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Xiaohang Xu, Yan Lin, Bo Meng, Xinbin Feng, Zhidong Xu, Yuping Jiang, Weilu Zhong, Yanhai Hu, Guangle Qiu. The impact of an abandoned mercury mine on the environment in the Xiushan region, Chongqing, southwestern China. Applied Geochemistry. Volume 88, Part B, 2018, pages 267-275, ISSN 0883-2927.

The article has been published in final form by Elsevier at http://dx.doi.org/10.1016/j.apgeochem.2017.04.005

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1	The impact of an abandoned mercury mine on the environment in
2	the Xiushan region, Chongqing, southwestern China
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26 Abstract

Mercury (Hg) contamination is a serious problem at Hg mining area of Xiushan 27 28 County, Chongqing, southwestern China. Concentrations of total mercury (THg) and methylmercury (MeHg) in paddy soil, surface water, and rice (grain) samples 29 wereanalyzed to exhibit regional distribution of Hg contamination. Simultaneously 30 31 gaseous elemental Hg (GEM) in ambient air in the vicinity of sampling sites were 32 measured in situ. Hg concentrations in surface water were highly elevated ranging 33 from 13-2390 ng/L, and MeHg concentrations varied between0.17-1.1 ng/L. Dissolved Hg and MeHg concentrations in surface water ranged from 4.7-470 ng/L 34 and 0.14-0.35 ng/L, respectively. High THg and MeHg concentrations were also 35 obtained in paddy soils from mining areas, ranging from 0.45-68 μ g/g and 0.13-4.8 36 ng/g, respectively. Similar to high concentrations in water and soil, Hg in rice (grain) 37 ranged from 4.7-550 ng/g and MeHg from 2.9-26 ng/g. Elevated Hg concentrations in 38 rice, as a staple food to residents, confirmed that rice consumption could be a vital 39 pathway of MeHg exposure to native populations. The spatial distribution 40 41 characteristics of Hg and MeHg pollution in the local environment indicated the origin from historic Hg mining sites in the Xiushan area. 42 43 44 Keywords: Mercury; Methylmercury; Paddy soil; Surface water; Rice; Hg mining 45 46

48 **1. Introduction**

Mercury (Hg), especially methylmercury (MeHg), is highly toxic which has 49 50 raised widely environmental concerns (Clarkson, 1998; UNEP, 2002). Hg can exist in the atmosphere as gaseous elemental Hg (GEM), and can be migrated for a long 51 distances due to its long lifetime (0.5-2year) in the atmosphere (Schroeder et al., 52 53 1998). Both inorganic and organic forms exist in the environment, and inorganic mercury (IHg) can be transformed into MeHg, the most toxic Hg form. The health 54 55 risks posed by MeHg due to its bio-magnification in the aquatic ecosystem are well known. Trace amounts of MeHg in water can lead to much higher and harmful 56 concentrations of MeHg in fish and their predators. The consumption of fish has been 57 considered as the major pathway for human MeHg exposure (WHO, 1990). 58 Mercury contamination is a serious problem in China, especially in historical Hg 59 mining regions in China (Horvat et al., 2003; Qiu et al., 2005; 2006a, b; 2008; 2012; 60 Feng et al., 2008). The major concern about Hg is its bio-accumulation, the 61 transformation from IHg to MeHg. Such strongly toxic Hg substances are readily 62 63 migrated to environmental compartments of water, soil and sediments. Generally, retorted ores (termed calcines) originated by low efficiency roasting of cinnabars in 64 Hg mining areas contain plenty of water soluble Hg compounds, such as the elemental 65 66 Hg and Hg salts (Kim et al., 2000; 2004). Under certain conditions, those soluble Hg compounds can be transformed into MeHg and finally flow into the human body 67 through the food chain (Watras et al., 1992; Weber, 1993). Though localized 68 69 ecological degeneration and health exposure risk from these contaminated sites, their combined impacts arise more widespread concern. Usually, the local residents of the 70 71 mining districts are exposed to long term health problems (Bose-O'Reilly et al., 2010). 72

The Xiushan Hg mine, namely Yangshikeng (YSK), is situated in the
southeastern part of Chongqing, southwestern China. The Hg mining activity of stateowned company at the YSK started at 1960s with a capacity of 25 tons metallic Hg
per year. Currently a private company of Yameiou holders the mining permit of the
YSK and its yearly output of Hg is approximately 30 tons. During sampling

campaigns a few of illegal artisanal Hg mining sites in the region were also observed.Rice was the staple food for native residents.

80 Currently, rice was proved to have strong enrichment ability of MeHg and rice consumption was confirmed to be the major MeHg exposure pathway for local 81 residents in Hg mining area (Feng et al., 2007; Qiu et al., 2008; Li et al., 2010). 82 However, up to now, only some THg mercury data was reported (Li, 2014; Zhang et 83 al., 2011). Due to the smaller scale than Wanshan, and Xunyang, few attention has 84 85 been paid in this area. There is still lack of research conducted to reveal the MeHg contamination in environmental media for further human health risk assessment in 86 87 Xiushan Hg mining area.

In this study, we analyzed THg and MeHg concentrations in water, rice and paddy 88 soil. Gaseous elemental mercury (GEM) in the ambient air were also measured in situ. 89 90 The purposes of this study are to evaluate the Hg pollution status in Xiushan by examining (1) GEM in the atmosphere; (2) Hg speciation in surface water as well as 91 92 THg and MeHg in rice and soil; and (3) potential environmental risks from MeHg 93 containing rice to local residents. The research is latest one in a range of researches related to abandoned Hg mining area in China in our research team, including several 94 published studies of the Wanshan (Qiu et al., 2005), Wuchuan (Qiu et al., 2006a), 95 96 Lanmuchang (Qiu et al., 2006b), Tongren (Li et al., 2008a), Xunyang (Qiu et al., 97 2011, 2012), and Yanwuping Hg mine (Qiu et al., 2013).

98 2. Study area

The Yangshikeng (YSK) Hg mining district (E: 108°53'2"-108°54'1"; N: 99 28°34'5"-28°36'4"), is located 32 km northwestern of the Xiushan County, covering 100 approximately 2.5 km² (Figure 1). The district is in a karstic and hilly region with 101 elevation ranging from 1631 to 246 m. It possesses a subtropical humid climate 102 103 featured by mild temperature and rich precipitation. The annual mean temperature is 16 °C. Average annual rainfall is about 1341 mm. Cinnabar is the main mineral in the 104 YSK is with grade ranging from 0.13% to 0.28%. The Shileixi creek is an affluent of 105 106 the Rongxi River, which is directly affected by the Hg mining activity in history and runs through the YSK area. And there exist 18.3 thousand residents living in the 107

108 downstream of YSK area, most of them are farmers who depend on the yield from the

109 rice paddy fields. Most of mine-wastes were deposited along the Shileixi Creek

110 without control. Significant quantities of mine-waste calcines were heaped around

abandoned mining and retorts processing sites.

