so far shown no significant effects on tree growth and tree vitality (crown density and crown colour).

The causes for the general low level of increment could be several. In specific periods deviation from the function may occur due to e.g. climatic factors. The increment function can be said to represent the average climate for a long period. Another explanation could be that the function is based on forest stands that have been treated regularly with thinning and therefor represent high vitality and high «class» stands.

From other investigations we know that increment in old pine and spruce forests will be little or even negatively effected for more than 20 years after liming (Staaf *et al.* 1996). To evaluate the forest growth more detailed increment analyses should be performed 10 years after liming. Then increment

cores should be taken and a calibration to pre-liming increment could be done. Then more information on long-term effects of this low dose dolomite application can be extracted.

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Appendix A. Water and soil chemistry

Abbreviation and references for sampling stations:

LIM-1: Stream outlet of catchment 1 in the limed catchment Fugleliåsen LIM-2: Stream outlet of catchment 2 in the limed catchment Fugleliåsen LIM-3: Stream outlet of catchment 3 in the limed catchment Fugleliåsen

LIM-4: Main stream outlet of the limed catchment Fugleliåsen

REF: Stream outlet of reference catchment Spjøtåsen

Limed: 1=unlimed; 2=limed

FU-06 to FU-xx: Sampling stations for soil solution in the limed catchment Fugleliåsen SP-28 to SP-xx: Sampling stations for soil solution in the limed catchment Spjøtåsen

Tot P	1-101		1 01	ı Qı	က	17	13	25	18	17	က	2	8	ო	4	13	9	8	=	8	2	2	Ø	2	က	12	7	6	10	7	က	8	8	က
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	_	_																																265
NO3-N	1-1011	16	9	_	50	325	79	160	210	32	33	1	13	22	13	34	12	52	53	16	22	09	09	9	38	245	235	170	230	230	38	37	38	44
ANC1	ued L-1	15	22	2	37	22	9	15	37	28	24	우	5	10	26	30	4	က	7	30	14	7-	7	12	32	14	-12	ကု	ç <u>,</u>	=	∞	8	တ	10
SULF	ma L-1	4.1	4.3	4.0	2.6	4.5	4.0	4.4	4.4	3.8	4.2	4.0	3.8	3.9	3.6	4.4	4.0	3.8	4.4	4.6	2.0	4.5	4.5	4.6	4.7	5.0	5.5	5.0	5.5	5.3	4.8	4.6	4.7	5.0
ਹ	ma L-1	3.7	3.4	3.6	3.4	2.7	2.6	2.7	2.7	2.5	3.5	3.4	3.3	3.5	2.1	3.4	3.7	3.0	3.6	3.5	3.2	3.3	3.5	3.2	3.3	2.4	2.5	2.7	2.5	2.8	2.3	2.3	2.2	2.4
×	ma L-1	0.2	0.2	0.5	0.5	0.4	0.1	0.4	0.3	0.2	0.1	0.0	0.1	0.1	0.1	0.5	0.3	0.4	0.5	0.4	0.2	0.1	0.1	0.1	0.1	0.5	0.4	0.4	0.5	0.3	0.2	0.1	0.2	0.2
	σ	2.2																																
T0C	mg L-1	4	က	4	∞	12	Ξ	12	13	14	Ω	5	4	5	တ	12	9	တ	9	13	5	5	4	5	∞	10	တ	တ	9	9	∞	7	7	80
Alo	ng L-1	100	98	94	121	282	292	265	299	256	143	156	107	129	162	272	566	238	272	232	146	164	115	141	139	230	230	212	248	193	186	227	204	165
Alr	µg L-1	145	123	130	142	328	433	458	394	312	180	260	180	193	168	337	326	332	348	271	201	275	227	217	180	291	302	329	307	227	232	316	564	190
Mg	mg L-1	0.4	0.4	0.4	0.3	9.0	0.4	0.4	0.5	0.5	0.4	0.3	0.3	0.4	0.3	0.5	0.4	0.3	0.4	0.5	0.5	0.4	0.4	0.4	0.5	0.5	0.4	0.4	0.4	0.5	0.4	0.3	0.4	0.4
	mg L-1	1.4	1.5	1.4	 3	2.1	1.2	1.7	1.8	1.9	1.6	0.1	1.3	1.4	<u>.</u>	1.4	1.	1.0	1.3	1.7	1.4	1.0	1.2	 5.	1.6	1.2	1.0	Ξ	1.2	1.5	1:	Ξ:	1.0	1.2
Hd	-log (H+)	5.2	5.4	5.3	5.1	4.9	4.5	4.7	4.7	4.7	5.4	4.7	5.2	5.2	5.4	4.7	4.6	4.7	4.6	4.6	5.2	4.8	2.0	2.0	4.9	4.6	4.5	4.6	4.6	4.6	4.7	4.6	4.7	4.6
Limed		-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-				-
DATO		26-May-93	26-May-93	26-May-93	26-May-93	3-Aug-93	3-Aug-93	3-Aug-93	3-Aug-93	3-Aug-93	31-Aug-93	31-Aug-93	31-Aug-93	31-Aug-93	31-Aug-93	6-Oct-93	6-Oct-93	6-Oct-93	6-Oct-93	6-Oct-93	7-Nov-93	7-Nov-93	7-Nov-93	7-Nov-93	7-Nov-93	11-Nov-93	11-Nov-93	11-Nov-93	11-Nov-93	11-Nov-93	14-Nov-93	14-Nov-93	14-Nov-93	14-Nov-93
St. no.		LIM-1	LIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	H H	- IM-1	LIM-2	FIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	RE	LIM-1	LIM-2	LIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	REF	LIM-1	LIM-3	LIM-4	RFF

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Ì	µg L-1 6									09																								
Tot N	рg L-1 235	320	250	250	300	370	400	360	400	382	330	385	330	355	230	200	210	215	195	240	220	215	210	230	220	185	195	202	190	240	201	235	245	260
N-SON	µg Г-1 81	160	108	103	92	185	210	175	195	139	165	202	160	170	89	9	75	72	33	73	79	72	74	64	49	40	45	51	31	48	29	69	74	65
ANC1	peq L-1 22	Ņ	∞	16	52	58	4	33	18	56	က	-32	-18	-13	φ	-50	တု	-7	0	φ	-18	우	ι'n	8	∞	-7	0	0	6	6	-	က	8	Ξ
SULF	mg L-1 4.7	4.8	4.9	4.7	4.9	4.6	4.8	4.8	4.8	5.0	4.8	5.1	5.0	4.9	4.6	4.6	4.7	4.5	4.4	4.4	4.4	4.7	4.5	4.2	3.9	3.8	4.0	4.0	3.4	4.0	3.9	4.2	4.1	3.8
	mg L-1 2.5	2.7	2.6	2.5	2.6	2.5	2.6	2.5	2.5	2.5	2.4	2.5	2.4	2.4	2.1	1.9	1.9	2.0	- .	1. 8.	1.4	1.4	1.6	1.4	1.2	1.0	1.	1.2	- -	1.2	Ξ	1.2	1.2	1.2
	mg L-1 0.3	0.2	0.2	0.2	0.3	0.3	0.3	0.2	0.3	0.3	9.0	0.3	0.2	0.3	0.3	0.2	0.2	0.2	0.5	0.3	0.2	0.1	0.2	0.2	0.5	0.5	0.1	0.2	0.5	0.2	0.5	0.1	0.5	0.2
Na J	mg L-1 2.2	2.4	2.3	2.3	2.3	2.2	2.4	5.6	2.3	2.3	1.9	1.9	1.9	1.9	1.7	1.6	1.7	1.7	1.6	7.	1.4	1.5	1.5	1.4	1.3	1.2	 	 5.	1.2	1 .	1.2	1.4	1.4	4.1
TOC	mg L-1 5	9	2	5	9	2	S	2	വ	7	S	2	വ	വ	9	Ŋ	വ	9	S	9	ഹ	2	9	2	9	2	2	9	S	7	5	2	9	9
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Aľ.	рg L-1 197	287	270	237	176	175	271	580	229	171	185	239	271	214	186	224	255	506	142	177	200	234	197	136	162	187	211	186	113	183	204	225	193	141
	<u> </u>	0.4	0.4	0.4	0.4	0.5	0.4	0.4	0.4	0.4	0.4	0.3	0.3	0.4	0.4	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.2	0.2	0.3	0.5	0.3	0.2	0.3	0.3	0.3
Ca		6.0	1:1	Ξ	1.2	 	1.0	Ξ	1.3	1.4	- -	0.9	1.0	1:0	0.9	0.7	0.9	0.9	0.9	0.9	0.7	0.8	0.9	0.9	0.8	9.0	0.8	0.7	0.8	0.8	9.0	0.8	0.8	6.0
Hd (+D) od	-10g (n+) 5.1	4.7	4.9	4.9	4.8	5.1	4.7	4.8	2.0	4.8	4.8	4.6	4.7	4.8	4.9	4.7	4.8	4.8	4.8	5.1	4.7	4.8	4.8	4.8	4.9	4.7	4.8	4.8	4.8	5.1	4.7	4.8	4.9	4.9
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DATO Limed	13-Mar-94	13-Mar-94	13-Mar-94	13-Mar-94	13-Mar-94	29-Mar-94	29-Mar-94	29-Mar-94	29-Mar-94	29-Mar-94	11-Apr-94	11-Apr-94	11-Apr-94	11-Apr-94	18-Apr-94	18-Apr-94	18-Apr-94	18-Apr-94	18-Apr-94	25-Apr-94	25-Apr-94	25-Apr-94	25-Apr-94	25-Apr-94	2-May-94	2-May-94	2-May-94	2-May-94	2-May-94	10-May-94	10-May-94	10-May-94	10-May-94	10-May-94
St. no.	LIM-1	LIM-2	LIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	# :	LIM-1	LIM-2	CIM-3	LIM-4	LIM-1	LIM-2	LIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	H :		LIM-2	LIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	REF

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Tot	ud L-1	270	160	137	285	245	640	370	330	540	425	370	210	225	230	300	310	210	200	300	325	270	190	185	265	302	202	215	210	220	290	205	205
N-SON	ua L-1	35	32	20	40	20	53	7	16	30	တ	9	-	တ	10	9	9	6	12	6	9	က	4	15	7	9	∞	9	31	22	24	9	45
ANC1	uea L-1	15	ည်	6	- ∞	26	20	ī	-	24	31	22	က	2	23	35	13	-23	-12	=	F	22	ι'n	4	=	20	46	6	33	21	Ξ	27	41
SULF	mg L-1	4.0	4.0	4.4	4.0	3.5	7.3	4.1	4.4	5.9	4.9	5.1	4.1	4.5	4.8	4.9	4.1	4.3	4.8	3.9	4.5	3.4	3.4	3.9	3.4	3.5	4.1	3.6	4.2	4.1	3.5	3.6	4.3
ರ	mg L-1	ر. دن	د .	1.6	1 ئ	1.4	1.8	1.8	1.7	1.8	1.7	4.	1.7	1.6	1.6	. 8.	1.6	1.9	1.9	1.6	1.9	1.4	1.6	1.6	7.5	1.6	1.6	2.0	1.7	1.7	2.0	1 .	1.8
×	mg L-1	0.2	0.1	0.1	0.2	0.2	0.2	0.1	0.1	0.2	0.1	0.1	0.0	0.0	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.2	0.1	0.1	0.1	0.1	0.1	0.0	0.1	0.1	0.1	0.1	0.1
	Ö)																														1.4	
100	mg L-1	∞,	ß	4	9	7	17	12	9	Ξ	12	12	∞	9	7	우	유	7	9	တ	12	유	7	7	6	12	9	7	S	9	6	7	വ
Alo	pg L-1	144	149	120	13	129	569	263	246	254	237	231	199	144	182	186	210	198	174	217	193	508	185	167	207	176	143	152	139	161	153	178	133
Alr	hg L-1	183	222	203	196	163	383	348	364	363	284	307	288	256	566	228	280	278	281	295	231	256	257	267	276	216	184	223	173	173	186	216	162
Mg	mg L-1	0.3	0.2	0.3	0.3	0.3	0.7	0.3	0.3	0.5	0.4	0.5	0.3	0.3	0.4	0.4	0.3	0.2	0.3	0.3	0.3	0.3	0.2	0.3	0.3	0.3	9.0	0.5	0.5	9.0	0.3	0.5	0.5
S	mg L-1	6.0	0.7	1.0	0.8	1.0	1.9	0.9	[:	1.6	1.5	 	6.0	1.2	ل دن	7.	1.0	0.7	1.0	1.0	1.2	0.9	0.7	0.9	0.8	1.0	<u>.</u> ა	0.9	<u>.</u> დ	ر. ئ	1.0	1.1	1.5
표	-log (H+)	5.5	4.7	2.0	2.0	4.9	4.6	4.4	4.6	4.6	4.5	4.6	4.6	4.8	4.8	4.8	4.8	4.6	4.8	4.7	4.7	4.9	4.6	4.8	4.7	4.7	5.6	5.3	5.9	0.9	5.1	5.4	5.6
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DATO Limed		18-May-94	18-May-94	18-May-94	18-May-94	18-May-94	25-Aug-94	25-Aug-94	25-Aug-94	25-Aug-94	25-Aug-94	31-Aug-94	31-Aug-94	31-Aug-94	31-Aug-94	31-Aug-94	12-Sep-94	12-Sep-94	12-Sep-94	12-Sep-94	12-Sep-94	19-Sep-94	19-Sep-94	19-Sep-94	19-Sep-94	19-Sep-94	26-Sep-94	26-Sep-94	26-Sep-94	26-Sep-94	26-Sep-94	29-Sep-94	29-Sep-94
St. no.		LIM-1	LIM-2	FIM-3	LIM-4	HE !	- IW-1	LIM-2	LIM-3	LIM-4	REF	- IM-1	LIM-2	E-W-3	LIM-4	쀭	- I	LIM-2	FIM-3	LIM-4	HEF.	LIM-1	LIM-2	ღ-WII	LIM-4	REF	LIM-1	LIM-2	FIM-3	LIM-4	REF	LIM-2	LIM-3