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113 **3. Methods**

114 **3.1. Samplings and preparation**

115 The sampling of soil, water, and rice was carried out in September 2011. During the sampling time, it was clear and the river runoff was at base flow condition. Water 116 samples (both unfiltered and filtered surface water, n=11) were collected from the 117 Shileixi creek river for Hg and MeHg analysis. Among samples, X15 is a sample of 118 leachate from abandoned mine-waste tailings of the YSK. Unfiltered samples were 119 directly gathered from the river for THg and total MeHg (TMeHg). Meanwhile, ultra 120 cleaned syringes and membrane filters (0.45 µm) were used for filtered samples for 121 Dissolved THg (DHg) and Dissolved MeHg (DMeHg). Ultra-pure HCl were added 122 123 into water samples on-site, obtaining an acid concentration of 0.4% (v/v) according to Method 1631E (USEPA, 2001). All the water samples were collected and preserved in 124 borosilicate glass bottles pre-cleaned by heating in a muffle furnace at 550° C for 3 h. 125 Soil samples (n=15) were gathered from paddy fields within or nearby the 126 Shileixi creek and Rongxi river (Figure 1). Each soil sample is a composite sample of 127 3-5 sub-samples from an area of 1 m^2 in diagonal sampling technique. Rice paddy soil 128 samples for all geochemical analysis were sealed in plastic bags to prevent cross 129 contamination. In the laboratory, all samples were kept at -20 °C in a refrigerator, then 130 131 appropriate amounts of samples were freeze-dried, homogenized in an agate mortar, 132 and passed through minus 200-mesh sieve.

Rice grain samples (n=15) were collected at the same time from to the same locations of soil samples. All the rice samples were sealed and stored in the same way as soil samples to prevent cross contaminations. Then the rice samples were freezedried, and ground with grinder (IKA-A11), and then sieved through 200 meshes, and stored in plastic bags for analysis.

Ambient air Hg (GEM, n=22) concentrations were measured in situ within the vicinity of the soil sampling locations. The Lumex RA-915⁺ (Lumex Ltd., Russia) was employed to measure GEM, and the detection limit for atmospheric Hg is 0.5 ng/m³ at the rate of 10 L/min. For the machine, after 5 min for obtaining stable baseline stage, an observation lasts for 30-60 min was conducted at each site.

143 **3.2. Analytical methods**

For rice THg analysis, approximately 0.5-1.0 g of rice sample was dissolved at 95-140 °C with a mixture of HNO₃ and H₂SO₄ (4:1, v/v) (Qiu et al., 2012; 2013). Rice samples were measured using cold vapor atomic fluorescence spectroscopy (CVAFS) in accordance with Method 1631E (USEPA, 2002). For rice MeHg determination, approximately 0.2-0.5 g rice sample was weighed for digestion using 25% KOHmethanol at 75-80 °C for 3 h. Then MeHg in rice samples was leached with dichloromethane (CH₂Cl₂) and back-extracted into water phase for determination

151 based on Method 1630 (USEPA, 2001; Liang et al., 1996).

For soil THg measurement, about 0.1-0.3 g of sample was heated at 95 °C for 30 152 153 min using water bath in a mixed solution of HCl and HNO₃ (3:1, v/v), then BrCl was added and the mixture heated to 95 °C for another 30 min (Feng et al., 2013; Meng et 154 al., 2016). Then an appropriate amount of digestive soil solution was determined 155 using cold-vapor atomic absorption spectrometry (CVAAS) (Feng et al., 2007; Qiu et 156 al., 2008). For soil MeHg determination, about 0.2-0.5 g soil sample was extracted 157 with CH₂Cl₂ after HNO₃ leaching (Liang et al., 1996). Then a proper amount of 158 digestive solution was used for determination by GC-CVAFS according to Method 159 1630 (USEPA, 2001). 160

For water THg and DHg, water samples were determined by BrCl oxidation and
SnCl₂ reduction (Bloom et al., 1983). Then appropriate amount of solution was
measured by CVAAS. For TMeHg and DMeHg, water samples were distilled, NaBEt4
ethylated, and analyzed by cold vapor atomic florescence spectrometry (CVAFS)
(Horvat et al., 1993; Liang et al., 1994, 1996). Specifically, a 45 mL aliquot sample
was added in a Teflon distillation vessel on an Al heating pan and distilled at 170 °C

167 for 3-4 h. The distillate then proceeded aqueous phase ethylation by adding 0.2 mL 2

M sodium acetate and 0.06 mL 1% sodium tetraethylborate step by step, then purging 168

- with N₂ to absorb MeHg onto a Tenax trap (Guo et al., 2008). MeHg was ultimately 169
- 170 desorbed with heating onto an isothermal GC column for peak separation and
- analyzed by CVAFS (Brooks Rand Model III). 171

The pH of soil was measured by taking 10 g from samples into 25 ml of reagent 172 water (Chinese National Standard Agency, 1988). For soil organic matter, the 173 K₂Cr₂O₇/ H₂SO₄ method was used to digest soil samples and then determined by 174 175 spectrophotometer according to the national environmental protection standard of China. For Humic acid, Fulvic acid, and Humin, Na₄P₂O₇, NaOH, K₂Cr₂O₇, and H₂SO₄ 176 were used according to the method recommended by the Ministry of Agriculture of 177 China (2010).

3.3 QA/QC 179

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QA and QC of THg and MeHg analysis were conducted using duplicates, method 180 blanks, matrix spikes, and certified reference materials (GBW07405; BCR580, TORT-181 2). 182

183 An average total Hg concentration of $0.30\pm0.01 \,\mu$ g/g (n=5) for the soil standard (GBW07405) was obtained, which was in accordance with certified value of 184 $0.29\pm0.04 \mu g/g$. For soil MeHg, BCR 580 was determined and the achieved MeHg 185 concentration of 74 \pm 5 ng/g (n=5) confirm the certified value of 75 \pm 4 ng/g. In soil 186 duplicates, the relative percent difference was less than 10% for THg and 8.4% for 187 MeHg. The detection limits for THg and MeHg were 0.01 ng/g and 0.002 ng/g, 188 189 respectively.

In rice, recovery rates on matrix spikes were 91-109% for Hg and 87-105% for 190 191 MeHg. TORT-2 was determined, and the obtained value (150±8 ng/g, n=5) met the 192 certified value (152±13ng/g). In rice duplicates, the relative percentage difference was less than 7% for Hg and 8.6% for MeHg. The detectable limits for Hg and MeHg 193 were 0.015 ng/g and 0.004 ng/g, respectively. 194

In water, recovery rate of matrix spikes were 92-109% for THg and 89-112% for 195 MeHg. The determination limits of THg and MeHg were 0.02 ng/L and 0.009 ng/L, 196 respectively. The relative percentage difference was <9 % for THg and <8.8% for 197

198 MeHg in water duplicates. The method blank cannot be detected. The field blank was

199 0.15 ± 0.04 ng/L for THg and 0.014 ng/L for MeHg, respectively.

For soil organic matter, the $K_2Cr_2O_7/H_2SO_4$ method was used to digest soil

samples and then determined by spectrophotometer according to the national

202 environmental protection standard of China. For Humic acid, Fulvic acid, and Humin,

- 203 Na₄P₂O₇, NaOH, K₂Cr₂O₇, and H₂SO₄ were used according to the method recommended
- by the Ministry of Agriculture of China (2010)
- 205 **4. Results and discussion**

206 **4.1. Water**

The THg concentrations of unfiltered surface waters from the YSK mining region 207 ranged from 13 to 1004 ng/L and averaged 180±310 ng/L. However, the leachate from 208 nearby abandoned large mine-waste tailings (calcines), sample Y13, exhibited 209 extremely high concentration of 2390 ng/L (Table 1). The THg concentrations 210 decreased to 150 ng/L at 4 km downstream of the tailing. The highest Hg 211 concentration obtained in leachate draining from the huge calcines pile is as high as 212 213 2390 ng/L. The DHg concentrations of filtered waters altered from 4.7 to 470 ng/L with the average of 68 ng/L. Similarly, the DHg concentrations were also variable and 214 the highest value was also found in sample Y13. The observations of high THg and 215 DHg in leachate are in accordance with previous studies on other Hg mining regions 216 in China (Qiu et al., 2005; 2006a, b; 2009; 2012). The source of Hg released into the 217 creek is mainly attributed to the waste calcines, the waste calcines contains large pool 218 219 of labile and water-soluble Hg which is main contributor for DHg concentration in the 220 creek (Gray et al., 2002, 2003; Kim et al., 2000, 2004).

The concentration of THg in filtered waters and unfiltered waters showed a sharply decrease with the downstream distances increase (**Figure 2**). At the upstreams of the rivers, the THg concentrations were quite low compared to the value of 50 ng/L (Type I, water for headwater and natural reserves, Chinese EPA, 2002) set by Chinese Hg standard for surface water. However, two samples of Y13 and Y14 exceeded the 1000 ng/L Chinese drinking water Hg guideline (GB5749-2006) and the 770 ng/L U.S. Environmental Protection Agency (USEPA, 1992) guideline recommended to protect against chronic effects to aquatic wildlife. The high level of THg in watercould therefore pose risks for both human and animals depending on the water as their

drinking sources.

The TMeHg and DMeHg of were relatively similar, ranging from 0.17 to 0.55 231 ng/L, and from 0.14 to 0.35 ng/L, with the mean concentrations of 0.42 ng/L and 0.23 232 233 ng/L, respectively. Surprisingly, the highest TMeHg of 1.1ng/L was obtained in sample Y15, which is not the nearest sample in the downstream of YSK, 234 235 demonstrating there may exist favorable conditions for Hg methylation. Meanwhile, the DMeHg rate (DMeHg/TMeHg) varied from 0.16 to 0.79 with the mean value of 236 0.63, and the lowest DMeHg rate was 0.16 obtained in Y15, suggesting the most of 237 MeHg in Y15 may bond to the particles (Qiu et al., 2011). 238

The THg concentrations of surface waters, gathered from the YSK Hg mining region, were comparable to those obtained in Hg mining regions worldwide (Table 2). Waters, influenced by calcines in Coast Range Hg mines, California, and in Palawan Quicksilver Hg mine, Philippines, exhibited the concentrations of 45000 ng/L and 31000 ng/L in Hg, respectively (Rytuba, 2000; Gray et al., 2003). The highest concentration of THg in surface water was reported to reach 23500 ng/L in China Hg mining regions (Zhang et al., 2009). Horvat et al. (2003) and Qiu et al. (2005) also reported that THg in the surface water ranged from 3.3 to 10580 ng/L, with highest concentrations emerged near the mine-waste tailings in the Wanshan Hg mining region. Similarly, the creek waters influenced by the leachate through calcines demonstrated high levels of Hg, which confirm that the mine-waste calcines is the mostly substantial Hg releasing source. The elevated Hg concentrations in waters, especially the two sites (site Y13 and Y14) beyond the Chinese drinking water Hg guideline (1000 ng/L) and chronic effects to wildlife in aquatic systems (770 ng/L) issued by USEPA (1992), which has adverse impacts on the growth and development of aquatic animal.

Moreover, in this study, the TMeHg concentrations is similar to that of water samples from Hg mine in Alaska, Idrija, Palwan, Yanwuping, Xunyang (**Table 2**). However, the MeHg concentrations are less than the results obtained in California (47 ng/L),,

Wanshan (25 ng/L), Almadén (30 ng/L) (**Table 2**), this indicated that the local conditions in Xiushan is less favorable for Hg methylation (Rytuba 2000; Gray et al., 2004; Qiu et al., 2005). Meanwhile, the elevated MeHg may also result in Hg accumulation in wildlife and the aquatic system, which will pose a serious threat to human health (Wan et al., 2004).

4.2. Soil

In the present study, THg concentrations in paddy soils varied from 0.45 to 68 μ g/g with a mean of 9.8±17.5 μ g/g (**Table 3**). The sample Y11, affected by minewastes discharged from an abandoned artisanal retort along the bank of Shileixi creek, manifested the highest concentration (**Figure 3**). Elevated THg were also obtained in the samples from the sites adjacent to abandoned Hg mines, such as Y10 and Y17, containing 13 μ g/g and 15 μ g/g, respectively.

Paddy soils from the YSK mining district also showed high MeHg concentrations, ranging from 0.21-3.7 ng/g with the mean value of 1.36 ng/g. In the upstream, soil MeHg concentrations were generally low (<1.1 ng/g). MeHg concentrations were higher in the samples which were closer in distance to the abandoned mine tailings, and the peak value (3.7 ng/g) was observed in artisanal Hg mines (Y11). In the downstream, soil MeHg decreased with the distances away from artisanal Hg mines.