Tot P	µg L-1		ı ro			9	က	4	വ	4	4	7	2	က	4								01												
N-4-N	µg L-1 µ	12	<u>6</u>	332	268	3	19	21	37	23	126	4	12	137	164	∞	5	5	9	13	27	9	2	19	30	19	9	2	19	19	99	∞	7	49	61
Tot	µg L-1 470	200	235	280	280	370	300	285	370	395	355	202	202	370	455	230	202	200	220	250	265	265	220	315	265	250	230	202	265	275	290	265	235	255	295
NO3-N	µg L-1	27	8	36	43	26	78	99	112	106	7	10	45	52	22	19	33	8	58	30	37	100	80	49	37	4	28	29	51	37	47	120	104	69	52
ANC1	peq L-1	18	43			29	34	49	51	83	42	27	41	33	28		30	38	58	15	75	47	30	28	38	70	33	33	54	52	22	30	56	53	15
SULF	mg L-1	4.4	4.9			3.7	3.4	3.5	3.7	3.7	4.0	3.6	4.3	4.0	3.6		3.8	3.9	4.2	3.8	4.6	3.7	4.4	4.0	4.3	4.0	4.0	4.0	4.2	4.2	4.1	3.9	4.2	4.3	4.6
ō	mg L-1	2.1	2.1			3.6	3.3	3.3	3.5	3.4	1.6	1 .8	1.8	1.7	1.7		2.9	2.8	3.3	3.1	3.1	2.7	3.0	2.8	2.9	2.8	2.7	2.7	2.9	2.9	2.6	2.5	2.7	2.7	2.9
×	mg L-1	0.1	0.1			0.5	0.3	0.3	0.4	0.4	0.1	0.1	0.1	0.1	0.1	0.3	0.5	0.1	0.5	0.5	0.2	0.1	0.1	0.2	0.1	0.5	0.1	0.1	0.1	0.1	0.2	0.1	0.1	0.1	0.1
Na	mg L-1 1.6	1.5	1.7	1.6	1.6	2.0	1 .	1.9	1 .	2.1	1.5	1.4	1.5	1.5	1.6	1.8	1.8		1.8 8.	1.9	2.2	1.9	1.9	1.9	2.2	2.1	1.9	1.9	2.0	2.1	6 .	1.	1.8	1.9	2.0
T0C	mg L-1	9	4			6	∞	80	∞	10	ω	7	2	7	F	∞	9	9	7	ω	7	9	9	9	တ	7	9	5	7	∞	6	9	2	9	∞
Alo	hg L-1 123	173	112	92	150	194	189	206	193	164	169	178	133	146	165	182	182	177	189	164	167	184	175	154	157	176	178	166	193	161	156	180	177	158	156
Alr	µg L-1 128	194	134	86	161	223	239	251	227	212	184	216	162	154	192	506	227	224	220	506	187	226	219	172	202	193	217	210	212	202	164	219	200	166	187
Š	mg L-1 0.6	9.0	9.0	9.0	0.3	0.8	0.7	0.7	0.7	0.5	0.5	0.5	0.5	0.5	0.3	0.7	0.7	9.0	0.7	0.4	0.8	0.7	0.7	0.7	0.5	0.7	0.7	9.0	0.7	0.4	0.7	0.7	9.0	0.7	0.4
	mg L-1 1.8	- -	1.7	1.6	1.2	1.7		1.6	1.6	5	4.	-	7.5	4.		_ .5	1.2	1.4	ر .	 ნ.	e:	<u>.</u> ა	ل نۍ	1.6	1.6	9.	1.2	1.4	۔ ئ	4.	1.5	1.2	<u>_</u> ა	1.6	4.
Hd	-log (H+) 6.5	5.5	5.8	6.4	5.6	5.3	2.0	5.5	5.3	4.7		5.4	5.6	5.9	5.2	5.5	5.2	5.4	5.4	4.7	5.9	2.5	5.5		4.9 6.4	. 53 6 7 8 9	51 32	5.5	2.7	4.9	0.9	5.5	5.5	5.9	5.0
DATO Limed		0	α .	C)	-	0	α .	C)	0	(α (CV (N (21	. .	CV ·	2	0	α .	-	0	7	0	7	 (01 (CJ :	≈	0	-	CJ	7	0	7	-
DATO	12-Oct-94	12-Oct-94	12-Oct-94	12-Oct-94	12-Oct-94	27-Oct-94	27-Oct-94	27-Oct-94	27-Oct-94	27-Oct-94	31-Oct-94	31-Oct-94	31-Oct-94	31-Oct-94	31-Oct-94	3-Nov-94	3-Nov-94	3-Nov-94	3-Nov-94	3-Nov-94	14-Nov-94	14-Nov-94	14-Nov-94	14-Nov-94	14-Nov-94	21-Nov-94	21-Nov-94	21-Nov-94	21-Nov-94	21-Nov-94	6-Dec-94	6-Dec-94	6-Dec-94	6-Dec-94	6-Dec-94
St. no.	LIM-1	LIM-2	r IM-3	LIM-4	보 :	- <u>-</u>	LIM-2	E-IM-3	LIM-4		- <u>W</u>	Z - K	ρ. Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε.Ε	LIM-4	보 :	- <u>I</u>	CIM-2	LIM-3	LIM-4	· 上:	LIM-1	LIM-2	. LIM-3	LIM-4	世 王 王		LIM-Z	. LIM-3	LIM-4	Ä	-1 -1	LIM-2	ΕΙΜ-3	LIM-4	REF

Tot P	µg L-1 4	t -		- ~	က	က		. —	٠ ۵	က	က	-	-	-	٠ ۵	8	8	2	N	8	8	-	-	-	7	8	-	-	α	8	-	-	-	-	-
	µg L-1	1 - rc	ייי כ	2 2	23	19	2	, LC	58°	4	53	2	2	32	24	14	32	10	16	41	7	∞	12	Ξ	34	25	88	∞	29	67	2	- ω	. ro	22	32
Tot N	µg L-1 290	3,5 7,5 7,5	205	780 780	270	260	225	210	270	295	230	245	225	250	255	315	325	295	305	340	245	305	285	270	310	415	430	305	415	375	220	285	270	290	265
	рg L-1 77	73	73	87	71	8	88	77	92	82	4	109	66	79	29	165	175	160	165	160	119	180	170	150	145	220	260	185	230	175	103	149	155	149	101
ANC1	peq L-1 35	22	27	် လ	Ŋ	43	24	23	19	6		7	58	4	20	38	30	59	30	ι'n	43	23	52	33	Ņ	58	15	16	58	-20	48	33	30	38	-
SULF	mg L-1 3.7	4.0	4.0	3.7	3.8	3.8	4.0	4.1	4.0	3.9		4.3	4.4	4.2	4.4	3.6	3.7	3.8	3.7	3.7	4.1	4.0	4.3	4.1	4.3	4.6	4.7	4.5	4.6	4.5	4.2	4.2	4.3	4.2	4.2
℧ :	mg L-1 2.7	2.6	2.6	2.6	2.7	2.8	2.8	2.8	3.0	2.7		3.0	2.9	2.8	3.0	4.2	4.7	4.0	4.2	4.2	4.2	4.4	4.0	4.1	4.1	4.8	4.9	4.5	4.7	4.3	4.1	4.2	3.9	4.2	3.9
	mg L-1 0.2	0.1	0.1	0.2	0.2	0.2	0.1	0.1	0.1	0.2	0.3	0.1	0.1	0.2	0.2	0.3	0.2	0.5	0.2	0.2	0.2	0.2	0.1	0.2	0.2	0.3	0.2	0.1	0.2	0.2	0.4	0.3	0.2	0.3	
	mg L-1 1.6	1.7	1.7	1.6	1.7	1.6	1.7	1.7	1.7	1.7	2.5	1 .8	1 .	1 .8	2.0	2.5	2.4	2.2	2.5	2.2	2.1	2.2	2.2	2.5	2.2	2.3	2.5	2.3	2.3	2.5	2.2	2.3	2.2	2.2	2.2
TOC	щg L-1 8	9	5	7	∞	7	9	വ	9	7	9	9	വ	S	7	IJ	ß	5	4	5	4	Ŋ	4	4	IJ	9	4	4	ა	Ŋ	4	4	4	4	വ
:	ру L-1 186	•		197																															
Al-	197 197	216	220	211	201	174	202	216	189	183	151	209	202	183	179	177	200	216	188	192	144	182	181	155	171	138	172	190	151	171	146	175	190	153	165
Mg W	g L-1 0.6	9.0	9.0	9.0	0.4	0.7	0.7	9.0	9.0	0.4	0.9	0.7	0.7	0.7	0.4	0.8	0.9	0.8	0.8	0.5	0.9	0.9	0.8	0.8	0.5	1.0	1.0	0.8	1.0	0.5	0.9	0.9	0.8	0.9	0.5
Ca Ca	1.3 1.3	1.2	1.3	<u>_</u> ნ:	1.2	9.	 	1.4	 5.	1.3	1.8	1.3	4.	1.5	1.5	1.5	4.	1.4	4.	1.2	1.7	4.	7.5	1.7	1 .	 &;	1.5	1.6	1 .	ل ئ	1.6	1.4	1.5	1.6	 5.
Hd (+H) sol	5.6	5.3	5.5	5.5	4.8	5.9	5.4	5.5	5.6	4.9	2.6	5.3	5.5	5.7	4.9	5.8	5.4	5.4	2.6	4.7	6.1	5.4	5.6	5.9	4.9	0.9	5.5	5.5	5.9	4.7	0.9	5.6	5.5	5.9	4.8
Limed	2	2	2	0	-	0	0	α	7	-	7	0	0	7		CI	7	0	N	-	7	7	0	N	-	8	8	CV	7	-	8	0	α	8	_
DATO Limed	12-Dec-94	12-Dec-94	12-Dec-94	12-Dec-94	12-Dec-94	21-Dec-94	21-Dec-94	21-Dec-94	21-Dec-94	21-Dec-94	3-Jan-95	3-Jan-95	3-Jan-95	3-Jan-95	3-Jan-95	15-Feb-95	15-Feb-95	15-Feb-95	15-Feb-95	15-Feb-95	14-Mar-95	14-Mar-95	14-Mar-95	14-Mar-95	14-Mar-95	27-Mar-95	27-Mar-95	27-Mar-95	27-Mar-95	27-Mar-95	5-Apr-95	5-Apr-95	5-Apr-95	5-Apr-95	5-Apr-95
St. no.	LIM-1	LIM-2	LIM-3	LIM-4	표 표 :	LIM-1	LIM-2	LIM-3	LIM-4	REF	LIM-1	LIM-2	CIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	REF	LM-1	LIM-2	LIM-3	LIM-4	HH.	- IM-1	LIM-2	LIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	REF

Tot P	рg L-1	J -	-	. 01	2	က	,	۷.	8	N	က	က	က	8	0	4	0	8	က	က	7	2	က	ນ	9	9	က	က	2	4	7	4	4	വ	S
풀	µg L-1																																		
Tot N	µg L-1 235	205	210	235	202	225	190	200	210	185	230	260	220	210	265	325	215	180	260	325	355	225	175	320	375	345	225	210	230	275	325	260	215	305	355
N-EON	рg L-1 р 104	78	88	105	24	99	24	64	2	88	44	75	51	51	34	53	64	43	38	53	33	34	19	35	19	18	30	58	56	13	4	36	24	16	9
ANC1	peq L-1	27	21	4	4-	3	15	12	23	ထု	43	45	4	39	Ξ	26	30	4	45	Ξ	80	45	48	29	19	78	25	46	89	19	83	09	72	79	78
Ę.	mg L-1 4 8	4.0	4.2	4.6	3.8	3.9	3.9	4.1	4.0	3.6	4.0	3.1	3.7	3.9	3.3	3.8	4.2	4.6	3.8	3.2	3.4	3.9	4.2	3.6	3.3	3.2	3.4	3.7	3.3	3.1	3.4	4.0	4.2	3.4	3.3
Ö	mg L-1 4.3	2.7	2.8	3.9	2.9	2.5	2.0	2.2	2.4	2.3	1.9	1.6	1.9	1.9	1 .8	1.9	2.0	2.3	1.9	1.8	<u>+-</u>	6.	2.1	1.8	1.7	1.6	1.7	1.9	1.7	1.6	1.4	1.6	6.1	1.5	1.5
	mg L-1 0.3	0.2	0.2	0.3	0.3	0.3	0.3	0.5	0.3	0.3	0.3	0.2	0.1	0.2	0.2	0.2	0.1	0.2	0.5	0.2	0.2	0.1	0.1	0.2	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.0	0.1	0.1	0.1
N	mg L-1 1.9	<u>6</u>	1.8	1.9	1.7	1.5	1.3	1.5	1.5	1.4	1.5	1.4	1.6	7.	1.4	1.5	1.5	1.7	1.4	1.3	1.6	1.6	1.7	1.6	1.4	4.	1.4	1.5	1.5	1.4	1.6	1.6	6.1	1.7	1.6
TOC	mg L-1 5	5	Ŋ	5	S.	9	വ	9	2	9	9	9	9	2	9	_	2	4	9	7	9	7	9	တ	6	თ	∞	7	∞	တ	6	ω	9	∞	တ
Alo .	hg L-1 128	142	157	139	=======================================	129	139	171	139	108	138	158	167	144	117	156	150	111	152	131	181	174	155	181	149	185	182	180	197	162	170	173	145	174	147
, Ar	рg L-1 145	169	198	158	158	147	166	203	160	145	153	182	195	160	144	160	167	137	160	157	188	191	175	188	175	198	195	200	506	184	178	185	162	177	174
	mg L-1 0.8	0.7	9.0	0.8	0.4	0.7	9.0	0.5	9.0	0.3	9.0	9.0	9.0	9.0	0.3	0.7	9.0	9.0	9.0	0.3	0.7	9.0	9.0	0.7	0.3	0.7	0.7	9.0	0.7	0.3	0.7	0.7	0.7	0.7	0.3
	mg L-1 1.5	Ξ:	1.2	1.4	0.1	1.2	0.9	1.0	Ξ	6.0	1.2	1.0	1.2	-	6.0	1.4	Ξ:	1.5	1.2	0.9	.5	1.2	_ .3	1.4	1.0	4.		1.2	<u>_</u> ნ	0.9	4.	1.2	1.5	. 3	1.0
Hd	-log (H+) 6.0	5.7	5.5	0.9	4.9	5.9	2.7	5.5	2.8	2.0	0.9	2.8	2.7	0.9	5.1	6.2	2.8	0.9	6.1	5.1	6.2	5.8	5.9	6.1	5.1	6.0	5.9	5.9	0.9	4.9	6.3	6.1	0.9	6.2	5.0
imed	7	8	N	0		0	0	0	0	-	CV ·	0	CV ·	01	-	Ø	α	Ø	0	-	C/I	Ø	0	N	.	CI I	C)	7	7	-	0	C)	0	Ø	-
DATO Limed	16-Apr-95	16-Apr-95	16-Apr-95	16-Apr-95	16-Apr-95	25-Apr-95	25-Apr-95	25-Apr-95	25-Apr-95	25-Apr-95	8-May-95	8-May-95	8-May-95	8-May-95	8-May-95	26-May-95	26-May-95	26-May-95	26-May-95	26-May-95	6-Jun-95	6-Jun-95	6-Jun-95	6-Jun-95	6-Jun-95	13-Jun-95	13-Jun-95	13-Jun-95	13-Jun-95	13-Jun-95	27-Jun-95	27-Jun-95	27-Jun-95	27-Jun-95	27-Jun-95
St. no.	LIM-1	LIM-2	LIM-3	LIM-4	世 :	LIM-1	LIM-2	LIM-3	LIM-4	HET.	- III	LIM-2	E-IM-3	LIM-4	REF.	LIM-1	LIM-2	LIM-3	LIM-4	REF	- IM-1	LIM-2	LIM-3	LIM-4	· 注:	LIM-1	LIM-2	LIM-3	LIM-4	HET.	LIM-1	LIM-2	FIM-3	LIM-4	REF

Tot P	ug L-1	6	တ													4				·		2											7
N-4-N	ng L-1	7	6	7	10	10	17	7	22	2	3	5	വ	2	7	9	19	5	7	17	56	23	21	49	72	51	144	2	2	7	5	5	10
Tot	ng L-1	089	029	780	670	190	425	305	300	450	410	320	265	260	320	320	270	310	265	280	290	300	305	315	375	382	640	400	405	310	332	370	300
N-SON	ng L-1	395	340	260	400	126	29	24	61	73	21	15	16	37	21	∞	2	103	97	34	17	33	46	27	101	149	8	210	235	96	195	220	113
ANC1	peq L-1	99	33	42	64	26	63	45	42	28	17	65	54	49	89	12	89	49	44	26	20	72	22	13	62	23	22	47	42	18	70	47	19
SULF	mg L-1	8.9	9.4	7.4	8.6	8.2	8.9	5.2	5.7	7.4	7.0	5.3	4.6	5.2	5.1	5.0	5.3	4.2	4.9	5.1	4.8	4.3	4.2	4.1	4.1	4.4	4.1	4.3	4.3	4.1	4.7	4.8	4.5
ō	mg L-1	2.5	2.3	2.7	2.4	2.7	2.7	2.6	2.6	2.7	2.6	2.1	2.1	2.2	2.2	2.1	2.5	2.6	2.5	2.5	2.4	2.6	2.7	2.6	2.9	3.0	2.7	2.7	2.6	2.5	2.7	2.9	2.8
×	mg L-1	0.3	0.3	0.3	0.3	0.7	0.4	0.3	0.5	0.3	0.4	0.4	0.3	0.2	0.3	0.3	0.3	0.2	0.2	0.2	0.2	0.3	0.3	0.2	0.3	0.2	0.2	0.2	0.2	0.1	0.2	0.2	0.1
Na	mg L-1	2.1	1.9	1.9	2.0	2.0	1.9	1.7	1.7	7.8	1.9	1.6	1.7	1.7	1.7	1.6	6.	6 .	1 .8	1 .8	1.9	1.8	1.7	1.7	1.9	2.0	2.0	1.7	1.7	1.7	2.0	6. 6.	1.9
700	mg L-1	တ	5	7	∞	=	Ę	10	∞	9	Ξ	Ξ	6	∞	유	Ξ	6	∞	9	ω	9	တ	∞	တ	7	_	∞	9	9	7	9	ß	9
Alo	µg L-1	150	182	147	156	188	203	203	191	194	175	207	211	191	210	178	188	180	162	83	162	75	37	73	43	25	128	142	139	125	129	137	129
Alr	pg L-1	264	231	194	187	290	239	240	228	232	259	228	225	225	236	226	196	204	184	202	212	119	26	72	42	26	131	151	154	159	141	149	163
Mg	mg L-1	1.4	1.5	1.3	1.4	0.7	1.4	0.9	1.0	1.2	0.5	0.9	0.8	0.9	0.9	0.4	0.9	0.8	0.8	0.9	0.4	0.9	0.9	0.4	0.8	0.9	0.4	0.9	0.9	0.4	1.0	6.0	
O	mg L-1	2.7	2.3	2.3	5.6	2.5	2.5	4.	1.6	2.1	2.0	1.6	1.2	1.5	1.6	1.3	1.7	გ	1.5	1.6	1.4	1.6	1.5	<u>.</u> ნ	1.6	1.6	4.	1.6	1.5	1.4	 8.	1.6	7.5
Hd	-log (H+)	0.9	5.4	5.6	5.8	4.8	5.8	5.5	5.5	2.7	4.7	5.8	5.6	5.6	5.6	4.7	5.8	5.6	5.6	5.8	4.8	6.1	6.0	4.9	5.8	5.7	5.0	0.9	5.9	5.2	6.1	5.9	5.1
Limed		0	Ø	α	0	-	0	0	N	7	-	Ø	0	0	7		7	CI	N	0	-	0	8	-	0	N	-	2	8	-	0	8	-
DATO Limed	•	7-Sep-95	7-Sep-95	7-Sep-95	7-Sep-95	7-Sep-95	14-Sep-95	14-Sep-95	14-Sep-95	14-Sep-95	14-Sep-95	18-Sep-95	18-Sep-95	18-Sep-95	18-Sep-95	18-Sep-95	4-Oct-95	4-Oct-95	4-Oct-95	4-Oct-95	4-Oct-95	23-Oct-95	23-Oct-95	23-Oct-95	20-Nov-95	20-Nov-95	20-Nov-95	4-Dec-95	4-Dec-95	4-Dec-95	19-Dec-95	19-Dec-95	19-Dec-95
St. no.		- M-1	LIM-2	LIM-3	LIM-4	REF	LM-1	LIM-2	LIM-3	LIM-4	REF	LIM-1	LIM-2	LIM-3	LIM-4	RF	LM-1	LIM-2	LIM-3	LIM-4	띪	LIM-1	LIM-4	Ή	LIM-1	LIM-4	Æ	LIM-1	LIM-4	吊	LIM-1	LIM-4	REF

Tot P	µg L-1 4	. ro	4	က	က	13	, LC	က	က	က	က	က	က	က	က	Ŋ	4	22	က	2	က	15	5	7	=	თ	9	12	Ξ	တ	4	4	က	6	4	4
	рg L-1 24	45	65	52	52	37	5	<u> </u>	24	; 2	15	42	12	တ	36	42	32	85	32	27	83	275	163	210	160	147	258	56	17	83	83	Ξ	12	ಕ	7	12
Tot N	hg L-1 660	730	630	260	200	410	520	465	320	430	405	315	355	340	315	400	360	350	310	290	275	595	405	200	665	290	200	495	465	445	320	280	290	325	280	275
N-SON	hg L-1 455	490	340	365	320	160	325	280	121	235	220	88	148	155	20	106	101	45	77	77	58	27	38	22	24	36	12	45	49	43	6	13	12	19	28	19
ANC1	peq L-1 28	33	4	26	46	12	22	42	10				29	22	53	74	89	25	89	22	23	99	53	16	66	83	30	8	72	15	9/	29	20	69	99	14
SULF	mg L-1 5.8	5.8	6.0	4.4	4.6	4.6	4.0	4.2	4.1				4.3	4.2	3.9	3.9	4.0	3.8	4.2	4.3	3.9	4.2	4.3	4.0	9.9	5.8	5.1	4.5	4.4	4.1	4.5	4.4	4.3	4.4	4.3	4.3
	тер L-1 2.9	3.0	3.0	2.5	2.5	2.4	2.0	2.1	6.1				1 .8	1.8	1.7	1.6	1.7	1.6	1.6	1.7	1.5	1.8	1.9	1.9	. .	2.0	2.2	2.4	2.4	2.5	2.4	2.4	2.5	2.5	2.5	2.4
¥ ,	IIIG L-1 0.5	0.4	0.4	0.4	0.4	0.4	0.4	0.4	0.3	0.4	0.4	0.3	0.4	0.3	0.3	0.3	0.3	0.2	0.3	0.3	0.2	0.3	0.2	0.2	0.2	0.1	0.1	0.3	0.2	0.3	0.1	0.1	0.1	0.3	0.2	0.1
	111g L-1	1.9	2.1	1.9	1.9	9.	1.7	1.7	1.6	1.6	1.6	1.5	1.7	1.7	1.7	1.6	1.6	1.5	1.6	1.6	1.5	1.6	1.6	1.6	1.7	1.8	 8.	1.4	1.4	1.4	1.6	1.6	1.7	1.6	1.6	1.7
TOC -	6 6	9	∞	7	9	7	9	9	7	9	9	7	7	9	7	တ	∞	თ	∞	7	∞	∞	^	∞	14	12	5	14	13	13	10	∞	9	=	თ	თ
Alo	ру С-1 123	120	119	119	126	113	126	141	117	133	138	121	152	149	126	172	161	142	135	137	101	110	11	124	240	206	190	237	233	201	175	182	163	202	202	176
Alr	165	140	189	132	141	153	129	147	149	131	142	145	148	153	148	172	167	164	145	155	161	110	112	133	247	215	241	249	248	239	189	195	204	204	207	204
Mg	1.0	- :	9.0	6.0	6.0	0.4	0.8	0.8	0.3	0.7	0.7	0.3	0.8	0.8	0.3	0.8	0.8	0.3	0.8	0.7	0.3	0.7	0.7	0.3	1.2	-	0.5	Ξ:	1.0	0.4	1.0	0.9	0.4	0.9	6.0	0.4
Ca 1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-		- 8.	1 .8	1.6	1.5	<u>_</u> ნ	4.	1.3	1.0	<u>_</u> ნ	- - -	1.0	4.	ر دن	1.0	4.	1 ن	- -	1.4	_ ა:	1.0	1.4	<u>_</u>	Ξ	2.2	6.	1.6	- 8.	1.7	1.4	1.7	1.6	1.4	1.6	1.5	L .3
Hd (+H) pol-	5.4	5.6	4.8 8.	5.8	2.7	4.7	5.8	2.7	4.8	5.8	2.7	4.9	2.8	2.7	4.9	2.7	5.8	4.9	5.9	5.8	4.8	6.7	6.3	5.4	5.6	5.8	2.0	5.5	5.6	4.7	6.0	5.9	4.8	5.8	5.8	4.8
	2	8	1	8	α .		7	7		7	0	-	0	Ø	-	α	7		C)	0	-	0	C)	_	0	C)	-	7	0		7	7	-	N	N	-
DATO Limed	12-Jan-96	12-Jan-96	12-Jan-96	18-Apr-96	18-Apr-96	18-Apr-96	22-Apr-96	22-Apr-96	22-Apr-96	28-Apr-96	28-Apr-96	28-Apr-96	7-May-96	7-May-96	7-May-96	20-May-96	20-May-96	20-May-96	29-May-96	29-May-96	29-May-96	20-Jun-96	20-Jun-96	20-Jun-96	2-Sep-96	7-Sep-96	7-Sep-96	30-Sep-96	30-Sep-96	30-Sep-96	16-Oct-96	16-Oct-96	16-Oct-96	28-Oct-96	28-Oct-96	28-Oct-96
St. no.	LIM-1	LIM-4	<u></u>	LIM-1	LIM-4	± :	[₩-1	LIM-4	REF	LIM-1	LIM-4	世 :	LIM-1	LIM-4	REF	LIM-1	LIM-4	HE :	LIM-1	LIM-4	# :	LIM-1	LIM-4	HH:	LIM-1	LIM-4	HH.	LIM-1	LIM-4	REF	LIM-1	LIM-4	REF	LIM-1	LIM-4	REF

Tot P	hg L-1	က	က	4	က	က	8	8	2	9	4	4	4	2	4	12	7	တ	2	7	တ	17	12	33	7	16	80	12	വ	7	9	တ	4	Ω
N-44N	µg L-1 12	10	12	31	50	19	, ro	9	5	13	15	17	22	78	84	92	22	75	12	22	202	158	82	102	56	42	50	52	24	20	6	18	ത	14
Tot N	lug L-1 250	245	260	285	270	255	260	285	255	410	360	340	255	230	325	375	340	380	270	325	540	610	530	715	455	490	425	450	310	515	330	330	460	340
NO3-N	µg г-1 14	24	16	37	27	25	96	126	61	155	150	118	82	49	89	30	34	25	∞	9	56	∞	59	17	2	13	22	2	51	22	114	22	98	68
ANC1	<u>-</u>	61							19	99	46	4	47	S	45	ι'n	29	21	8	36	8	32	93	25	105	46	94	37	74	33	69	56	61	14
SULF	g L-1 3.9	4.0	4.0	4.3	4.4	4.2	4.1	4.2	4.0	3.4	3.6	3.6	3.3	3.4	3.8	3.3	3.2	2.9	3.5	3.1	3.7	5.9	4.3	3.0	3.9	3.1	3.6	3.3	3.8	3.6	3.8	3.7	3.8	3.7
<u></u>									2.3	2.6	2.6	2.7	3.2	3.2	2.5	2.9	2.1	2.2	1.7	1.9	2.0	2.3	2.2	2.3	2.2	2.3	3.0	3.0	3.0	2.8	3.0	2.8	2.9	2.7
₩ 7 -	0.2	0.1	0.1	0.5	0.1	0.1	0.2	0.1	0.1	0.3	0.5	0.5	0.5	0.5	0.2	0.2	0.1	0.5	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.1	0.3	0.3	0.5	0.1	0.2	0.1	0.2	0.1
Na 1-1-		1.6	1.7	1.7	1.7	1.8	1.7	1.6	1.7	1.6	1.6	1.7	1.8	1.8	1.5	1.5	1.4	1.4	1.5	1.4	1.6	1.4	1.6	1.6	1.7	1.6	1.8	1.8	1.8	6.	1 .	1.9	1.7	1.8
TOC	9 9	ω	6	∞	7	ω	9	9	7	7	7	7	9	വ	ည	7	∞	∞	တ	10	တ	12	12	15	12	7	12	13	∞	9	6	9	6	6
Alo	153	158	141	137	136	123	127	129	119	112	110	100	131	112	66	87	136	118	161	145	123	173	48	51	212	179	195	172	176	148	183	156	146	156
Alr	158 158	165	174	158	162	176	136	141	151	135	137	145	144	147	66	103	144	143	172	174	126	195	71	74	217	218	196	200	169	173	183	182	192	191
Mg ma L-1		0.9	0.4	0.9	6.0	0.4		0.8	0.4	0.9	0.8	0.4	0.8	0.4	0.7	0.3	0.7	0.3	0.8	0.4	0.8	0.4	0.9	0.4	1.0	0.4	- -	0.4	0.9	9.0	0.9	0.4	0.9	0.4
Ca ma L-1		1.4	1.2	1.4	4.	1.3	1.3	1.4	1.3	1.3		1.	1.3	=======================================	1.4		1.3	1.1	4.	1.2	1.6	<u>_</u> ნ	1.9	1.6	1 .9	5	1.7	4.	1.7	1.4	1.6	1.4	1.5	1.2
Hd (+H) bol-		5.9	4.8	6.1	6.0	4.9	6.1	0.9	5.0	5.8	2.7	4.8	5.8	4.8	6.2	5.5	6.1	5.1	6.1	5.0	6.3	5.2	0.9	5.1	6.2	2.0	6.2	2.0	6.2	5.1	5.8	4.9	5.6	4.7
		7		7	7	-	0	N	_	0	0	-	0	-	7	-	0	-	7	-	7	-	0	-	0	-	01	****	0		N	-	N	-
DATO Limed	7-Nov-96	2-Nov-96	2-Nov-96	21-Nov-96	21-Nov-96	21-Nov-96	4-Dec-96	4-Dec-96	4-Dec-96	8-Dec-96	8-Dec-96	8-Dec-96	3-Mar-97	3-Mar-97	5-May-97	5-May-97	23-Jun-97	23-Jun-97	2-Jul-97	2-Jul-97	7-Aug-97	7-Aug-97	4-Sep-97	4-Sep-97	18-Sep-97	18-Sep-97	14-Oct-97	14-Oct-97	4-Nov-97	4-Nov-97	10-Nov-97	10-Nov-97	13-Nov-97	13-Nov-97
St. no.	LIM-1	LIM-4	REF	-IM-1	LIM-4	REF	LIM-1	LIM-4	REF	LIM-1	LIM-4	REF	LIM-4	REF	LIM-4	REF	LIM-4	REF	LIM-4	REF	LIM-4	HE I	LIM-4	REF	LIM-4	REF	LIM-4	HE H	LIM-4	REF	LIM-4	REF	LIM-4	REF