In this study, soil THg exhibited a typical distribution that downward trends appeared with the distance farther away from artisanal Hg mines both in upstream and downstream. The surface water runs through the mining sites, which were seriously polluted by Hg (Qiu et al., 2012; 2013), rice paddies are therefore affected by the incoming water when paddy fields are drowned during the rice growing season. Water from upstream will bring down Hg-enriched particles transported from upstreamcontaminated sites during flooded seasons (Horvat et al., 2003). Soil MeHg had the similar distribution tendency to soil THg. Hg contaminated mine-water runoffs were frequently applied to irrigate paddy fields near Hg mining and retorting sites and periodic irrigation in the paddy fields can produce anaerobic conditions in flooded soil

(Gilmour and Henry, 1991; Weber, 1993; Horvat et al., 2003). Under the appropriate pH, organic matter, anaerobic condition, sulfate reducing bacteria and other Hg methylation bacteria will play a key role to convert Hg into MeHg (Qiu et al., 2008; 2013; Rothenberg et al., 2014)

Mineral residue and calcines in the YSK Hg mining area are the major sources of Hg pollution for the local environment. For comparisons, concentrations of THg and MeHg in soils from other Hg mining areas are present in **Table 3**. In comparison with the results in this study, at Almadén, Yanwuping, Idrija, and in California, Alaska, Palawan, Wanshan, Xunyang, higher THg in soils were obtained. Similar to the results for high THg, MeHg concentrations of 7.3 ng/g, 11 ng/g, 23 ng/g, 41 ng/g, and 80 ng/g in soil from the Yangwuping, Xunyang, Wanshan, Alaska, and Idrija mining area were higher than the highest concentration obtained in this study. Though THg and MeHg concentratios in soil samples from the YSK mining area (up to 68 μ g/g THg, and 3.7 ng/g MeHg) are less than those obtained from worldwide Hg mines, soils in the YSK mining district are still heavily polluted with Hg compared to 1.0 μ g/g regulated by Chinese standards quality of soil (GB15618, 2008). The remarkable concentration present in the surface soils are an obvious index of the Hg mining activities.

4.3 GEM

The GEM concentrations varied from 29 to 4209 ng/m³, with an average of 498 ± 682 ng/m³. The spatial distribution was featured by the remarkable variation and some extremely elevated concentration (**Figure 4**). The highest concentration was obtained at the downwind direction site (Y11) of the cinnabar ores retorting facility, indicating that the retorting activity and its nearby huge pile of mine-waste tailings may be an origin of Hg⁰ in the ambient air.

In the current study, the GEM concentrations were remarkably high by comparison with the results reported in urban cities and remote regions in China (Fu et al., 2012). Compared to other historic Hg mining regions in China, which were in the range of 7.4-410 ng/m³, and 12-180 ng/m³, in Xunyang, and Wanshan (Qiu et al.,

2012, 2013), the average GEM value ($498 \pm 682 \text{ ng/m}^3$) achieved in this study was significantly elevated,. The average value of GEM in the YSK Hg mining area is about ten folds of the air quality reference standard for type II (50 ng/m^3) issued by the Ministry of Environment Protection of China (MEP), posing a potential risk to local residents.

4.4. Rice

Results of rice samples and corresponding locations were shown in **Table 2** and **Figure 1**. Samples gathered within and near Hg mining sites contained elevated concentration of THg, varying from 12 ng/g to 384 ng/g, with the mean value of 47.5±97.3 ng/g. MeHg concentration in rice are also high, varying from 9.2 ng/g to 64 ng/g with the mean value of 12.1±15 ng/g. The peak concentrations of THg and MeHg were 384 ng/g, 64 ng/g collected from abandoned artisanal Hg mines. The rice collected far away from Hg mining sites, exhibited generally low Hg, ranging from 4.7-18 ng/g.

There exists remarkable positive relationships between rice MeHg (MeHg_R), and soil THg (r=0.975, p<0.01) (**Table 4**). Significant positive correlations can also be observed between GEM and soil THg (r=0.995, p<0.01), which implying that soil THg may be a source to GEM, and the freshly Hg deposition in soil is readily to be methylated (Meng et al., 2010). The positive relationship (r=0.999, p<0.01) between rice THg and soil THg indicates that soil THg can contribute a lot to rice THg (Yin et al., 2013).

In comparison with rice samples from unpolluted region, which mean Hg of 3.3 ± 1.3 ng/g and MeHg of 2.9 ± 1.0 ng/g (Meng et al., 2010), samples gathered from the YSK Hg mining district contained high concentrations of both THg and MeHg. For rice samples in the present study, peak concentration of 384 ng/g and 120 ng/g in THg were 19 times and 6 times of the Hg limit (20 ng/g) suggested by the National Food Safety Standard (GB 2672, 2012). Meanwhile, there also existed other sites downstream from Hg mines are higher than the limit. This phenomenon displayed that rice plants in the area were seriously polluted with Hg, posing a potential risk and harm of Hg to the local residents.

The probable daily intake (PDI) values of IHg and MeHg for general rice consumption adult were estimated. It was supposed that regular amount of rice intake was 400 g per day, meanwhile body weight for one person was 60 kg. The evaluated PDI ranged widely from 0.031 to 2.56 mg/kg (mean: 0.32 mg/kg) for THg and from 0.061 to 0.427 mg/kg (average: 0.081 mg/kg) for MeHg. In our search, this large range demonstrates the wide range of Hg concentrations obtained in rice. The PDI value of 0.427 mg/kg for MeHg might suggest that rice consumption is an important approach for MeHg entering into human body, which poses a risk to human health. PDI values of IHg at abandoned Hg mine sites were 0.8 and 2.56 mg/kg per day, which were higher than the PTWI for IHg (0.57 mg/kg/d) issued by JECFA (2010). For MeHg, the highest PDI value of 0.427 mg/kg is more elevated than the new dietary reference dose (RfD) of 0.23 mg/kg body weight per day certified by the UN Committee (WHO, 2003), however the second PDI value of 0.14 mg/kg is slightly higher than the RfD of 0.1 mg/kg body weight per day recommended by the USEPA (1997).

4.5. Correlation between MeHg and organic matter

Notable positive correlations (r=0.788, p<0.01) was found between soil MeHg and humic acid (**Table 5**). Meanwhile, significant positive relationships (r=0.605, p<0.05) can be observed between soil MeHg and Fulvic acid. These positive relationships implied that humic acid may contribute a lot to affect the MeHg concentration in soil, and fulvic acid can also influence the MeHg in soil in some degree (Hintelmann et al., 1995, 1997). MeHg in rice tissue exhibited a decreasing trend, when increase the humic acid in hydroponic experiment (Su et al., 2016). The relationship in this study could be an explanation for the low MeHg in rice tissues, and humic acid may be bond to the MeHg and thus decrease the mobility of MeHg to the aboveground of rice plant (Ullrich et al., 2001).