700	l/su	j D				19.9	1 T	<u>.</u>	25.4	<u>;</u>				30.1			20.1			43.6			40.4	29.5	3.9	3.7	3.2	3.5	3.2	3.1	2.7	3.5	က	က	3.4	3.7	3.4
Alo	//	i c	•		27.1	500	9 8	414	367	450	2	372	414	454	266	712	320	752		702	612	853	463	389	165	157	135	147	136	144	121	152	138	142	127	136	142
Ā	<u> </u> /201	i D			282	294	362	481	39.5	479) -	402	435	480	624	722	432	762		782	650	810	497	429	411	439	444	445	296	792	552	029	461	428	416	458	468
¥	ma/i	l b				0.73	1.36	2																	0.18	0.18	0.21	0.19	0.18	0.23	0.22	0.32	0.43	0.28	0.24	0.28	0.26
Na	I/om	1.77				1.78	63	1 78	1.65	189	2	3.11	2.27				2.05		1.68			2.76	2.36		1.66	1.55	1 .	1.31	. 8:	1.95	1.96	1.8	1.66	1.54	1.69	1.97	1.8
Mg	ma/L	0.35				0.35	0.35	0.28	0.15	0.15	0.17	0.32	0.29	0.32	0.23		0.3	0.41	0.39	0.44	0.47	0.85	0.65	0.72	0.25	0.2	0.19	0.25	0.27	0.25	0.24	0.24	0.22	0.24	0.22	0.26	0.24
రొ	ma/L	.23	1.37	0.68	:	0.79	0.78	0.76	0.38	0.41	0.51	0.68	0.74	92.0	0.58		0.49	0.72	0.64	0.7	97.0	2.46	0.77	0.84	0.49	0.33	0.37	0.16	0.4	0.36	0.37	0.39	0.39	0.39	0.4	0.41	0.38
Tot-N	na/L) -				280	490	!																	116	117	105	102	66	66	78	84	92	92	113	120	66
NO3-N	na/L			4		19	102	က	99			0.5					0.05						0.05		0.5	0.5	7	0.5	Ŋ	ო	4	4	က	က	თ	4	4
N-4-N	na/L					83	54																		∞	7	က	12	က	က	က	က	ဖ	က	ო	က	ო
ï	mg/L					4.2	2.7																		2	4.8	4.6	4.8	4.8	4.8	4.5	5.5	4.4	5.4	5.4	6.3	6.1
804	mg/L	ı				4.4	4.6																		4.6	4.5	4.2	4.4	4	3.9	4.1	4.7	5.3	വ	5.3	4.8	4.6
ច	mg/L					4	3.3																		1.8	1.7	4.4	ر ن	3.5	3.3	က	က	1.4	1.4	1.7	1 .	2.8
ALK	mmol/L																								0.012	0.012	0.015	0.018	0.01	0.014	0.014	0.018	0.025	0.029	0.019	0.026	0.024
kond.	mS/m					4.58	4.61	4.31	3.36	3.71							4.52						2.08		2.68	2.56	23	2.05	2.92	2.73	5.6	2.65	2.36	2.24	2.36	2.42	2.53
Hd	-log (H+)			4.6		4.13	4.16	4.17	4.32	4.25	4.18	4.2	4.17	4.03	4.12	4.12	4.19	4.12	4.29	4.15	4.19	4.39	4.24	4.21	4.59	4.56	4.62	4.73	4.58	4.66	4.69	4.68	4.86	4.86	4.78	4.87	4.79
Limed		-	-	-	_	_	α	7	7	7	7	7	21	7	21	01	0	7	7	7	7	Ø	Q	7	-	-	-	-	-	0	N	CJ	0	Ø	7	N	01
Depth L	Ę	വ	വ	ß	ß	ស	വ	വ	2	ιΩ	വ	വ	ഹ	വ	ഹ	വ	വ	ഹ	വ	വ	വ	വ	S.	ည	5	<u>5</u>	5	ن	5	12	15	5	12	5	15	5	15
Date		8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	8-Sep-94	11-Sep-94	9-Sep-94	11-Oct-94	31-Oct-94	6-Nov-94	30-Nov-94	15-Dec-94	1-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95
ло.		90	1 90	•			•	•	_	-	90			•						90		•		•	9	_	_				.,	-	_	90	90	90	90
Loc.		2	2	ß	교	교	5	교	2	근	급	⊒ i	⊋ ;	⊋ ;	⊋ ;	구 [⊋ ;	2	₽ (2	급 i	근 :	급 ;	급 :	⊡ i	⊋ i	글 i	2 i	구 i	급 :	2	2	₽	2	급 :	근	2

10C	19/L	5 4	3.7	3.5	3.6	က	3.1	6.6	38	3.6																				6.5		5.9		9.9	5.7		7.3	
Alo	152	171	152	178	177	116	128	155	152	153				14	45											51	99		163	191	197	213	218	246		231	267	256
Ā.	194 403	498	486	494	483	405	389	364	457	448				35	37											41	83		244	273	274	460	446	443		460	464	465
¥ :	119/L 0.45	0.41	0.11	0.15		0.13	0.08	0.12	0.08	0.08											5.6											0.43						
Na	1/gm	2.42	1.92	1.87	1.88	2.92	1.62	1.86	1.85	2.04	1.78				1.87		1.33	1.59			2.06		1.73			2.92		1.66			1.32	1.94	2.06	2.04	1.46	1.38	1.84	2.04
Mg	0.24	0.32	0.31	0.31	0.33	0.34	0.37	0.35	0.38	0.38	0.29				0.32		0.13	0.15			0.33		0.3	0.51		0.64	0.55	0.3			0.22	0.34	0.3	0.26	0.25	0.23	0.25	0.25
ပီ ရ	0.34	0.42	0.37	0.36	0.4	1.37	0.47	0.42	0.41	0.4	1.1	1.05			1.02		0.49	0.56			1.2		0.4	1.72		1.92	1.76	0.79	0.71	0.5	0.23	96.0	0.55	0.51	0.38	0.36	0.37	0.38
Tot-N	108	117	96	78		96	95	138	110	-																						160						
NO3-N	1 C	က	0.5	വ	4	S	0.5	0.5	0.5	12	ო	က					35											4	α	-	4	∞		0.5	∞			
NH4-N	1 6	က	က	1		9	က	თ	9	က																						œ						
iS i	7.3	9	7	7.3		6.1	9.9	7	7.6	6.9																						5.2						
S04	5.4	4.4	4.8	4.8		Ŋ	4.4	4.8	4.4	4.4																						3.7						
ا ا	7.	4.3	2.3	2.4		2.2	5.6	1.7	2.7	5.9																						4.2						
ALK	0.029	0.016	0.024	0.028	0.027	0.028	0.026	0.033	0.03	0.023																					0.013					0.015	0.012	0.013
kond.			2.58																		2.62										2.12	3.42				2.44	2.71	2.83
Hq (##) pol-	4.66	4.69	4.79	4.85	4.84	4.94	4.91	4.99	4.9	4.84			4.03		3.85	3.85		4.01	3.76	3.83	3.93	3.89	3.94	3.73	3.71	3.64	3.7	4.62	4.58	4.56	4.69	4.44	4.53	4.47		4.67	4.66	4.67
Limed		2	8	CJ	7	7	7	0	α	0	-	-	-	-	Ø	α	0	7	7	Ø	7	CJ	8	N	7	01	7	-	-	-	-	-	0	CJ	0	0	01	Ø
Depth	5	15	15	15	15	15	15	5	15	15	2	2	ည	2	2	2	2	Ŋ	Ŋ	IJ	2	വ	Ŋ	വ	ល	വ	Ŋ	5	15	15	15	15	15	15	5	<u>က</u>	5	5
Date	0-May-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	9-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	8-Sep-94	1-Sep-94	9-Sep-94	11-Oct-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95	12-Jun-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95
9	06 1	7 90						•					07																					•	•			. 20
Loc.	IJ	P	2	₽	2	2	급	교	금	급	근	근	⊋ ;	구 :	₽	2	2	2	2	₽	ß	3	교	근	2	금	2	2	2	2	己	3	2	2	3	⊋;	급 ;	J.

TOC	1/500	9.6	9	ď	9 6	9 6	ò				8.9	5.6	7.2	5.5	3.1	8.9	3.4	2.8	4.2		<u>6</u> .	2.4	4.2	6.1	9.9	4.6	8.9	4	3.6	3.6	3.4	3.8	6.1	3.7	4
Alo	//	55. 1 45.	224	280	248	27.5	247	20 70 70	194	220	323	213	151	130	74	193	106	103	134		25	69	104	174	180	103	216	128	135	26	107	118	171	113	121
Ā	/01	1 4 1 8	437	503	544	57.5	7,78	0.04	479	418	446	459	162	139	82	194	113	112	153		29	11	127	174	188	127	250	161	150	127	136	140	192	140	154
×	l/om	i D									0.17	0.16	0.13	0.12	0.15	0.39	0.28	0.15	0.11		0.14	0.16	0.46	0.54	0.51	0.04		0.05	0.05	0.08	0.05	0.02	0.18	0.17	60.0
N e	l/om	i ñ					1 73	2 14	: i	1.92	1.91	2.05		1.82	1.44	1.62	1.79	1.96	1.98	1.93	1.59	1.65	2.05	2.1	2.1	1.81		1.84	1.76	1.65	 83	1.54	2.03	1.75	1.92
M	ma/l	0.22	0.23	0.26	0.3	62 0	0.34		0.31	0.29	0.32	0.3	0.42	0.31	0.32	0.37	0.42	0.38	0.37	0.5	0.38	0.14	0.54	0.5	0.46	0.56	1.34	0.57	0.52	0.5	99.0	0.62	0.7	0.56	0.55
బ	ma/l	0.35	0.34	0.41	0.38	0.37	0.6	0.45	0.39	0.38	0.38	0.35	2.64	2.25	1.92	1.86	2.35	1.97	1.91	2.22	1.86	2.16	2.88	2.24	2.01	2.21	5.37	2.34	1.94	1.67	2.58	2.29	2.75	2.05	1.94
Tot-N	na/L	È									230	160	210	175	122	202	121	66	108		26	6	200	185	185	165		126	128	138	113	125	210	135	141
NO3-N	na/L	2			0.5	0.5	<u>;</u>				0.5	2	0.5	0.5	0.5	0.5	0.5	0.5	0.5		ო	0	0.5	0.5	0.5	0.5	က	0.5	0.5	0.5	Ω	0.5	0.5	0.5	4
NH4-N	na/L	<u>.</u>									32	7	က	∞	က	9	က	က	က		ω	က	വ	Ŋ	വ	14		က	13	က	Ŋ	က	თ	9	ო
ខ	ma/L	.									თ	7.8	4.6	4.5	4.4	4	4.1	4.3	4.4		3.7	4.3	3.8	Ŋ	4.3	4.1		4.8	5.1	5.2	3.8	4.7	4.9	5.3	4.6
804	mg/L	•									4.3	4.3	6.3	5.4	4.2	5.5	4.2	3.7	4		5.1	5.2	7.4	5.6	5.5	7		2.8	5.2	5.3	_	5.7	9.9	S	4.6
ច	mg/L	r									2.3	2.4	5.6	2.2	1.4	5.6	3.5	5.8	2.7		7	1.7	3.3	2.7	2.7	2.3		2.4	2.3	2.4	3.4	2.9	3.6	2.8	3.1
ALK					0.002	0.012					0.025	0.015	0.089	0.02	0.08	90.0	0.068	0.067	0.051		0.07	0.087	90.0	0.072	0.067	0.05	0.08	0.058	0.07	0.052	0.046	0.056	0.058	0.073	0.059
kond.	mS/m				2.83	2.92					2.69	2.59	3.1	5.8	2.33	5.6	5.8	2.41	2.41		2.44	2.63	3.48	3.05	2.86	2.94	6.8	2.98	2.64	2.44	3.39	3.03	3.6	2.83	2.72
Ħ	-log (H+)	4.6	4.66	4.48	4.55	4.63	4.55	4.89	4.6	4.73	4.81	4.71	5.39	5.3	5.39	5.61	5.46	5.72	5.59		5.9	5.92	5.96	5.96	5.83	2.69	2.99	5.53	2.62	5.73	5.54	5.57	5.58	5.9	5.65
Limed	·	0	۲3	7	7	7	7	7	2	0	7	7	-	-		-		0	21	0	7	7	0	7	0	0	7	8	0	Ø	CJ	21	7	N	Ø
Depth	E	12	15	15	15	15	15	15	5	5	5	5	വ												വ				ស	വ	2	ည	ນ	2	വ
Date		0-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	2-Nov-97	8-Sep-94	1-Sep-94	9-Sep-94	11-Oct-94	31-Oct-94	6-Nov-94	0-Nov-94	5-Dec-94	1-May-95	1-Jun-95	13-Jul-95	8-Sep-95	15-Oct-95	2-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	7-Sep-97	16-Oct-97	2-Nov-97
ē.		07 1			02		•																							_		44	-	44	44
Loc.		2	2	군	근	己	근	3	급	교	2	⊋ ;	⊋ i	⊋ i	⊋i	⊋ i	⊋ ;	⊋i	2	₽ ;	2∶	2 ∶	2∶	급 :	2 ;	⊋ ;	⊋ ;	⊋ ;	급 :	2 ;	근	⊋	2	2	3

700	1/00	2 2	9	23	2.1	8.	9.	<u>~</u>	9.1	8.	1.7	2.3	2.5	6:1	1 .6	2.1	2.2	2.1	2.2	5.	3.8	2.2	2.5	2.3	1.9	2	2.2		1.9	1.9	1.6	1.5	1.5	2.1	1.9
Alo	1/211	48	45	92	83	74	99	5.	25	88	45	7	88	45	47	29	77	2	29	42	20	29	29	62	41	22	45	52	25	72	09	40	34	89	29
Alr	[/0]	82	82	62	97	94	8	83	65	47	69	93	6	42	75	129	115	9	80	22	76	80	108	86	74	88	89	32	78	8	98	61	99	83	90
¥	ma/l	0.17	0.14	0.12	0.07	0.05	0.04		0.15	0.07	0.04	0.18	90.0	0.14	0.04	0.17	0.03	0.03		0.03	0.02	0.07	0.03	0.01	0.14	0.13	0.18		90.0		0.1	0.1	0.03	0.08	0.05
S.	ma/L	20.2	1.83	1.61	1.75	1.94	2.01	1.62	1.56	1.62	1.86	2.04	5.09	1.72	1.78	2.66	1.86	1.74	1.69	1.71	1.54	1.99	1.73	1.92	1.68	1.65	1.37	1.47	1.76	1.51	7.	7.	1.88	1.75	1.87
Mg	ma/L	0.38	0.31	0.36	0.44	0.39	0.38	0.39	0.4	0.42	0.45	0.46	0.39	0.49	0.54	0.88	9.0	0.54	0.53	0.47	0.64	0.63	0.57	0.51	0.42	0.38	0.36	0.35	0.44	0.4	0.41	0.45	0.48	0.46	0.43
రొ	ma/L	2.64	2.34	2.06	5.64	2.26	2.18	2.08	1.91	2.27	2.65	2.39	1.98	2.12	2.36	3.99	2.78	2.3	2.05	2.48	2.66	2.86	2.41	2.05	3.45	3.3	2.76	2.34	3.25	2.4	2.25	2.5	2.83	2.62	2.48
Tot-N	na/L	128	102	96	87	75	09		29	69	77	66	8	78	72	93	83	7		7	74	125	104	88	72	95	8		74		48	81	89	₩	99
NO3-N	na/L	0.5	0.5	က	0.5	0.5	0.5	4	α	7	0.5	4	4	4	4	4	0.5	0.5		Ŋ	0.5	0.5	0.5	4	0.5	0.5	4	က	-	4	က	7	0.5	7	0.5
NH4-N	ng/L		က	က	က	13	က		က	က	က	ည	က	က	2	ო	က	თ		S	က	∞	က	က	က	ω	က		က		က	က	က	က	က
ïS	mg/L	5.1	4.7	4.5	4.3	4.5	4.6		3.7	4.4	4.5	5.3	4.9	4.7	4.6	5.1	S	5.2		4.6	5.1	5.3	5.3	4.8	4.7	4.7	4.5		4.4		3.6	4.3	4.4	5	ល
S04	mg/L	4.4	4.1	5.1	4.5	3.5	3.7		5.3	5.4	7.5	5.6	4.3	9.9	9.9	13.4	6.3	5.1		6.7	5.9	6.4	5.3	4.7	4.1	3.6	2.8		4.8		4.7	4.4	7.9	5.4	4.4
ច	mg/L	2.3	- 8:	1.9	3.3	5.9	2.7		1.9	1.9	2.4	2.5	2.5	2	2.5	3.8	2.5	2.3		2.8	2.8	3.2	2.7	2.9	1 .	- 8:	5.		2.4		1.9	1.7	8	2.1	2.5
ALK	mmol/L	0.137	0.115	0.088	0.081	0.09	0.08	0.079	0.073	0.09	0.064	0.091	0.09	0.084	0.068	0.048	0.083	0.098		90.0	0.083	0.083	0.086	0.071	0.166	0.17	0.161	0.146	0.13	0.11	0.097	0.121	0.073	0.104	0.115
kond.	mS/m	3.08	2.83	2.51	2.98	2.24	2.47	2.59	2.53	2.7	3.13	2.99	2.61	2.85	2.92	2.07	3.21	2.81		3.09	3.22	3.51	2.99	2.73	3.33	3.21	2.61	2.24	3.05	2.53	2.62	2.69	3.26	2.88	2.8
Hd	-log (H+)	5.51	5.44	5.91	5.48	5.79	5.85	5.37	5.87	5.93	5.95	6.01	5.96	5.91	5.95	5.66	5.73	5.75	6.22	6.11	5.71	5.88	5.94	5.69	5.65	5.62	5.81	6.81	5.73	6.53	5.86	6.03	5.91	6.14	60.9
Limed		-	-	-	-	Ø	ς,	CJ	7	0	0	0	0	01	0	Ø	8	Ø	α	Q	N	α	N	01	-	-	-	-	-	7	Ø	0	CΙ	7	Ø
Depth	E	15	15	15	15	5	15	15	15	15	15	15	15	15	5		5	12	15	15	15	15	12	15	2	S	S	വ	ည	വ	വ	വ	S	വ	2
Date		8-Sep-94	1-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	15-Dec-94	1-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	59-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	15-Dec-94	1-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95
0		44	44				`	44	_						44						44	44	44	44	42		45			5	. 45		45	42	45
Loc.		3	2	2	2	⊒ (2	3	⊋ ;	2	₽ (2 ∶	2 ∶	⊋ ;	2	⊋ ;	2	⊋ :	급 [2	2	₽ ;	근	₽ (₽ ;	⊋ ;	2	2	2	2	2	2	2	2	₽

202	, ,	η α	σ	σ. —		2.1	2.1	9	2.1	2.1	2.3	2.1	6:	<u>6</u>	1.7	2.1	1.7	7.	5.	1.6	7,	<u>t.</u>	4.	. 8.	9.	1.7	1.2	5.6	1.7	4.	Ø	4.	7.	9.1	1.7	1.7
Alo	1000	9 9 0	3 2	23	9 6	2	99	88	£	4	84	84	8	Ξ	9	20	20	20	14	8	4		Ξ	ន	8	2	22	88	32	35	2	4	13	5	15	15
Ar		194 194	<u> </u>	<u>~</u>	26	103	9	26	83	8	73	89	32	32	22	3	31	36	19	32	3	14	27	44	37	52	45	44	20	20	32	22	27	56	83	78
×	1/500	, F	0.06	0.11	0.04	0.04	90.0	0.04	0.07	90.0	90.0	0.01	0.26	0.23	0.27	0.18	0.12	0.11	0.1	0.1	0.11	0.2	0.05	0.14	0.1	0.1	0.0	0.12	0.12	0.08	0.1	0.08	0.1	0.09	0.08	90.0
Na	1/202	, r.	1.7	1.91	1.57	1.44	1.47	1.79	1.47	1.63	1.68	1.7	1.93	1.87	1.67	1.58	1.88	1.81	9:	1.79	1.63	1.67	1.87	1.78	1.92	1.68	1.75	1.64	1.91	1.63	1.57	1.81	1.6	1.76	ل 9:	1.81
M) /v	0.42	0.51	0.5	0.59	0.52	0.52	0.59	0.68	99.0	0.7	0.58	0.62	0.58	0.56	0.57	0.63	0.57	0.56	0.55	0.55	9.0	0.64	0.57	0.53	0.51	0.59	0.48	0.71	0.7	0.7	0.71	0.75	0.75	0.87	0.77
ఔ	l/pm	2.16	2.59	2.85	3.22	2.57	2.31	3.3	3.36	3.39	3.62	2.91	5.69	5.73	4.87	4.83	5.24	5.22	5.22	4.95	3.81	4.23	4.92	4.44	4	3.3	3.71	3.5	4.74	4.46	4.39	4.87	4.8	5.15	5.51	5.4
Tot-N	//	74	89	84	87	26	83	22	89	92	84	75	77	<u>8</u>	83	8	69	22	44		48	69	89	71	99	104	62	84	138	51	74	65	54	80	99	69
NO3-N	1/611	က	4	4	0.5	0.5	0.5	4	0.5	0.5	0.5	4	0.5	0.5	7	က	0.5	0.5	0.5	4	4	4	0.5	4	0.5	4	0.5	4	0.5	0.5	0.5	4	0.5	0.5	0.5	4
N-4-N	na/L	က -	က	ω	က	7	12	က	S	ß	വ	က	က	13	7	16	ဖ	က	က		7	9	ß	က	က	က	က	က	မ	တ	က	က	က	က	က	က
ïS	ma/L	4.5	2	4.7	Ŋ	5.2	5.2	5.2	5.2	5.3	5.4	3.6	5.5	5.2	5.1	S	5.2	5.1	വ		4.3	4.9	4.9	5.2	5.1	4.9	5.1	4	5,4	5.6	5.5	5.4	2.7	5.6	9	5.5
804	ma/L	4.8	5.6	8.4	2.2	4.1	3.6	9.9	6.1	5.6	5.4	2.3	Ω	Ω.	3.5	1.5	3.2	-	2.5		4.5	4.6	10.1	6.3	4.4	4.8	5.5	8.5	6.5	4.2	3.3	9	5.8	6.3	വ	3.3
ច	mg/L	1.6	2	5.9	2.1	8	2.3	2.4	2.5	2.1	2.5	2.1	വ	2.5	1.7	1.9	2.1	1.2	2.3		2.2	1.9	2.1	2.5	2.4	1.8	C)	C)	2.4	2.1	2.1	2.3	α	2.2	5.6	2.3
ALK	mmol/L	0.102	0.098	0.056	0.12	0.127	0.121	0.108	0.128	0.138	0.158	0.14	0.269	0.275	0.281	0.299	0.278	0.279	0.274	0.256	0.195	0.218	0.136	0.176	0.196	0.164	0.161	0.093	0.205	0.246	0.251	0.205	0.212	0.22	0.281	0.283
kond.	mS/m	2.63	5.9	3.53	3.22	2.78	2.46	3.51	3.46	3.48	3.56	2.91	4.77	4.59	3.91	3.45	4.01	3.8	3.7	3.77	3.62	3.81	4.4	3.85	3.5	3.35	3.46	3.58	4.31	3.93	3.5	4.22	4.2	4.46	4.65	4.14
표	-log (H+)	5.75	5.98	5.87	5.99	5.96	6.16	6.01	5.89	6.05	6.31	5.98	5.79	5.95	5.86	6.32	5.96	6.15	6.23	6.65	6.28	98.9	6.36	6.42	6.35	5.99	6.43	6.16	6.1	6.25	6.67	6.56	6.12	6.41	9.9	6.34
Limed		7	ca	Ø	01	ત	0	01	CI	0	ત્ય -	8	_	_	_	-	-	7	7	7	0	7	α	N	0	01	CJ :	2	7	0	0	7	7	7	7	7
Depth	E	2	2	ည	വ	S.	ις.	വ	22	S)	2	ည	र्	5	ट	15	15	5	15	5	5	15	5	12	5	5	<u>र</u>	5	15	15	15	5	15	15	15	15
Date		0-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	8-Sep-94	1-Sep-94	19-Sep-94	11-0ct-94	31-Oct-94	16-Nov-94	30-Nov-94	15-Dec-94	11-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	2-Nov-97
9		45	45	45	45	5 1				•											_					•		45		45	45	45	45	45 1	45	. 45
Loc.		급	<u>.</u>	2 i	⊋ i	2 i	⊋ i	구 i	⊋ i	⊋ i	⊋ i	2 i	2 i	⊋ i	⊋ i	2 i	⊋ i	2 i	⊋ ;	2	⊋ i	⊋ i	⊋ ;	⊋ ;	⊋ ;	⊋ ;	2 i	⊋ ;	급 :	급 -	2 ∶	2	2	급	2	D.