5. Conclusions

Elevated concentrations of THg and MeHg in both filtered and unfiltered surface

water can be primarily attributed to the calcines from retorting ores. High concentrations of THg and MeHg in paddy soils in the vicinity of the abandoned artisanal Hg mines were observed. GEM levels in the atmosphere around demonstrated similar distribution pattern to the soil Hg concentration. Rice MeHg had a positive correlation with soil MeHg, indicating soil MeHg may be the major source of MeHg in rice. The significantly positive correlations between soil MeHg and humic acid indicated its key role in controlling the MeHg production in soil. Results of this study point out that the artisanal Hg mines and abandoned calcines tailings caused seriously Hg pollution in Xiushan County, and the concentration levels of Hg in both water and rice exceed the safe level and therefore pose potential health risk for local residents.

Acknowledgments

This study was supported by the National Key Basic Research Program of China (2013CB430004), the Sino-Norwegian Cooperative Project (SINOMER III) to XF and GQ, the Sino-Swedish Cooperation Project (SMaReF), the External Cooperation Program of BIC, Chinese Academy of Sciences, Grant No.132852KYSB20130003, and by the National Natural Science Foundation of China (41173126, 41373135, and 41203091). The authors gratefully acknowledge the support of all persons involved in the project.

References

- Bloom, N. S., & Crecelius, E. A. (1983). Determination of mercury in seawater at sub-nanogram per liter levels. Marine Chemistry, 14(1), 49-59.
- Bose-O'Reilly, S., Drasch, G., Beinhoff, C., Rodrigues-Filho, S., Roider, G., Lettmeier, B., ... & Siebert, U. (2010). Health assessment of artisanal gold miners in Indonesia. Science of the total Environment, 408(4), 713-725.
- Chinese National Standard Agency. (1988). Determination of pH value in forest soil.

GB 7859-87, UDC 634.0.114:631.422, pp. 171-173. (in Chinese).

Chinese EPA. (2002). Chinese Standards for Surface Water Quality (GB3838—2002). Apr. 28, Chinese. EPA, Beijing.

Chinese MOH. (2006). Standard for drinking water quality. GB5749-2006.

- Chinese MOA. (2010). Determination of humus content in soil. Beijing: China Agriculture Press, 1-3. NY/T 1867-2010.
- Clarkson, T. W. (1998). Human toxicology of mercury. The Journal of trace elements in experimental medicine, 11(2-3), 303-317.
- Fang, F., Wang, Q., & Li, J. (2004). Urban environmental mercury in Changchun, a metropolitan city in Northeastern China: source, cycle, and fate. Science of the total environment, 330(1), 159-170.
- Feng, X., Shang, L., Wang, S., Tang, S., & Zheng, W. (2004). Temporal variation of total gaseous mercury in the air of Guiyang, China. Journal of Geophysical Research: Atmospheres, 109(D3).
- Feng, X., Li, P., Qiu, G., Wang, S., Li, G., Shang, L., ... & Fu, X. (2007). Human exposure to methylmercury through rice intake in mercury mining areas, Guizhou Province, China. Environmental science & technology, 42(1), 326-332.
- Feng, X., & Qiu, G. (2008). Mercury pollution in Guizhou, Southwestern China—an overview. Science of the Total Environment, 400(1), 227-237.
- Feng, X., Yin, R., Yu, B., & Du, B. (2013). Mercury isotope variations in surface soils in different contaminated areas in Guizhou Province, China. Chinese Science Bulletin, 58(2), 249-255.
- Fu, X. W., Feng, X., Liang, P., Zhang, H., Ji, J., & Liu, P. (2012). Temporal trend and sources of speciated atmospheric mercury at Waliguan GAW station, Northwestern China. Atmospheric Chemistry and Physics, 12(4), 1951-1964.
- Ganguli, P. M., Mason, R. P., Abu-Saba, K. E., Anderson, R. S., & Flegal, A. R.(2000). Mercury speciation in drainage from the New Idria mercury mine, California. Environmental science & technology, 34(22), 4773-4779.
- GB 2762, (2012). Maximum levels of contaminants in foods. The national standard of the people's republic of China.

- GB15618, (2008). Standards for soil environmental quality. National Standards of the People's Republic of China
- Gilmour, C. C., & Henry, E. A. (1991). Mercury methylation in aquatic systems affected by acid deposition. Environmental pollution, 71(2), 131-169.
- Gray, J. E., Crock, J. G., & Fey, D. L. (2002). Environmental geochemistry of abandoned mercury mines in West-Central Nevada, USA. Applied Geochemistry, 17(8), 1069-1079.
- Gray, J. E., Greaves, I. A., Bustos, D. M., & Krabbenhoft, D. P. (2003). Mercury and methylmercury contents in mine-waste calcine, water, and sediment collected from the Palawan Quicksilver Mine, Philippines. Environmental Geology, 43(3), 298-307.
- Gray, J. E., Hines, M. E., Higueras, P. L., Adatto, I., & Lasorsa, B. K. (2004). Mercury speciation and microbial transformations in mine wastes, stream sediments, and surface waters at the Almadén mining district, Spain. Environmental science & technology, 38(16), 4285-4292.
- Guo, Y., Feng, X., Li, Z., He, T., Yan, H., Meng, B., ... & Qiu, G. (2008). Distribution and wet deposition fluxes of total and methyl mercury in Wujiang River Basin, Guizhou, China. Atmospheric Environment, 42(30), 7096-7103.
- Hintelmann, H., Welbourn, P. M., & Evans, R. D. (1995). Binding of methylmercury compounds by humic and fulvic acids. Water, Air, and Soil Pollution, 80(1-4), 1031-1034.
- Hintelmann, H., Welbourn, P. M., & Evans, R. D. (1997). Measurement of complexation of methylmercury (II) compounds by freshwater humic substances using equilibrium dialysis. Environmental Science & Technology, 31(2), 489-495.
- Horvat, M., Nolde, N., Fajon, V., Jereb, V., Logar, M., Lojen, S., ... & Drobne, D. (2003). Total mercury, methylmercury and selenium in mercury polluted areas in the province Guizhou, China. Science of the Total Environment, 304(1), 231-256.
- JECFA, 2010. Summary and conclusions of the seventy-second meeting of the joint 16

FAO/WHO expert committee on food additives Rome, Italy.