Ĺ) 6	1 F	r 60	ς α <	o F	4.1	- 1	ř		o:	. 4 . 7.	?	4.1	3.5	8	6	27	š	5.5	4.2	3.1	2.7	2.4	2.5	2.2	2.9	2.7	4.6	3.1	2.5	က		3.7	က	5.6	2.3	က	3.3
οĮδ] [1 6	3	121	- -	117	0	3 5	3 2	1 2	9 9	2	88	98	20	88	2 92	>	75	83	83	99	67	79	ß	8	23	107	87	75	94	141	88	8	26	5	29	89
Δir	1/51	5 6 1 6	2	187	3	184	194	530	244	188	190	2	192	196	211	181	183	2	117	148	136	168	282	508	196	162	153	190	191	191	166	230	213	213	221	509	184	190
¥	- F	75.	2.24	0 63	9	-	0.52	10:0		0.68	0.39		0.18	0.21			0.07		0.28	0.2	0.19	0.13	0.43	0.22	0.19	0.94	0.21	0.41	0.5	0.31	0.08		0.11		0.26	0.05	0.04	0.23
e N	[/em	163	128	1 74	• •	1.22	1.52	186	90.0	1.86	1.66		1.61	1.68	1.61	2.03	4		1.56	1.47	1.2	1.17	1.86	1.62	1.66	1.38	1.52	1.6	1.74	1.9	1.61		1.64		1.65	1.67	1.45	1.89
Ma) /um	68.0	1,34	0.84		0.44	0.39	0.52	0.57	0.88	0.66	:	0.71	0.82	0.92	0.7	0.57	•	0.67	0.59	0.51	0.43	0.76	0.56	0.58	0.44	0.43	0.67	0.61	0.7	0.61	0.69	0.71	9.76	0.77	0.67	0.67	0.77
යී	ma/l	0.94	1.54	0.88))	0.36	0.39	0.5	0.46	0.66	0.51		0.47	0.56	0.58	0.58	0.44		0.71	0.57	0.57	0.52	0.86	0.64	0.73	0.53	0.54	0.61	0.59	9.0	0.57	0.61	0.55	9.0	0.59	0.58	0.53	0.48
Tot-N	na/L	455	230	225		215	185	2		620	190		175	123			160		220	190	131	120	130	113	98	117	123	190	132	202	130		150		102	104	125	147
NO3-N	na/L		Ŋ	30	r !	33	9	117		470	12		0.5	0.5		=			4	-	4	6	13	7	7	ည	9	52	œ	110	တ		4		4	ß	0.5	0.5
NH4-N	na/L		32	7		31	ıO	•		22	∞		က	13			ιΩ		12	7	က	-	က	က	က	က	က	თ	ო	თ	Ŋ		က		က	2	က	က
ï	mg/L	4.7	വ	4.2		3.3	5.2			9.9	7.1		5.3	5.6			6.8		4.8	4.6	3.7	4	4.3	3.7	3.9	3.4	4.7	5.6	5.4	5.9	6.5		5.2		5.6	5.9	6.4	5.9
S04	mg/L	7.7	5.3	3.7		4.2	4.4			4.1	4.5		4.9	4.3			4.6		5.6	4.8	4.6	4.1	2.8	2.5	3.7	4.8	4.6	4.8	4.2	4.3	4.4		4.8		4.5	3.8	4.9	4.1
ច	mg/L	2.3	6.	4		1.4	-			2.9	2.4		1.9	3.6			1.4		1 .8	<u>-</u> ნ:	1.2	1.6	5.9	- 6:	2.9	1.8	6.0	α	5.6	5.8	2.1		1.9		3.5	3.1	7.5	3.1
ALK	mmol/L	0.038	690.0	0.02		0.034	0.038		0.029	0.037	0.027		0.024	0.025		0.03	0.028		0.033	0.032	0.032	0.028	0.018	0.022	0.024	0.035	0.041	0.034	0.03	0.041	0.03		0.036		0.033	0.03	0.042	0.042
kond.	mS/m		4.23	2.84		2.11	2.01		2.74	3.04	2.52		2.56	2.86		2.62	2.11		2.47	2.36	2.13	2.02	3.31	2.3	2.36	2.5	2.03	2.41	2.41	2.67	2.29		2.45		2.5	2.52	2.25	2.54
Ħ	-log (H+)	5.04	5.09	4.91	4.97	5.08	5.16	4.85	5	5.01	5.11		4.97	4.96	4.88	5.21	5.09		4.97	4.92	2	2	4.75	4.93	4.97	5.17	5.24	5.17	5.04	4.99	5.13	5.05	5.1	5.14	5.18	2.07	5.18	5.43
Limed		-	-	۲3	α	α	0	α	α	0	0	7	Ø	01	0	63	0	7	-	-	-	-	_	7	7	01	Q	Q	0	0	C)	0	8	7	α	7	Ø	Ø
Depth	Ę	2	2	2	Ŋ	5	ß	5	S	S	വ	വ	2	വ	വ	Ω	Ω	S.	15	15	15	15	15	15	15	15	15	15	ا	15	5	15	15	15	15	15	15	5
Date		8-Sep-94	11-Sep-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	16-Oct-97
9		48	48	48		•		48							•										• •	•				•		48	48			48	48	48
Loc.		₽	2	₽	2	3	₽	Ð	5	2	2	급	근 :	₽	2	근	2	3	근	2	2	2	2	근 .	2	2 ;	2 ;	2 ∶	2 i	2	급 :	근	근	근	2	2∶	2	<u> </u>

201	1/200	j j			11.0	! C	† 2		17.5	17.4	7.7										4.4	:						5.5	;				9.7		8.3
Alo	7	J 9			257	31.5	2	334	447	429	183	258	314								85	121	i !		132	133		120		134			156		184
Ā	/61	ı D			350	390	3	453	5.05	598	277	387	434								145	163)		184	202		207		238			234		309
×	l/bm) D			er.	2															0.88) !													
Na	ma/l	1. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5.			1.86	163	1.63	1.89	193	1.94		2.15			6		1.57				1.98	88.		1.43		1.58				9.1		1.83	1.59		1.81
M	l/bm	5.0			1.13	0.65	0.66	1.05	0.58	0.55	1.68	1.05	1.43	!	1.16		0.53				0.62	0.49	!	0.57		0.39		0.98		9.0		0.91	9.0		0.91
చి	ma/l	0 0	0.86	0.55	1.33	0.81	0.87	1.23	0.87	0.96	1.79	1.04	1.35		1.22		1.07	0.99	0.57	1.19	1.02	0.83		0.82		0.71		1.23		0.64		0.83	0.68		0.75
Tot-N	na/l	2			1300																305														380
NO3-N	na/L	415	455	106	825	137	<u>.</u>		595	124	2840						23	51	5		120							1080							21
NH4-N	na/L) -			74	0	•														9														
ï	ma/L)			2.9																5.1														
804	ma/L	,			6.4																3.4														
ច	mg/L				4.8																വ														
ALK	mmol/L							0.016	0.037	0.046											0.014				0.031	0.034									0.028
kond.	mS/m				5.08	3.67		5.3	3.79	2.76											3.23				2.71	2.27									3.12
Ħ	-log (H+)	4.96	4.95	2.07	4.59	4.79	4.82	4.71	4.96	5.05	5.01	4.8	4.65	4.71	5.01	5.26	5.03		5.05		4.68	4.91	4.86	4.93	4.95	5.03	4.88	4.93	5.22	4.78	4.81	4.84	4.99	5	4.93
Limed	•	-	-	-	-	۲3	2	α	7	α	2	7	7	7	۲3	α	-	-	-	-		8	۲3	7	8	7	7	7	7	8	7	7	7	c۷	Ø
Depth	ᇊ	Ω	ιΩ	Ŋ	ည	Ŋ	Ŋ	ည	വ	ω.	ω	Ω	ιΩ	S	വ	Ŋ	5	15	15	15	15	5	15	15	15	5	15	15	15	15	15	15	15	15	15
Date		8-Sep-94	11-Sep-94	19-Sep-94	31-Oct-94	16-Nov-94	30-Nov-94	15-Dec-94	11-May-95	21-Jun-95	10-May-96	3-Oct-96	24-Oct-96	13-Nov-96	8-Jul-97	16-Oct-97	8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	15-Dec-94	11-May-95	21-Jun-95	28-Sep-95	10-May-96	12-Jun-96	3-Oct-96	24-Oct-96	13-Nov-96	8-Jul-97	7-Sep-97	16-Oct-97
no.			•	49	-	•	•		_	49	_											•	•	•			-					49 1		49	•
Loc.		근	IJ	3	2	3	Ð	IJ	2	2	2	2	2	근	₽	3	3	₽	Ð	3	근	2	2	Ð	3	2	2	2	근	3	2	근	2	2	5

T0C	ma/L	က ထ	3.0	3.9	4.5	3.4	3.2	3.2	3.9	3.4	4.5	4.6	3.8	3.7	3.2	3.9		3.7	3.4	3.5	3.3	3.4	3.9	4	က	3.9	4	4	3.7	3.2	3.2	က	3.8	3.3	4.1	4.1	3.6	3.5
Alo	ua/L	13	122	129	141	121	135	119	153	135	148	144	133	128	129	11	133	124	135	127	5	110	113	118	110	113	114	130	125	117	133	121	162	135	145	132	127	131
Air	uq/L	171	198	190	220	220	222	225	250	218	212	219	202	217	191	193	208	226	224	231	198	166	159	184	200	162	185	185	199	526	233	222	274	218	199	195	216	224
×	mg/L	0.11	0.09	0.14	0.14	0.24	0.22	0.21	0.27	0.19	0.07	0.04	0.2	0.19	0.32	0.13		90.0	0.12	0.15	0.0	0.03	0.08	0.1	90.0	0.11	0.08	0.14	0.1	0.2	0.22	0.22	0.28	0.21	0.08	0.03	0.19	0.19
Na	mg/L	1.69	1.51	1.23	1.46	1.74	1.86	1.93	1.88	1,5	1.33	1.55	2.01	1.95	1.63	1.56		1.61	1.67	1.55	1.56	1.49	1.83	1.52	1.79	1.73	1.62	1.23	1.67	1.84	1.95	1.99	1.96	1.56	1.35	1.62	5.09	2.05
Mg	mg/L	0.48	0.42	0.38	0.39	0.56	0.49	0.51	0.53	0.38	0.31	0.29	0.55	0.56	0.5	0.44	0.41	0.54	0.53	0.53	0.4	0.43	0.46	0.44	0.5	0.5	0.46	0.39	0.42	0.58	0.51	0.53	0.59	0.39	0.33	0.3	0.58	0.56
రొ	mg/L	1.32	1.17	-	1.03	1.4	1.32	1.41	1.43	1.02	6.0	0.91	1.4	1.39	1.39	1.19	1.41	1.4	1.38	1.3	1.21	1.2	1.32	1.13	1.3	1.42	1.36	1.06	1.15	1.43	1.38	1.48	1.52	1.08	0.97	6.0	1.57	1.42
Tot-N	hg/L	132	146	150	270	200	185	195	202	147	138	260	195	345	490	190		145	123	160	149	125	360	1775	325	135	144	140	180	520	190	138		155	138	180	185	230
NO3-N	µg/L	4	4	IJ	111	69	89	90	65	20	ဖ	80	75	215	415	20		∞	13	ဗ္တ	23	0.5	61	0.5	500	9	9	7	24	8	9/	26	92	54	∞	99	72	120
NH4-N	hg/L	က	က	က	Ξ	က	က	က	က	က	က	က	თ	က	က	∞		က	4	က	S.	က	40	က	က	က	က	က	6	က	က	က		က	က	က	က	ო
ï	mg/L	3.9	3.9	3.6	3.8	3.8	3.9	4.1	4.3	5.6	က	က	4.4	4.1	4	4		3.9	4.2	4.2	3.9	3.9	4.1	4.4	4.3	3.9	4.1	3.6	4	3.9	3.9	4.1	4.5	2.7	က	2.7	4.5	4.1
804	mg/L	6.7	6.2	4.2	5.3	4.1	4.1	4.5	6.8	4.2	3.9	4.7	5.6	5.1	4.7	5.3		5.5	5.4	4.8	4.7	4.4	4.6	4.1	4	6.1	5.6	4.5	5.4	4.3	3.8	4.8	4.9	4.5	4.3	S	5.6	5.4
ਠ	mg/L	7	. 8:	<u>-</u> .	1.7	3.4	က	က	5.3	1.7		9.0	2.4	2.7	1.9	1.7		2.3	2.3	2.6	2.5	CI	2.5	2.4	2.8	1.7	<u>.</u>	<u>.</u>	1.6	3.4	5.6	က	3.7	1.7		0.7	2.4	2.6
ALK	mmol/L	0.04	0.041	0.042	0.035	0.037	0.035	0.034	0.039	0.043	0.051	0.039	0.044	0.037	0.041	0.033		0.042	0.045	0.043	0.048	0.045	0.044	0.047	0.04	0.043	0.044	0.042	0.046	0.038	0.036	0.043	0.041	0.043	0.053	0.042	0.054	0.047
kond.	mS/m	2.62	2.44	2.05	2.23	2.68	2.54	2.53	2.71	2.02	1.83	1.81	2.66	2.76	2.65	2.33		5.6	2.54	2.4	2.39	2.28	2.45	2.24	2.5	2.68	2.63	2.08	2.3	2.79	2.59	2.57	2.82	2.15	1.91	1.83	2.71	2.72
Hd	-log (H+)	5.15	5.11	5.19	5.24	5.14	5.24	5.21	5.33	5.4	5.56	5.51	5.33	5.23	5.18	5.29	5.14	5.34	5.32	5.39	5.3	5.43	5.6	5.49	5.33	5.13	5.1	5.14	5.39	5.11	5.22	5.33	5.36	5.4	5.56	5.52	5.36	5.45
Limed		-	-	-	-	-	-	-	-		-	-		-	-	-	_		-	-	-	-	-	-	-	-					-	_	-	-	-	-	_	-
Depth	ᇊ	2	വ	വ	Ŋ	വ	w	S)	S.	വ	Ŋ				S	2					ည	വ	2	2					ਨ	5	15	15			15	15	15	5
Date		8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	15-Dec-94	11-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	15-Dec-94	11-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95
ë				88	82	88	8	28		•	28				•						28								88	28	83	58	8	8	88	88	82	58
Loc.		SP	S	SP	SP	SP	SP	SP	SP	SP	SP	S	SP	SP	S	SP	SP	SP	SP	SP	S	SP	SP	SP	SP	S	S	SP	SP	S	SP	SP	SP	SP	SP	SP	S	SP