- Kim, C. S., Brown, G. E., & Rytuba, J. J. (2000). Characterization and speciation of mercury-bearing mine wastes using X-ray absorption spectroscopy. Science of the Total Environment, 261(1), 157-168.
- Kim, C. S., Rytuba, J. J., & Brown, G. E. (2004). Geological and anthropogenic factors influencing mercury speciation in mine wastes: an EXAFS spectroscopy study. Applied Geochemistry, 19(3), 379-393.
- Li, L. (2014). Study on environmental pollution status and risk assessment of mercury in Xikou mercury mine area. (Doctoral dissertation, Chongqing University). In Chinese.
- Li, P., Feng, X., Shang, L., Qiu, G., Meng, B., Liang, P., & Zhang, H. (2008). Mercury pollution from artisanal mercury mining in Tongren, Guizhou, China. Applied Geochemistry, 23(8), 2055-2064.
- Liang, L., Horvat, M., & Bloom, N. S. (1994). An improved speciation method for mercury by GC/CVAFS after aqueous phase ethylation and room temperature precollection. Talanta, 41(3), 371-379.
- Liang, L., Horvat, M., Cernichiari, E., Gelein, B., & Balogh, S. (1996). Simple solvent extraction technique for elimination of matrix interferences in the determination of methylmercury in environmental and biological samples by ethylation-gas chromatography-cold vapor atomic fluorescence spectrometry. Talanta, 43(11), 1883-1888.
- Liu, N., Qiu, G., Landis, M. S., Feng, X., Fu, X., & Shang, L. (2011). Atmospheric mercury species measured in Guiyang, Guizhou province, southwest China. Atmospheric Research, 100(1), 93-102.
- Meng, B., Feng, X., Qiu, G., Cai, Y., Wang, D., Li, P., ... & Sommar, J. (2010).
 Distribution patterns of inorganic mercury and methylmercury in tissues of rice (Oryza sativa L.) plants and possible bioaccumulation pathways. Journal of agricultural and food chemistry, 58(8), 4951-4958.
- Meng, B., Feng, X., Qiu, G., Li, Z., Yao, H., Shang, L., & Yan, H. (2016). The impacts of organic matter on the distribution and methylation of mercury in a

hydroelectric reservoir in Wujiang River, Southwest China. Environmental Toxicology and Chemistry, 35(1), 191-199.

- Meng, B., Feng, X., Qiu, G., Liang, P., Li, P., Chen, C., & Shang, L. (2011). The process of methylmercury accumulation in rice (Oryza sativa L.). Environmental science & technology, 45(7), 2711-2717.
- Qiu, G., Feng, X., & Jiang, G. (2012). Synthesis of current data for Hg in areas of geologic resource extraction contamination and aquatic systems in China. Science of the total environment, 421, 59-72.
- Qiu, G., Feng, X., Li, P., Wang, S., Li, G., Shang, L., & Fu, X. (2008). Methylmercury accumulation in rice (Oryza sativa L.) grown at abandoned mercury mines in Guizhou, China. Journal of agricultural and food chemistry, 56(7), 2465-2468.
- Qiu, G., Feng, X., Meng, B., & Wang, X. (2011). Methylmercury in rice (Oryza sativa L.) grown from the Xunyang Hg mining area, Shaanxi province, northwestern China. Pure and Applied Chemistry, 84(2), 281-289.
- Qiu, G., Feng, X., Meng, B., Sommar, J., & Gu, C. (2012). Environmental geochemistry of an active Hg mine in Xunyang, Shaanxi Province, China. Applied geochemistry, 27(12), 2280-2288.
- Qiu, G., Feng, X., Meng, B., Zhang, C., Gu, C., Du, B., & Lin, Y. (2013).
 Environmental geochemistry of an abandoned mercury mine in Yanwuping,
 Guizhou Province, China. Environmental research, 125, 124-130.
- Qiu, G., Feng, X., Wang, S., & Shang, L. (2005). Mercury and methylmercury in riparian soil, sediments, mine-waste calcines, and moss from abandoned Hg mines in east Guizhou province, southwestern China. Applied Geochemistry, 20(3), 627-638.
- Qiu, G., Feng, X., Wang, S., & Shang, L. (2006a). Environmental contamination of mercury from Hg-mining areas in Wuchuan, northeastern Guizhou, China. Environmental Pollution, 142(3), 549-558.
- Qiu, G., Feng, X., Wang, S., & Xiao, T. (2006b). Mercury contaminations from historic mining to water, soil and vegetation in Lanmuchang, Guizhou, southwestern China. Science of the total environment, 368(1), 56-68.

- Rothenberg, S. E., Windham-Myers, L., & Creswell, J. E. (2014). Rice methylmercury exposure and mitigation: A comprehensive review. Environmental research, 133, 407-423.
- Rytuba, J. J. (2000). Mercury mine drainage and processes that control its environmental impact. Science of the Total Environment, 260(1), 57-71.
- Schroeder, W. H., Anlauf, K. G., Barrie, L. A., Lu, J. Y., Steffen, A., Schneeberger, D.
 R., & Berg, T. (1998). Arctic springtime depletion of mercury. Nature, 394(6691), 331-332.
- Su, Y. B., Chang, W. C., Hsi, H. C., & Lin, C. C. (2016). Investigation of biogeochemical controls on the formation, uptake and accumulation of methylmercury in rice paddies in the vicinity of a coal-fired power plant and a municipal solid waste incinerator in Taiwan. Chemosphere, 154, 375-384.
- Ullrich, S. M., Tanton, T. W., & Abdrashitova, S. A. (2001). Mercury in the aquatic environment: a review of factors affecting methylation. Critical reviews in environmental science and technology, 31(3), 241-293.
- UNEP Chemicals, & Inter-Organization Programme for the Sound Management of Chemicals. (2002). Global mercury assessment. UNEP Chemicals.
- US EPA. (1992). Water quality standards, Establishment of numeric criteria for priority toxic pollutants. States' compliance, Final rule, Federal Register, 40 CFR Part 131, vol. 246, pp.847-860.
- US EPA. (1997). Mercury study report to congress. EPA-452/R-97-0003. USEPA Office of Air Planning and Standards, Office of Research and Development.
- US EPA. (2001). Method 1630: Methyl mercury in water by distillation, aqueous ethylation, purge and trap, and CVAFS. Draft January 2001. U.S. Environmental Protection Agency, Office of Water, Office of Science and Technology Engineering and Analysis Division (4303), 1200 Pennsylvania Avenue NW, Washington, D.C. 20460. pp. 1-41.
- Wan, Q., Feng, X., Lu, J., Zheng, W., Song, X., Han, S., & Xu, H. (2009).Atmospheric mercury in Changbai Mountain area, northeastern China I. The seasonal distribution pattern of total gaseous mercury and its potential sources.

Environmental research, 109(3), 201-206.

- Wang, Q., Kim, D., Dionysiou, D. D., Sorial, G. A., & Timberlake, D. (2004). Sources and remediation for mercury contamination in aquatic systems—a literature review. Environmental pollution, 131(2), 323-336.
- Watras, C. J., & Bloom, N. S. (1992). Mercury and methylmercury, in individual zooplankton: Implications for bioaccumulation. Limnology and Oceanography, 37(6), 1313-1318.
- Weber, J. H. (1993). Review of possible paths for abiotic methylation of mercury (II) in the aquatic environment. Chemosphere, 26(11), 2063-2077.
- WHO. (2003). UN Committee Recommends new Dietary Limits for Mercury available at http://www.who.int/mediacenter/news/notes/2003/np20/en/ (Cited October 1, 2016)
- WHO. (1990). Methylmercury. Environmental health criteria 101. Geneva: World Health Organization, 1-144.
- Xu, X., Liu, N., Landis, M. S., Feng, X., & Qiu, G. (2016). Characteristics and distributions of atmospheric mercury emitted from anthropogenic sources in Guiyang, southwestern China. Acta Geochimica, 35(3), 240-250.
- Yang, Y., Chen, H., & Wang, D. (2009). Spatial and temporal distribution of gaseous elemental mercury in Chongqing, China. Environmental monitoring and assessment, 156(1-4), 479-489.
- Zhang, H., Feng, X., Larssen, T., Shang, L., & Li, P. (2010). Bioaccumulation of methylmercury versus inorganic mercury in rice (Oryza sativa L.) grain. Environmental science & technology, 44(12), 4499-4504.
- Zhang, L., Jin, Y., Lu, J., & Zhang, C. (2009). Concentration, distribution and bioaccumulation of mercury in the Xunyang mercury mining area, Shaanxi Province, China. Applied Geochemistry, 24(5), 950-956.
- Zhang, L., Lin, X., Liu, Q., Xu, Li., & Zhang, J. (2011). Effects of Mercury Mine exploitation on Local Soil and Water Environment in Xiuhsan. Journal of Southwest China Normal University (Natural Science Edition), 36(6), 105-109.

Zhao, L., Qiu, G., Anderson, C. W., Meng, B., Wang, D., Shang, L., ... & Feng, X.

(2016). Mercury methylation in rice paddies and its possible controlling factors in the Hg mining area, Guizhou Province, Southwest China. Environmental Pollution, 215, 1-9.

Figures

Figure 1 The locations of sampling points in Xiushan area, China.

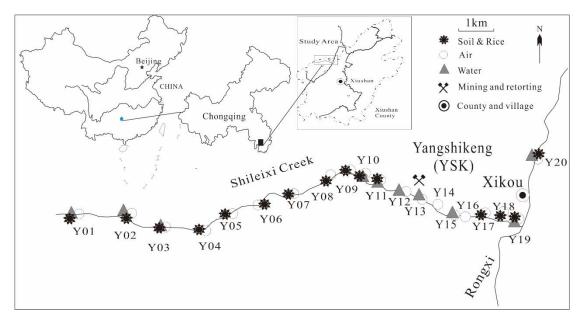
Figure 2 The distribution of different Hg species in water samples from the YSK Hg mining area, Chongqing, China.

Figure 3 The THg and MeHg distribution in paddy soils gathered from the YSK Hg mining area in Xiushan County, Chongqing, China.

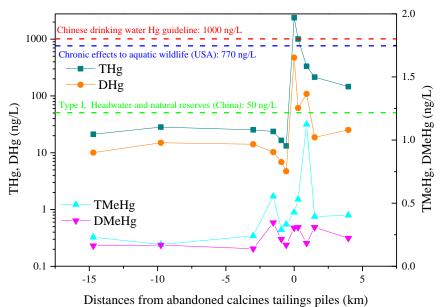
Figure 4 The GEM concentrations distribution in the ambient air influenced by artisanal Hg mines in Xiushan Hg mining region.

Figure 5 The Correlations between soil MeHg and humilic acid, and fulvic acid in our research.









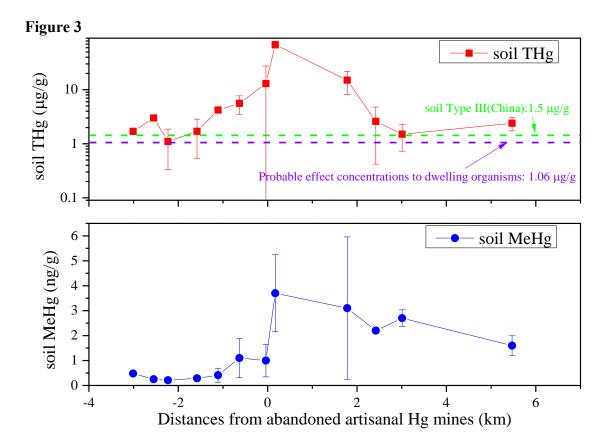
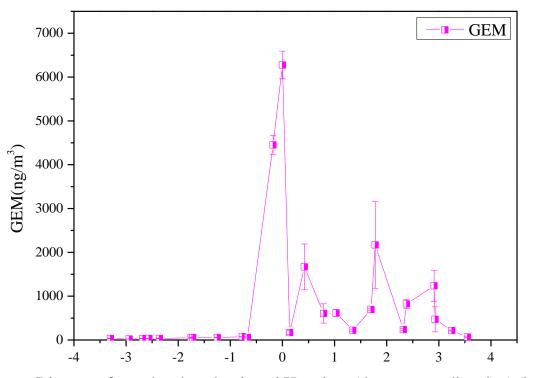
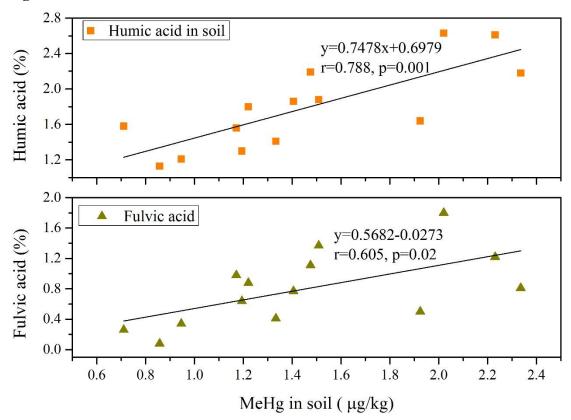


Figure 4



Distances from abandoned artisanal Hg mines (downstream direction) (km)





Tables

 Table 1 Total mercury and methylmercury in surface waters from the YSK abandoned mining areas

Table 2 The Hg and MeHg concentrations in surface water and soil collected from Hg mining regions worldwide

Table 3 Total mercury and methylmercury in rice and soil gathered from the YSK abandoned mining areas

Table 4 Correlations among soil THg (THgs), soil MeHg (MeHgs), rice THg (THg_R), rice MeHg (MeHg_R), and GEM in this study. Asterisk (**) suggests correlation is significant at the 0.01 level, and (*) denotes correlation is significant at the 0.05 level.