TOC	mg/L	. e.	3.5	3.3	3.4	3.5	3.3	2.8	3.2	3.4	3.7	က	3.5	3.6	3.6	3.2	5.8	3.1	2.8	3.4	33	3.9	3.9	3.3	3.3	က	3.4	3.3	3.4	3.4	2.7	3.1	3.4	3.3	2.9
Alo	ng/L	126	105	106	114	132	124	108	108	133	117	109	108	108	110	112	86	112	108	131	126	136	127	123	116	117	100	106	123	109	105	110	106	106	105
Aľ	rg/L	204	201	186	217	233	236	190	154	167	199	214	175	199	199	227	255	260	231	271	558	212	202	204	244	217	508	223	236	231	189	176	200	212	242
¥	mg/L	0.32	0.17	0.07	0.04	0.1	0.16	0.11	0.04	0.02	90.0	0.03	0.12	0.13	0.14	0.14	0.18	0.18	0.18	0.23	0.26	0.13	0.05	0.23	0.22	0.35	0.22	0.12	0.1		0.1	90.0	0.07	0.07	0.04
s S	mg/L	1.68	1.6	1.29	1.61	1.66	1.65	1.67	1.56	1.87	1.62	1.91	1.53	1.5	1.14	1.25	1.61	1.71	1.72	1.56	1.51	1.6	1.46	1.58	1.67	1.59	1.4	1.57	1.44	1.38	1.54	1.41	1.62	1.6	1.64
Mg	mg/L	0.51	0.46	0.36	0.54	0.54	0.55	0.43	0.47	0.48	0.48	0.52	0.45	0.42	0.38	0.4	0.57	0.49	0.49	0.5	0.36	0.41	0.29	0.56	0.58	0.49	0.44	0.47	0.5	0.52	40	0.41	0.44	0.43	0.5
రొ	mg/L	1.43	1.28	1.16	1.44	1.42	1.43	1.21	1.36	1.41	1.28	1.39	1.16	1.1	96.0	96.0	1.33	1.22	1.28	1.19	0.97	1.08	0.85	1.31	1.35	1.33	1.19	1.23	1.23	1.22	1.19	1.16	1.21	1.14	1.27
Tot-N	µg/L	490	155	195	138	117	132	122	202	202	155	310	120	141	129	202	190	180	147	185	147	126	190	190	232	535	165	125	107		175	125	220	144	330
NO3-N	hg/L	410	45	74	15	13	54	හි	7	20	12	195	4	ß	4	8	75	9/	26	82	44	4	26	83	125	445	51	4	Ŋ	16	40	-	25	∞	215
N-4-N	hg/L	က	7	က	က	Ø	က	വ	က	Ŋ	က	က	ო	က	က	∞	က	က	က	က	က	က	က	ß	က	က	Ŋ	က	7		က	က	14	က	ო
ï	mg/L	4	4.1	2.1	3.9	4.2	4.2	4	3.9	4.4	4.3	4.3	3.7	3.9	3.4	3.5	3.5	3.6	3.8	3.8	5.6	2.9	2.9	4.1	3.8	3.8	3.9	3.6	4		3.7	3.7	4	4	4
S04	mg/L	4.9	5.4	4.8	5.3	5.5	Ŋ	4.8	4.5	4.8	4.6	4.6	5.3	6.1	4.7	4.8	3.8	3.7	4.3	4.3	4.5	4.2	4.8	5.4	5.1	4.7	5.2	Ŋ	5.2		4.5	4.5	4.5	4.3	4
ਠ	mg/L	7	6 .	- :	2.3	2.3	2.6	2.3	7	2.2	2.4	2.8	7.	1.6	-:	4.	3.7	2.8	2.7	က	1.7	6.0	9.0	2.1	2.4	- ∞:	1.6	- 8:	2.1		2.1	1.6	. 8.	2.4	2.7
ALK	mmol/L	0.049	0.042	0.042	0.043	0.048	0.046	0.041	0.053	0.054	0.045	0.04	0.036	0.039	0.036	0.04	0.033	0.032	0.038	0.035	0.042	0.052	0.042	0.048	0.045	0.045	0.046	0.038	0.041	0.04	0.035	0.048	0.02	0.042	0.036
kond.	mS/m	2.74	2.43	2.03	2.64	2.65	2.55	2.47	2.37	2.56	2.38	2.61	2.48	2.52	5.06	2.13	2.73	2.46	2.39	2.48	2.1	1.88	1.78	2.43	2.54	2.63	2.29	2.4	2.4	2.23	2.35	2.19	2.31	2.28	2.43
Ħ	-log (H+)	5.18	5.48	5.41	5.33	5.39	5.45	5.43	5.51	5.53	5.51	5.33	5.06	5.06	2.07	5.26	5.04	5.13	5.23	5.19	5.28	5.5	5.49	5.45	5.27	5.09	5.4	5.23	5.35	5.33	5.34	5.44	5.49	5.37	5.22
Limed	•	-	-	-	-	-	_	-	-	-	-	-	-	-	-	-	-	_	-	-	-	-	-	-	-	•	-	-	-	-	-	-	-	-	-
Depth	CH	15	15	15	15	15	15	15	15	15	15	15	വ	2	S)	Ŋ	ស	വ	ß	S.	2	2	5	Ŋ	2	Ŋ	S.	Ŋ	വ	2	2	2	വ	വ	വ
Date		0-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	3-Sep-94	1-Sep-94	3-Sep-94	11-Oct-94	1-Oct-94	16-Nov-94	0-Nov-94	15-Dec-94	I-May-95	1-Jun-95	13-Jul-95	28-Sep-95	5-Oct-95	10-May-96	12-Jun-96	3-Oct-96	24-Oct-96	3-Nov-96	10-Jun-97	8-Jul-97	7-Sep-97	6-Oct-97	2-Nov-97
		28 10					_																			•	_			-	•		-	_	
Loc.		SP																																	

700	ma/L	3.4	3.4	3.3	3.4	က	က	3.3	3.2	4.1	3.9	3.5	3.2	4.6	3.4	2.9		3.6	3.3	2.8	3.3	3.3	3.5	က	13.8	11,4	11.5	F	9.5	9.4	9.6		9.5	11.8	10.3	14.2	11.9
Alo	ug/L	5	112	112	118	114	108	128	129	133	131	116	139	114	105	101	106	129	122	105	112	112	109	109	343	363	343	318	292	300	300	391	276	308	220	360	330
Ą	uq/L	170	198	200	220	251	225	270	224	212	506	217	217	199	212	239	189	218	253	202	180	191	200	236	438	451	418	423	428	398	431	499	337	403	333	464	427
×	mg/L	0.11	0.11	0.12	0.1	0.18	0.16	0.21	0.25	0.11	0.09	0.21	0.19	0.36	0.221	0.24		0.09	0.12	0.09	0.07	90.0	0.08	0.04	0.18	0.15	0.18	0.25	0.19	0.13	0.13	0	0.19	0.13	0	0.21	0.24
Na	mg/L	1.53	1.49	1.14	1.43	1.74	1.74	1.49	1.44	1.27	1.54	1.62	1.81	1.51	1.48	1.91	1.49	1.45	1.42	1.54	1.41	1.68	1.52	1.64	1.38	1.32	1.05	1.21	1.66	1.51	1.54	1.7	1.12	1.35	1.49	1.54	1.62
Mg	mg/L	0.45	0.43	0.4	0.51	0.49	0.49	0.48	0.36	0.31	0.29	0.57	0.58	0.49	0.45	0.62	0.46	0.5	0.51	0.41	0.42	0.47	0.44	0.51	0.27	0.22	0.2	0.31	0.43	0.38	4.0	0.47	0.23	0.23	0.23	0.29	0.34
రొ	mg/L	1.19	1.12	-	1.15	1.24	1.3	1.18	0.99	0.84	0.87	1.33	1.42	1.33	1.21	1.63	1.19	1.26	1.21	1.19	1.15	1.28	1.12	1.29	0.54	0.44	9.4	0.51	0.72	0.61	0.61	0.75	0.41	0.47	0.5	0.54	0.59
Tot-N	hg/L	126	135	128	205	180	146	170	141	129	160	185	240	535	180	405		117	138	104	117	182	155	345	250	260	212	215	190	185	170		202	215		282	250
N-EON	hg/L	4	ω	Ŋ	79	9/	23	77	4	4	17	29	130	440	51	282		7	8	S	CI	44	12	232	cu	-		Ø	വ	0.5	0	0.5	0	α	0.5	0.5	4
NH4-N	µg/L	က	က	က	9	က	က	က	က	က	က	က	က	က	∞	ო		က	က	ß	က	က	9	က	က	7	က	တ	9	7	က		7	7		9	7
ß	mg/L	3.7	4	3.3	3.7	3.7	3.8	3.7	5.6	2.8	2.8	4.1	4	3.8	3.9	2.8		4	4	3.8	3.8	4.2	4.1	4	4.4	4.2	3.4	4.9	4.7	3.6	4.6		7	3.1			5.2
804	mg/L	6.2	5.3	4.6	4.9	4.1	4.2	4.2	4.4	4.1	4.6	5.4	5.1	4.6	5.1	3.9		5.2	4.8	4.3	4.4	4.6	4.3	4.1	3.8	3.1	5.9	3.4	3.6	2.8	3.6		2.6	3.2		3.1	3.4
ਠ	mg/L	1.7	5.	-:	1.4	ო	2.7	5.9	1.7	6.0	0.8	2.2	2.4	1.8 8.	1.7	4.8		2.1	2.3	2.4	1.6	1 .9	2.4	5.6	4.	1.2		1.8	3.6	4.8	2.1		1.2	0.8		1.8	2.5
ALK	mmol/L	0.037	0.039	0.036	0.043	0.032	0.04	0.036	0.04	0.049	0.043	0.049	0.052	0.045	0.044	0.033		0.04	0.042	0.036	0.046	0.055	0.043	0.038									0.014	0.014			
			2.49	2.04	2.15	2.43	2.4	2.4	2.09	1.84	1.8 8.	2.53	5.6	2.62	2.33	3.17		2.37	2.29	2.33	2.18	2.4	2.27	2.46	2.83	2.84	2.54	2.8	3.51	2.88	2.79	3.53	2.03	2.31		5.69	3.03
Hd	-log (H+)	5.07	5.05	5.05	5.34	5.13	5.24	5.21	5.35	5.48	5.45	5.28	5.34	5.11	5.47	5.12	5.41	5.36	5.27	5.31	5.38	5.57	5.44	5.25	4.42	4.44	4.5	4.47	4.34	4.44	4.46	4.38	4.65	4.64	4.57	4.54	4.46
Limed	•	-			_	-	-	-	-	-	-		-	-	-	-	-	-	-	-	-	-	-	-	-	,-	-	τ	-	-	-	-	-	-	-	, -	
Depth	띩	15	15	5	5	15	15	15	5	15	15	15	15	15	12	5	15	5	ट	5	15	5	5	15	Ω.	2	Ŋ	S	2	co	ည	Ŋ	2	2	Ŋ	വ	2
Date		8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	16-Nov-94	30-Nov-94	15-Dec-94	11-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	3-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	8-Sep-94	1-Sep-94	9-Sep-94	11-Oct-94	31-Oct-94	6-Nov-94	30-Nov-94	15-Dec-94	11-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95
G			•	•			• •		-																37						`	•	_			••	
Loc.		SP	SP	S	SP	S	SP	Sp	SP	SP	SP	SP	SP	S	S D	S	SP	SP	S	S	S	S	S	SP	SP	SP	S	S	SP	SP	SP	SP	SP	SP	SP	S	SP

TOC	/om	10.8	5	6.0	12.1	80	2	96	14.6	16.6	13.6	12.9	13.9	12.6	10.6	8.5	9.1	თ	10.4	9.6	5	=	5	1.1	11.2	28.9	7	10.7		8.6	14.3	14.3	13.3	11.3
Alo	1/011	293	8	564	356	353	569	264	349	375	351	345	372	363	312	275	308	287	370	309	349	583	376	346	318	860	358	386	290	268	345	376	349	344
Alr	na/L	366	422	180	474	431	353	368	424	462	432	452	463	471	411	431	416	441	490	373	452	453	449	412	422	3290	482	458	385	436	453	452	452	459
¥	ma/L	0.2	0.08	0.18	0.08	0.08	!	90.0	0.06	0.0	90.0	0.04	0.17	0.18	0.19	0.19	0.13	0.11		0.22	0.19		0.22	0.22	0.3	0.61	0.25	0.15		0.13	0.15	0.14	0.15	0.12
Na	ma/L	1.25	65	1.88	1.43	1.35	65	1.44	13.	1.58	1.62	1.69	1.37	1.02	1.14	1.64	1.48	1.43	1.4	1.09	1.34	1.44	1.42	1.52	1.25	1.75	1.36	1.33	1.22	1.47	1.31	1.56	1.64	1.71
Mg	mg/L	0.28	0.33	0.61	0.33	0.36	0.39	0.3	0.27	0.31	0.35	0.4	0.26	0.18	0.26	0.42	0.37	0.33	0.37	0.2	0.2	0.21	0.26	0.28	0.28	0.39	0.31	0.33	0.33	0.27	0.25	0.25	0.29	0.33
రొ	mg/L	0.45	0.53	0.99	0.53	0.53	0.52	0.53	0.46	0.54	0.54	0.55	0.51	0.35	0.43	69.0	9.0	0.58	0.64	0.38	0.4	0.44	0.48	0.47	0.44	99.0	0.5	0.52	0.49	0.47	0.44	0.41	0.44	0.49
Tot-N	rg/L	520	230	200	245	195		175	250	315	280	235	225	215	195	180	175	165		170	215		225	200	202	710	220	175		170	235	285	250	220
NO3-N	µg/L	4	4	0.05	0.5	0.5		0.5	0.5	0.5	0.5	0.5	0.5	က	7	8	0.5	0.5	0.5	2	0	0.5	0.5	0.5	0.5	0.5	0.5	0.5		0.5	0.5	0.5	0.5	4
NH4-N	µg/L	5	Ξ	32	9	7		5	9	თ	우	9	က	က	12	က	က	ო		9	7		7	ស	9	83	က	7		9	10	=	유	က
S	mg/L	3.4	4.6	8.5	4.7	4.5		5.4	5.2	6.4	5.4	വ	4.5	3.5	4.9	ß	4.1	4.6		1 .	3.1		5.1	5.4	ιΩ	8.7	Ŋ	4.8		5.8	2.7	7.1	6.5	6.2
804	mg/L	3.1	3.5	5.3	3.5	3.7		2.8	2.6	3.3	3.2	3.4	3.8	2.8	2.8	3.4	2.5	3.4		2.4	5.9		2.9	5.8	3.4	4	3.3	3.5		3.3	5.6	2.8	က	3.1
ö	mg/L	-	1.2	3.4	1.6	ا 9:		2.3	د .	1.4	7	2.3	1.4	Ξ	1.6	3.6	1.7	CJ		1:1	0.8		1.6	2.2	-	2.5	1.6	1 .		01	1.3	ر ن	2.1	2.3
ALK	mmol/L	0.019						0.005	0.005	0.005	0.005									0.014	0.015	0.011		0.015	0.005					0.005	0.005	0.014	0.011	0.005
kond.			5.66	4.69	3.05	3.08		2.77	2.51	2.82	2.91	3.05	5.8	2.3	2.42	3.47	2.76	2.67	3.11	1.95	2.2	2.33	5.2	2.59	2.5	4.05	2.75	2.88		2.77	2.45	2.47	2.73	2.7
Ħ	-log (H+)	4.54	4.55	4.3	4.47	4.48	4.46	4.58	4.62	4.56	4.6	4.48	4.45	4.51	4.5	4.36	4.45	4.48	4.44	4.65	4.65	4.61	4.56	4.61	4.58	4.43	4.51	4.53	4.47	4.57	4.63	4.65	4.65	4.57
Limed	,	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-			_	-	-	-	-	-	-	-
Depth	Ę	വ	2	IJ	Ŋ	S.	Ŋ	S	Ω	വ	ស	S	15	15	12	15	15	15	15	15	15	15	15	15	15	15	15	15	15	15	5	5	15	5
Date		0-May-96	2-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	8-Sep-94	19-Sep-94	1-Oct-94	1-Oct-94	16-Nov-94	0-Nov-94	5-Dec-94	1-May-95	21-Jun-95	13-Jul-95	28-Sep-95	15-Oct-95	2-Jun-96	29-Aug-96	3-Oct-96	4-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97
0		_		N			•			•	37																					•	•	
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700	mg/L				4	<u>.</u>						28	ì			30.9	3				40.6	30.4	1				22.8		16.6	6.8	12.3	12.8	10.1	Ξ	12.9	6.8	7
Alo	hg/L			25.4	5 5	226	077			237	272	309				363	3	383	367	433	396	394	3				468	38,	271	218	309	328	287	294	314	216	157
Ą	hg/L			255	201	95. 23.	2004			257	303	321				396)	440	387	426	394	416	-				596	494	477	348	770	526	451	256	544	394	394
¥ ,	mg/L	0.5	1				ر بر	5																			0.16	0.15			1.21	0.7	ب 1.	Ξ	0.89	1.34	0.77
S.	mg/L 1 57	<u>?</u>		1 18	 	<u>.</u>	<u>.</u>		55	65	33						1.06	1.56	1.45	2.16	1.83	}					2.67		1.44	1.43	3.59	2.48	2.49	4.6	3.56	2.52	3.27
Mg	mg/L	3		0.32	300	0.00	2		0.29	0.35	0.42	0.51	0.43			0.43	0.39	0.4	0.34	0.41	0.41	0.42	<u> </u>				0.68		0.39	0.46	1.15	0.68	0.63	1.07	0.79	0.57	0.67
చ్ర :	mg/L 0.88	3	1.38	0.58	0.63	2	i		0.54	0.53	0.0	0.68	9.0			0.59	0.49	0.68	0.46	0.54	0.46	0.44	- - -				2.34	1.74	1.38	 5.	3.62	2.14	1.96	3.26	2.52	1.68	1.88
Tot-N	hg/L				300)																						630	595	380	730	200	405	430	200		
NO3-N	194 1. 6.	•	က	က	4	•						œ									N		7	ဖ			α	N	8	വ	16	8	88	∞	4		0.5
NH4-N	ng/r				5)																							80	49	166	33	က	က	9		
is s	IIIg/L				3.1	:																							4.4	3.6	4.8	4.2	4.4	4.9	4.4		
S04	- 1 - 1				4.1																								4	3.4	4.7	4.9	4.7	6.7	8.4		
تا ق) D				3.4																								1.9	3.8	16.4	4.9	5.1	12	5.5		
ALK	1																										0.046			0.027		0.021	0.022	0.023	0.029		0.016
kond.								3.55																					2.64	2.97	8.06	3.93	4.01	6.81	5.07		4.83
Hd (##)	4.12		4.39		4.1	4.1	4.15	4.26	4.23	4.04	4.09	4.07	4.3	4.01	4.05	3.98	4.04	4.18	4.12	4.06	4.04	3.98			4.6	4.46	4.97	4.99	4.95	4.88	4.56	4. 8.	4.78	4.77	4.84	4.66	4.69
Limed		-	-	-	-	-	-	-	-	-	-	-		-	-	-	-	-	-	-	-	-	-	_		-	-	-	-	-	-	-	-	-	-	-	-
Depth	2	5	ည	2	S	2	2	2	ည	ည	2	വ	ည	വ	വ	2	Ŋ	ß	ည	വ	ည	Ŋ	15	15	15	15	2	ည	S)	2	2	S.	ည	വ	S	S.	ស
Date	8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	3-Sep-94	11-Sep-94	19-Sep-94	12-Nov-97	8-Sep-94	11-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95
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Loc.				•							S																										

700	mg/L	9.6	8.8		26.9	10.0	9 0	- - - - - - - - - - - - - - - - - - -	10.3	2		9.6	9	11.55	6.1	;		7.5	7.7	8.3	2.9	6.5	8	7.4		7.2	7.7	6.7	9	8.3	!	9.4	7.9
Alo	hg/L	168	190		305	304	247	151	232	 	361	191	205	240	157		272	232	237	237	207	192	240	163		206	528	224	171	184	123	191	195
Ąľ	rig/L	510	430		340	440	75	934	756		580	271	301	364	321		469	476	530	220	423	436	630	562	!	636	662	579	646	280	255	423	521
×	mg/L			0.81	9	0.35	}							1.33	.3	1.07	0.61	5		0.75	0.51	0.27			0.65	0.12	0.0	0.13	0.18	0.04		0.5	0.35
Na	mg/L	5.89			1 79	2.28	996	6.33	5.8	3.4		2.52	2.56	2.16	1.85	3.01	2.55	2.96	4.11	3.77	2.98	3.43	5.01	4.85		3.99	4	3.95	5.98	5.14	2.9	2.75	3.3
Mg	mg/L	1.12	0.73		0.48	0.56	0.77	1.26	1.17	0.45	0.89	0.47	0.44	0.47	0.45	1.03	0.72	0.67	0.78	0.76	0.51	0.53	0.88	0.85		0.67	0.63	99.0	0.87	99.0	0.31	0.37	0.5
రొ	mg/L	2.99	1.79		1.44	1.58	6	3.24	2.99	1.06	2.12	1.46	1.38	1.4	1.2	2.93	2.07	1.91	2.15	2.21	1.3	1.38	2.02	1.95		1.53	1.37	1.32	2.15	1.53	0.7	0.81	0.93
Tot-N	µg/L	785	330		405	355						330	330	440	215			295	280	280			385	260		235	235	230	250	275		310	245
N-EON	µg/L	202	Ξ		0.5	0.5		ις	8			-	,	-	0.5			67	က	က	4	0.5	15	4		0.5	0.5	0.5	4	0.5		-	4
NH4-N	hg/L	158			8	თ						က	က	က	∞			က	5	ß			8	12		12	12	5	က	7		13	ო
ឆ	mg/L	5.8				4.3						4.4	4.6	4.9	3.6			4.4	4.6	4.5			5.4	4.5		4	4.4	4.7	3.7	3.2		3.3	2.7
804	mg/L	3.4				2.7						7.1	9.7	9	4.4			5.4	6.8	6.8			8.6	9.7		6.9	7.2	6.9	6.5	7.2		5.5	4.1
ច	mg/L	18.7				3.8						3.9	3.8	3.4	4.8			6.1	8.3	8.1			9.5	9.5		7	6.8	7.1	12.7	8.5		3.7	6.5
ALK	mmol/L		0.019			0.024		0.005				0.028	0.026	0.026	0.022			0.013	0.025	0.023		0.05	0.026	0.017		0.012	0.014	0.018	0.012	0.014		0.028	0.012
kond.	E/SE	8.4	4.01		3.14	3.7		8.54	8.22					3.48	3.14			4.63	5.41	5.34		4.18	6.55	5.5		4.98	5.09	4.61	7.01	5.71		3.38	3.96
Ħ.	-log (H+)	4.66	4./3		4.81	4.79	4.7	4.55	4.52	4.76	4.59	4.91	4.86	4.83	4.82	4.59	4.7	4.66	4.81	4.77	4.69	4.73	5.01	4.67		4.59	4.66	4.71	4.66	4.7	4.85	4.84	4.66
Limed	•	- ,	_		_	-	-	-	-	-	-	-	-	-	-	-	_	-	-	-	-	_	-	-	-	-	-		_	-	-	-	-
Depth	Ę 1	υr	n i	S.	2	22	Ŋ	വ	5	വ	2	15	15	15	15	15	15	15	15	15	5	15	5	15	15	15	5	5	15	5	ភ	5	र्घ
Date		U-May-96	08-11nc-21	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	16-Oct-97	12-Nov-97	8-Sep-94	1-Sep-94	19-Sep-94	11-Oct-94	31-Oct-94	16-Nov-94	0-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	13-Nov-96	10-Jun-97	8-Jul-97	17-Sep-97	16-Oct-97	2-Nov-97
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Loc.		y 0	, ,	J.	S D	SP	S	S	SP	S.	S D	g B	ر م												S	SP	S	S D	S D	SP.	S	S	S P

T0C	mg/L																8	2											8	?				7.4	6	17.3
Alo	hg/L				777	7					307	ŝ		480	2		330	33,00	3				8	}					143	2 5	ì	352	3	149	146	262
Alr	hg/L				16.1	† †					670	5		980	3		542	20.5	3				184						366	228	} !	624	- 1 3	278	320	352
¥	mg/L																								0.21	!										0.25
Na	mg/L	5.3		2 68	2 6	1	2 52	9	2 61	i				1.94		2.5	4 28	ì	208	ì			000	ì	2.77					1.83	!	3.73	;	2.71		1.75
Mg	mg/L	5.5		104	2.0	5	0.78	3	0.7	;	1.01	33		0.66	0.87	90	0.84	0.94	0.51)			0.64		0.62				0.86	0.47	0.54	1.03		0.46	0.5	0.34
යි [.]	mg/L	5 5	:	2.41	187	<u> </u>	1,66	3	1 71	• •	2.34	2.4	i	1.24	1 62	12	1.16	1.37	1.27	1.24	į		1.38		1.28				1.76	1.04	1.12	1.62		0.93	0.95	1.16
Tot-N	hg/L																																			295
NO3-N	hg/L	٠ م	1				9	?											-	۰ ۵					4				1185							0.5
N-4-N	hg/L																																			ო
io s	IIIg/L																																			4.8
S04	1.g/L																																			4.9
٥ -	-19 -1																																			1.8
ALK																																				0.023
kond.																																				2.84
Hd (#H)	4.59	4.68	4.69	4.45	4.82	4.58		4.6	4.48	4.43	4.53	4.53	4.5	4.46	4.37	4.64	4.56	4.47	4.68		4.78	4.61	4.62	4.7		4.72	4.59	4.54	4.54	4.65	4.58	4.51	5.89	4.74	4.64	4.77
Limed		_	-	-	-	_	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	_	-
Depth L	2	Ŋ	IJ	Ŋ	Ŋ	Ŋ	Ŋ	Ŋ	Ŋ	Ŋ	വ	വ	വ	ß	Ŋ	Ŋ	Ŋ	Ŋ	15	15	5	15	15	5	15	15	12	15	15	15	15	15	15	15	15	15
Date [8-Sep-94	11-Sep-94	19-Sep-94	31-Oct-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95	10-May-96	12-Jun-96	29-Aug-96	3-Oct-96	24-Oct-96	7-Aug-97	16-Oct-97	12-Nov-97	8-Sep-94	11-Sep-94	19-Sep-94	31-Oct-94	16-Nov-94	30-Nov-94	11-May-95	21-Jun-95	28-Sep-95	15-Oct-95	12-May-96	3-Oct-96	24-Oct-96	8-Jul-97	17-Sep-97	16-Oct-97	12-Nov-97	11-Sep-94
ė			•								50 1					20				•					•					20					50	•
Loc.	SP	SP	SP	SP	SP	SP	SP	SP	SP	S	SP	SP	SP	S	SP	SP	SP	SP																		

Appendix B. Hydrology

Abbreviation and references for sampling stations:

LIM-1: Stream outlet of catchment 1 in the limed catchment Fugleliåsen

LIM-4: Main stream outlet of the limed catchment Fugleliåsen

REF: Stream outlet of reference catchment Spjøtåsen

	LIM-1	LIM-4	REF		LIM-1	LIM-4	REF
DATO	m3/period	m3/period		DATO		m3/period	
1-Mar-94		•	•	12-Jan-96			907
13-Mar-94	42409	73455	35718	18-Apr-96			27005
29-Mar-94	12056	20882	14520	22-Apr-96			13419
11-Apr-94	90137	156122	60915	28-Apr-96	16009	29703	11203
18-Apr-94	27255	47206	23256	7-May-96		20578	7690
25-Apr-94	28191	48829	20565	20-May-96			8553
2-May-94	38872	67329	28031	29-May-96	7424	18413	6707
9-May-94	14053	24482	12153	20-Jun-96			3152
18-May-94	5632	9755	3450	7-Sep-96			5167
25-Aug-94	12747	30574	9079	30-Sep-96		51575	21310
31-Aug-94	22617	57695	16508	16-Oct-96		26823	10589
11-Sep-94	29915	59405	22739	28-Oct-96		59484	22306
19-Sep-94 29-Sep-94	30121 3188	58337 5960	22593 372	7-Nov-96		37884	15207
12-Oct-94	1799	2428	1338	21-Nov-96	8094	19409	7497
26-Oct-94	8774	17119	15401	4-Dec-96	9081 19230	28174	6213
31-Oct-94	22534	1/113	7212	8-Dec-96 31-Dec-96	7679	29127	17316 7382
3-Nov-94	11524	69380	8300	31-Dec-30 3-Mar-97	7079	18918 104865	41315
14-Nov-94	5331	10642	4321	5-May-97 5-May-97		104803	36057
21-Nov-94	9675	18438	7314	23-Jun-97		32745	9340
6-Dec-94	6146	11686	4907	2-Jul-97		17242	5711
12-Dec-94	20238	38509	14988	7-Aug-97		5320	3937
21-Dec-94	14412	27807	11131	4-Sep-97		9252	5668
31-Dec-94	6536	13172	5316	18-Sep-97		19720	8800
3-Jan-95	1122	2449	1095	14-Oct-97		35028	14427
15-Feb-95	53026	115920	63605	4-Nov-97		18359	7886
14-Mar-95	31353	65867	30239	10-Nov-97		29766	9648
27-Mar-95	25161	48771	25208	13-Nov-97		39454	14800
5-Apr-95	42085	29331	38208	31-Dec-97		163590	61217
16-Apr-95	39339	68756	23288				
25-Apr-95	30706	53875	15301				
8-May-95	23320	45086	13549				
26-May-95	9291	19022	7794				
6-Jun-95	11336	18702	7054				
13-Jun-95	16064	24167	10057	•			
27-Jun-95	22051	31656	13272				
7-Sep-95	1069	1663	1052				
14-Sep-95	17515	29530	11848				
18-Sep-95	27845	54684	22350				
4-Oct-95 23-Oct-95	13389 16168	26732	10949				
20-Nov-95	4779	36374	13771				
4-Dec-95	4779 2715	11416 6821	3737 1859				
19-Dec-95	4400	10787	2907				
31-Dec-95	858	2307	2907 975				
01-060-00	000	2007	9/3				

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