Table 5 Correlations among soil MeHg, organic matter, humic acid, humilic acid, and fulvic acid in our research. Asterisk (**) suggests correlation is significant at the 0.01 level, and (*) denotes correlation is significant at the 0.05 level.

		Unfilte	Unfiltered water		d water
Sample ID	Location	THg	TMeHg	DHg	DMeHg
		(ng/L)	(ng/L)	(ng/L)	(ng/L)
Upstream f	from Hg mines				
Y01	Shileixi Creek	21	0.23	10	0.16
Y02	Shileixi Creek	28	0.17	15	0.17
Y03	Shileixi Creek	25	0.24	14	0.14
Abandoned	l artisanal Hg mines				

Table 1

Y10	Shileixi Creek	24	0.55	10	0.35
Y11	Shileixi Creek	16	0.29	6.9	0.22
Y12	Shileixi Creek	13	0.33	4.7	0.17
Downstream	n from abandoned industrialized Hg mines				
Y13	Leachate from calcines	2390	0.43	473	0.30
Y14	Shileixi Creek	1004	0.53	61	0.31
Y15	Shileixi Creek	333	1.1	110	0.18
Y19	Shileixi Creek	214	0.39	19	0.31
Y20	Rongxi River	146	0.40	25	0.22
Recommend	led Hg guildelines for water				
Chronic effe	ects to aquatic wildlife (USEPA, 1992)	770			
Headwater a	and natural reserves (Chinese EPA, 2002)	50			
Chinese drin	nking water Hg guideline (Chinese MOH,	1000			
2006)		1000			

Table 2

lable 2								
	Unfiltered water		Filtered water Soil					
Location	THg	TMeHg	DHg	DMeHg	THg	MeHg	Reference	
	(ng/L)	(ng/L)	(ng/L)	(ng/L)	$(\mu g/g)$	(ng/g)		
Hg mines,							Ganguli et al., 2000; Rytuba, 2000;	
California, USA	0.65-45000	0.017-47	0.2-140	0.017-1.2	0.6-1500		Domagalski et al., 2001; Thomas et al., 2002;	
Camonna, USA							Nacht et al., 2004; Suchanek et al., 2009	
Hg mines, Alaska, USA	0.1-2500	0.01-1.2	1.3-50		0.03-5326	0.02-41	Gray et al., 2000; Bailey et al., 2002	
							Higueras et al., 2003; Berzas Nevado et al.,	
Almadén Hg mine,	<d.l20300 0.<="" td=""><td>0.04-30 2.5-136</td><td>2.5-136</td><td></td><td colspan="2">3.4-8889</td><td colspan="2">2003, 2009; Gray et al., 2004; Millán et a</td></d.l20300>	0.04-30 2.5-136	2.5-136		3.4-8889		2003, 2009; Gray et al., 2004; Millán et a	
Spain							2006; Bueno et al., 2009	
Iduiio II o mino							Hines et al., 2000; Gnamuš et al., 2000;	
Idrija Hg mine, Slovenia	0.20-708	0.01-0.6	0.03-134	0.10-0.27	0.18-2759	0.32-80	Horvat et al., 2002; Gosar et al., 2006;	
Slovenia							Kocman et al., 2004, 2011	
Palawan Hg mine, Philippines	1.0-31000	<0.02-3.1	8.0-30000		0.012-566		Gray et al., 2003; Maramba et al., 2006	
							Horvat et al., 2003; Qiu et al., 2005, 2009;	
Wanshan Hg mine,	3.3-10580	0.012-25	1.1-655	<0.005-6.0	0.10-790	0.10-23	Zhang et al., 2010a, b; Søvik et al., 2011; Lin	
China							et al., 2010, 2011	
Xunyang Hg mine,	6.2-23500	0.022-3.7	0.21-3500	0.015-0.52	1.3-750	1.2-11	Zhang et al., 2009; Qiu et al., 2012a	
China	0.2-23300	0.022-3.7	0.21-3300	0.015-0.52	1.3-730	1.2-11	Zhang et al., 2009, Qiu et al., 2012a	

Yanwuping Hg mine, China	3.8-51	0.20-2.7	0.85-27	0.038-0.38	0.24-240	0.34-7.3	Qiu et al., 2013
Xiushan Hg mine, China	21-2390	0.17-1.1	4.7-473	0.14-0.35	0.48-68	0.21-3.7	This study

3 Table 3

		Se	oil	Rice		
Sample ID	Location	THg	MeHg	THg	MeHg	
		$(\mu g/g)$	(ng/g)	(ng/g)	(ng/g)	
Upstream fr	om Hg mines					
Y01	Shileixi Creek	0.48	0.88	4.8	3.0	
Y02	Shileixi Creek	1.0	1.1	12	9.2	
Y03	Shileixi Creek	4.9	1.3	14	7.8	
Y04	Shileixi Creek	1.7	0.48	16	5.7	
Y05	Shileixi Creek	3.0	0.25	10	5.3	
Y06	Shileixi Creek	1.1	0.21	7.1	4.7	
Y07	Shileixi Creek	1.7	0.29	14	5.8	
Y08	Shileixi Creek	4.2	0.41	16	7.1	
Y09	Shileixi Creek	5.6	1.1	18	12	
Abandoned	artisanal Hg mines					
Y10	Shileixi Creek	13	1.0	120	21	
Y11	Shileixi Creek	68	3.7	384	64	
Downstream	n from Hg mines					
Y17	Shileixi Creek	15	3.1	12	8.7	
Y18	Shileixi Creek	2.6	2.2	34	9.3	
Y19	Shileixi Creek	1.5	2.7	36	8.4	
Y20	Rongxi River	2.4	1.6	14	9.2	
Recommend	led Hg guildelines					
Probable effect concentration to dwelling		1.00				
organisms (Macdonald et al., 2000)	1.06				
Available for agricultural soil (Chinese		1.5				
MOA, 2010)	1.5				
Limit for ce	real food (Chinese, GB 2762-					
2012)				20		

Table 4

	THgs	MeHgs	THg _R	MeHg _R	GEM
THgs	1				
MeHgs	0.655**	1			
THgr	0.965**	0.591*	1		
MeHgr	0.975**	0.621*	0.991*	1	
GEM	0.995**	0.728*	0.999*	0.993*	1

Table 5

Correlations

	MeHgs	Organic Matter	Humic acid	Fulvic acid	Humin
MeHg _s	1				
Organic Matter	0.388	1			
Humic acid	0.725**	0.342	1		
Fulvic acid	0.536*	0.119	0.838**	1	
Humin	0.385	0.4	0.362	-0.136	1